

SYNTHESIS OF NOVEL ANALOGS OF BELACTOSIN A AND  
CARFILZOMIB

by

Mahesh K. Yadab

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Thesis Committee:

Dr. Norma K. Dunlap

Dr. Scott Handy

Dr. Kevin L. Bicker

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## ABSTRACT

Belactosin A, a naturally occurring proteasome inhibitor with potent anti-tumor activity, was discovered in the late 1980's. The key structural features of the natural product include a cyclopropane ring with a terminal  $\beta$ -lactone "serine trap". Activity is conferred by acylation of the  $\beta$ -lactone with a threonine of the proteasome. Carfilzomib is a tetrapeptide epoxyketone, and an analog of epoxomicin with improved properties. In phase I and phase II clinical trials, carfilzomib exhibited higher selectivity, equal potency and less peripheral neuropathy than bortezomib and salinosporamide. Its activity is due to the formation of a highly stable six membered morpholine ring by the interaction between epoxyketone, free hydroxy and  $\alpha$ -amino groups of the threonine.

Several syntheses of belactosin A and carfilzomib have been reported, however only a handful of analogs have been reported. An approach to the synthesis of novel hybrid analogs is reported here, with the synthesis of phenylalanine, leucine and valine cyclopropyl analogs of belactosin A with the epoxyketone of carfilzomib. Key steps in the synthetic route include a cyclopropanation and formation of an enone followed by epoxidation.

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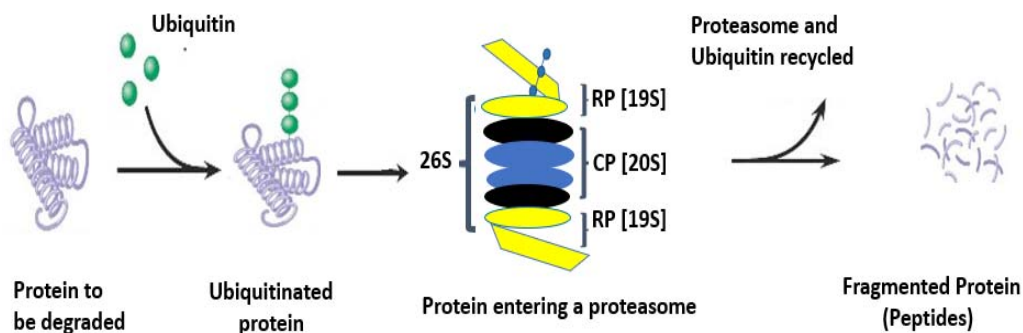
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## CHAPTER I: INTRODUCTION

### Proteasome structure and function

The traditional methods for treatment of cancers are surgery, radiation and chemotherapy. However, many treatments cause side effects, and not all are effective. A novel treatment for cancer chemotherapy is by proteasome inhibition.<sup>1</sup> The 26S proteasome is an ATP-dependent multi-subunit proteolytic complex expressed in the nucleus and cytoplasm of all eukaryotic cells.<sup>2</sup> It is responsible for regulation of the mutant, damaged and misfolded proteins by the ubiquitin proteasome pathway (UPP), and is a subject of research in anticancer therapy.<sup>3</sup> The proteasome consists of two 19S regulatory particles (RP) and a 20S catalytic core particle (CP). The 20S CP is barrel-shaped and is composed of four stacked rings ( $\alpha_7, \beta_7, \beta_7, \alpha_7$ ) with 28 subunits (**Figure 1**). Two outer rings on either side are composed of seven different  $\alpha$ -subunits while the central two rings are composed of seven different  $\beta$ -subunits ( $\beta_1, \beta_2, \beta_3, \beta_4, \beta_5, \beta_6, \beta_7$ ) with six active sites; two of each chymotrypsin-like (CT-L), trypsin-like (T-L) and caspase-like (C-L)<sup>2,4,5</sup>. The 19S RP binds polyubiquitinated proteins and, due to its ATP activities, polyubiquitinated proteins begin to unfold and insert the protein into the 20S CP, where the protein breaks into smaller peptides having 3-22 amino acids in length.

The protein to be degraded is flagged by polyubiquitination.<sup>6</sup> This process is catalyzed by the enzyme E1 (ubiquitin-activating enzyme), E2 (ubiquitin-conjugation enzyme) and E3 (ubiquitin ligase) which activate free ubiquitin and attach it to the target protein.



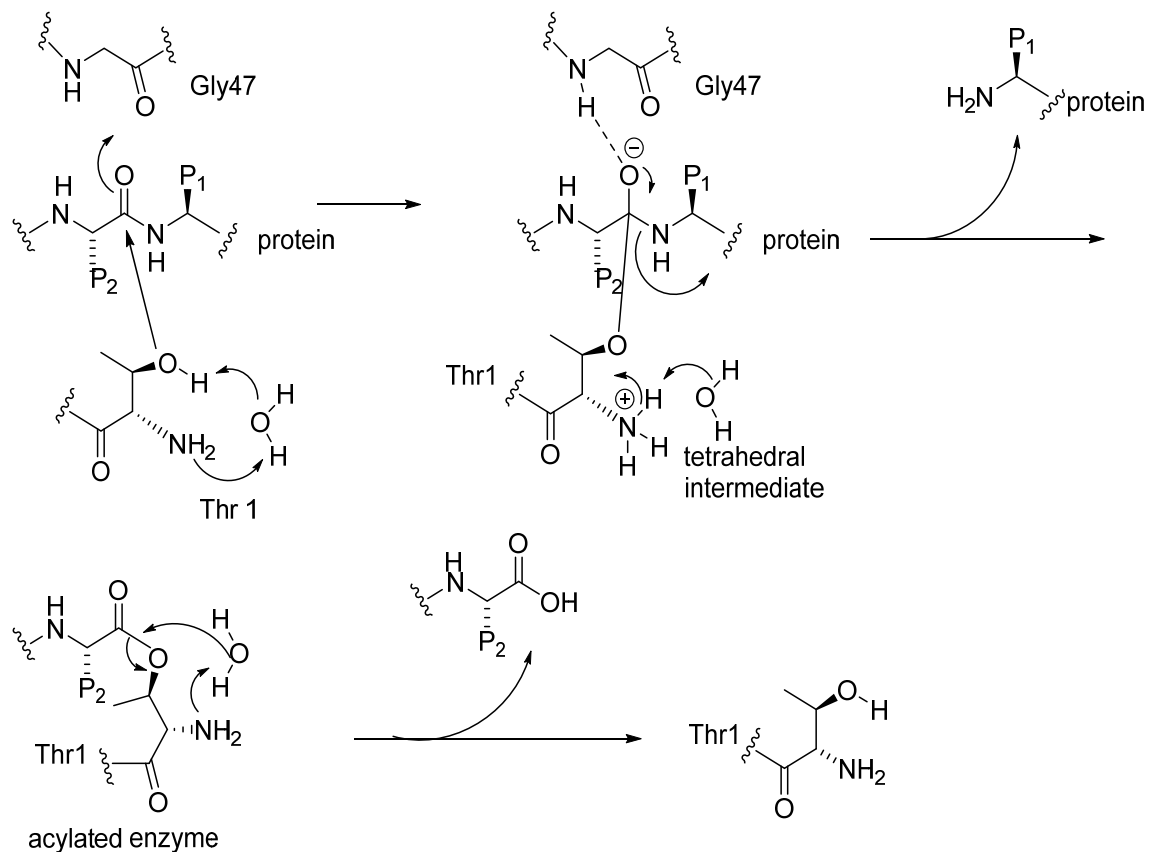
**Figure 1:** The simplified scheme of the ubiquitin-proteasome pathway

The process starts by the formation of a thioester bond between E1 and ubiquitin in the presence of ATP. This activated ubiquitin then transfers to the E2 enzyme and E2 presents ubiquitin to E3 which brings ubiquitin. This process is repeated several times, creating a polyubiquitin chain that tags the protein for degradation by the proteasome.

### **Proteasome's catalytic mechanism**

The proteasome plays a vital role to control protein quality, antigen processing, signal transduction, cell differentiation, cell-cycle progression and apoptosis.<sup>7</sup> Proteasomes contain a new class of proteolytic enzyme called threonine (Thr) proteases,<sup>8</sup> which are activated by the catalytic triad having acidic (Asp or Gly), a basic histidine (His) and the nucleophilic (Thr) residue.<sup>9</sup> The N-terminal threonine of  $\beta$ -subunits are utilized by all the proteolytic sites present in proteasomes. According to X-ray crystallography, in presence of a water molecule the nucleophilic (Thr-O) attack is activated by N-terminal Thr1 to add to the carbonyl carbon of a protein substrate to form an intermediate tetrahedral oxyanion (**Figure 2**), which forms an intramolecular hydrogen bond with the Gly 47-N (oxyanion hole).

A water molecule is activated by the deprotonated Thr1 amino group to hydrolyze the acyl-enzyme to regenerate the Thr1-OH (free proteasomes) and second product (**Figure 2**).<sup>7</sup>

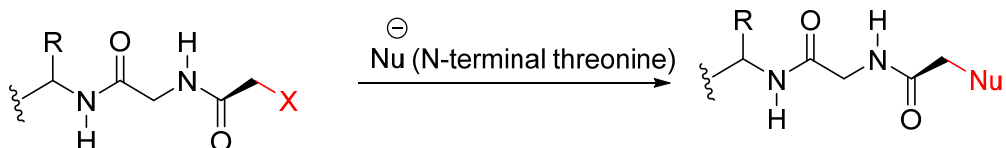


**Figure 2:** General proposed mechanism for the proteasome-catalyzed hydrolysis of peptides

### Proteasome inhibitors

The proteasome crucial role was discovered in the late 1980s.<sup>8</sup> The cell cycle depends on degradation of protein, and proteasome inhibition mechanism is a new way to fight against non-small cell lung carcinoma (NSCLC),<sup>10</sup> chemo resistant multiple myeloma

(MM)<sup>11</sup>, and several other types of cancers. According to one model, proteasome inhibitors caused apoptosis mostly in transformed cancer cells.<sup>12</sup> Common features of inhibitors include being peptide-like and having a reactive group “serine trap” to form a covalent bond (**Figure 3**) with the nucleophilic threonine.<sup>8</sup>



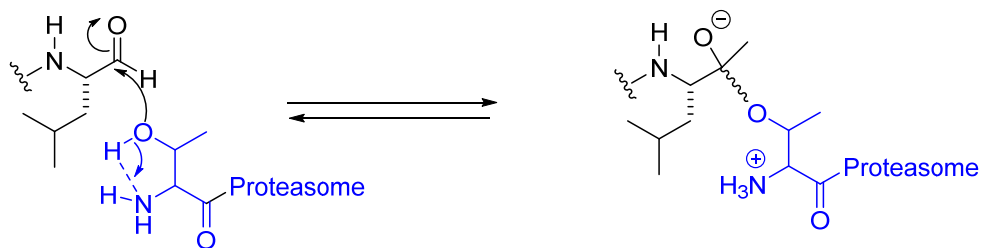
**Figure 3:** Inhibitors showing active site (a serine trap X)

Proteasome inhibitors are basically two types on the basis of sources: chemically synthesized small molecules, and derivatives of natural products.<sup>13</sup> All inhibitors act on CT-L activity, but also have a weaker effect on T-L and C-L sites. Synthetic inhibitors are classified as peptide aldehydes, peptide boronates, peptide lactones and peptide epoxyketones, and will be discussed further.

### 1.1 Peptide aldehyde

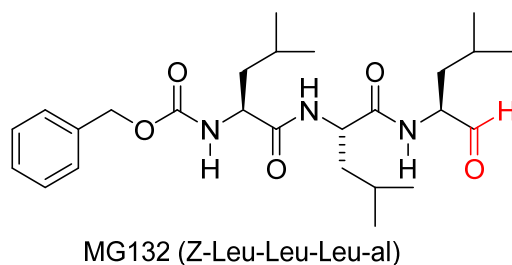
Peptide aldehydes were the first reported reversible proteasome inhibitors. Aldehyde inhibitors have slow binding activity. These also block other serine and cysteine proteases leading to off-target effects. Because of fast dissociation rates, these are quickly metabolized into inactive acids by the oxidation process and taken out of the cell through the multi-drug resistance (MDR) system carrier.<sup>7,8,13</sup> The electrophilic carbon atom of the carbonyl group forms a covalent bond with the nucleophile, which is semi-reversible (**Figure 4**).





**Figure 4:** General mechanism of aldehyde inhibitors

Only a few aldehyde inhibitors, like MG132 (Z-Leu-Leu-Leu-al) in **Figure 5**, are commercially available, and can inhibit CT-L activity of proteasomes. None are approved as drugs but are used as research tools to study the proteasome.



**Figure 5:** Structure of MG132 (Z-Leu-Leu-Leu-al) (a serine trap in red)

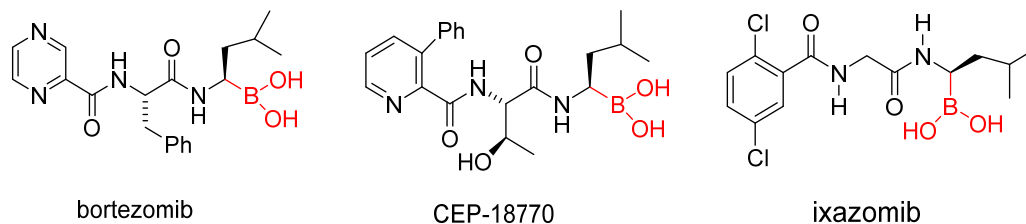
## 1.2 Peptide boronates

Peptide boronates are formed by replacing the aldehyde functionality of aldehyde inhibitors by a boronic acid moiety and are much improved over aldehyde inhibitors.<sup>14</sup> Even low nanomolar concentrations inhibit CT-L activity and have no effect on cysteine proteases.<sup>7</sup> Because of much slower dissociation rate of the boronate-proteasome adducts with respect to the aldehyde-proteasome adducts, these are practically irreversible, despite being reversible inhibitors. A covalent tetrahedral boron adduct is formed between the electrophilic boron atom and the active site Thr1-O (**Figure 6**).



**Figure 6:** General mechanism of boronate inhibitors

Boronate inhibitors are found to be more specific than aldehyde inhibitors and show very poor inhibitory effect on cysteine proteases due to the weak interaction between boron and sulfur. Bortezomib (**Figure 7**) is a dipeptide boronate inhibitor, marketed as Velcade. It was approved by the FDA to be used clinically for the treatment of multiple myeloma and mantle cell lymphoma in 2003 and 2006, respectively.<sup>15</sup> The drug is currently given with the various chemotherapeutic agents (doxorubicin, thalidomide, melphalan) in several active clinical trials for the treatment of several types of cancers, but has severe side effects like nausea, diarrhea, and peripheral neuropathy.<sup>7,16</sup> Ixazomib (**Figure 7**) is also known as MLN9708. Ixazomib is the first oral compound and second-generation proteasome inhibitor approved by the US FDA for the treatment of multiple myeloma (MM). Ixazomib has better pharmacokinetic and pharmacodynamic terms with respect to bortezomib. By the use of ixazomib severe peripheral neuropathy was hardly noticed, but side effects like thrombocytopenia, nausea, vomiting, diarrhea, fatigue, and rash were very common.<sup>17,18</sup>

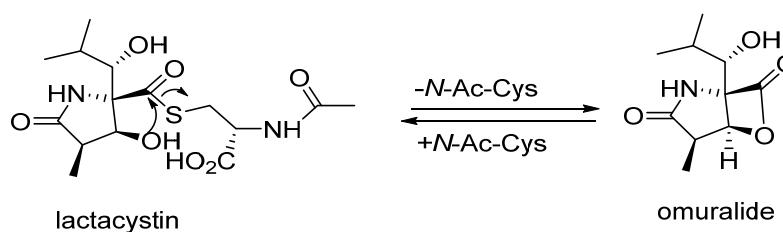


**Figure 7:** Structure of bortezomib, CEP-18770 and ixazomib (a serine trap in red)

CEP-18770 (**Figure 7**) is found to be far better than bortezomib in terms of various biological activities like enzymatic, cellular and antiproliferative and pharmacological properties,<sup>7</sup> but is not yet approved.

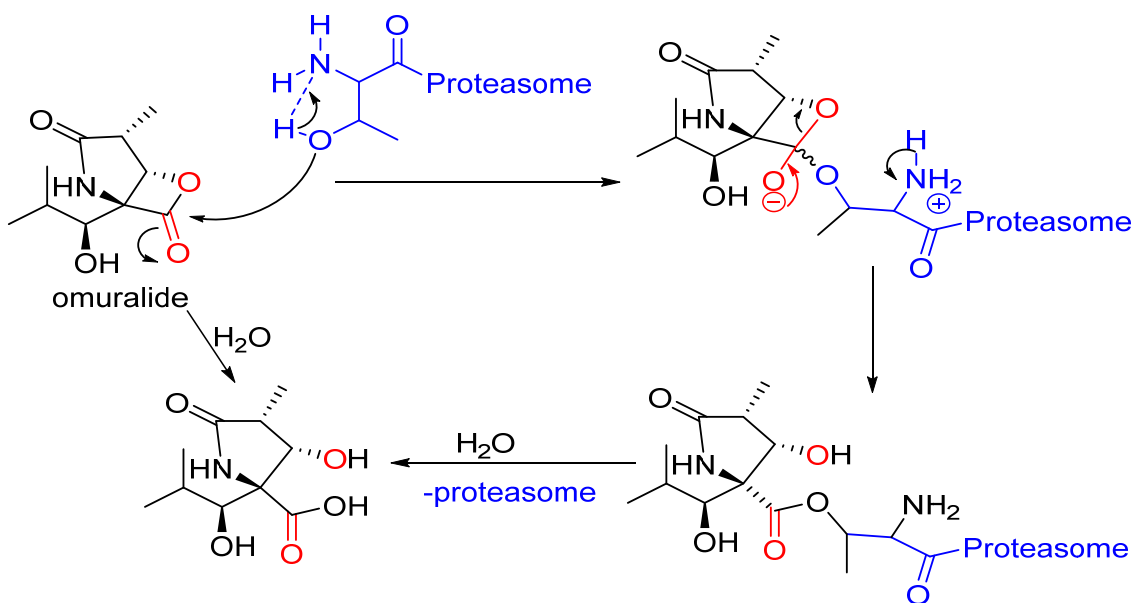
### 1.3 $\beta$ -lactones and lactacystin inhibitors

Lactacystin (**Figure 8**) was discovered by Omura and co-workers, and later synthesized by Corey and co-workers.<sup>8</sup> This is the first natural non-peptidic proteasome inhibitor and was isolated from a strain of *Streptomyces* which produces omuralide or lactacystin-  $\beta$ -lactone.



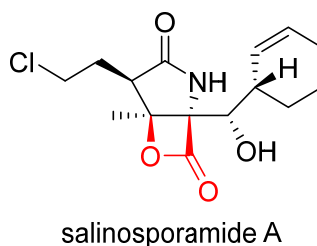
**Figure 8:** Lactacystin interconversion to omuralide

Omuralide's low nanomolar concentrations can inhibit the CT-L activity of the proteasome (**Figure 8**) without inhibiting most of the serine and cysteine proteases. The  $\beta$ -lactone forms an adduct with proteasome, which is slowly ( $t_{1/2} \approx 20$  hr) hydrolyzed by water, though it is regarded as an irreversible inhibitor and more selective than aldehyde inhibitors.<sup>7,8,14,19</sup>



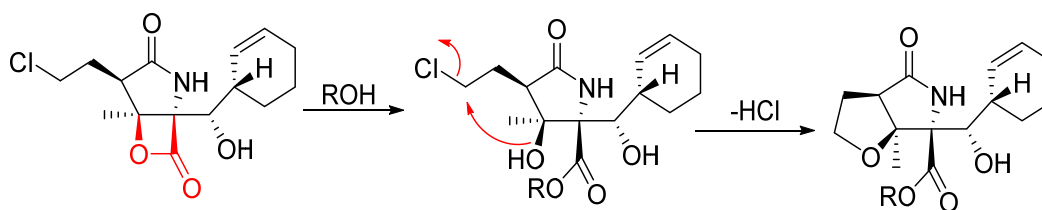
**Figure 9:** General mechanism of  $\beta$ -lactones and lactacystin inhibitors

Salinosporamide A (**Figure 10**) was isolated from the marine actinomycete *Salinispora tropica* by Fenical *et al.* in 2003.<sup>14</sup> It is very important  $\gamma$ -lactam- $\beta$ -lactone.



**Figure 10:** Salinosporamide A (a serine trap in red)

Salinosporamide A has a unique structure, that is a fused  $\gamma$ -lactam- $\beta$ -lactone bicyclic ring structure similar to lactacystin. According to X-ray crystallography structure, Groll *et al.* proposed it as an irreversible proteasome inhibitor. The N-terminal Thr-O group is acylated through opening of the  $\beta$ -lactone ring (**Figure 11**), and the C6 hydroxy group displaces the chloride group to form the tetrahydrofuran intermediate.<sup>14</sup>

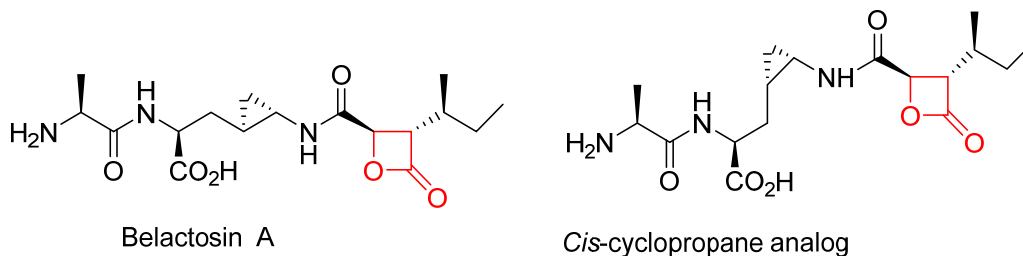


**Figure 11:** Mechanism of the irreversible inhibition of Salinosporamide A

It stops the cell proliferation of various tumor cells lines ( $IC_{50} < 10$  nM) as well as bortezomib-resistant myeloma cell lines. It is 35 times more potent than omuralide, which irreversibly inhibits 20S proteasome CT-L activity with  $IC_{50}$ s of 2.5 to 4.3 nM.

#### 1.4 Belactosin and derivatives

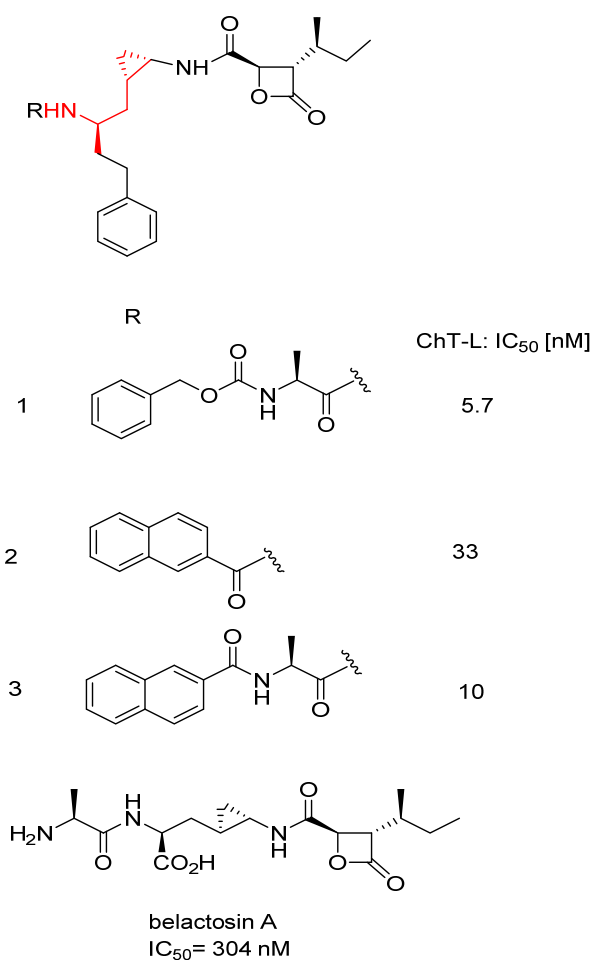
Belactosin A (**Figure 12**) is a tripeptide natural product. Asai and co-worker discovered and isolated it from *Streptomyces sp.* in 2000.<sup>19</sup> It is a reversible proteasome inhibitor with antitumor activity. From X-ray crystallographic analysis, it is found that  $\beta$ -lactone ring opening inhibits CT-L activity of proteasome by acylating the active-site Thr residue.<sup>20,21</sup>



**Figure 12:** Structure of belactosin A and *cis*-cyclopropane analog

The studies by Shuto *et al.* show that the natural belactosin A with the *trans*-cyclopropane structure is less potent than the unnatural *cis* isomer (**Figure 12**).<sup>22</sup> While investigating the synthesis of cyclopropane derivatives with a desired stereochemistry, Shuto *et al.* studied

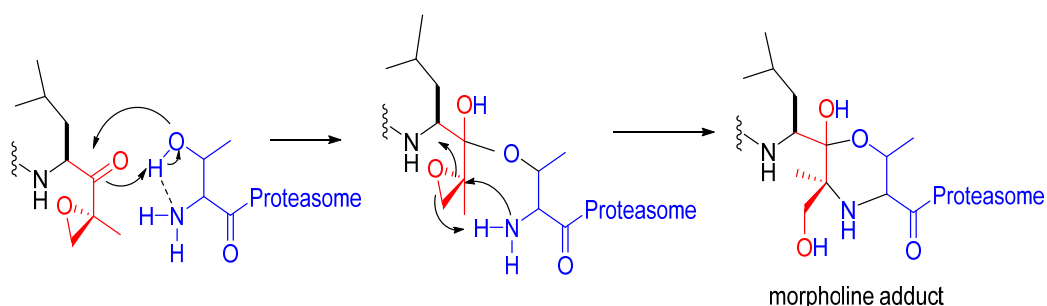
systematic structure-activity relationship (SAR) of belactosin A and found compounds **1-3 (Figure 13)** were highly effective proteasome inhibitors.<sup>23</sup> In addition to finding that the trans-cyclopropane stereochemistry is not required, they also found that replacing the carboxylic acid with a hydrophobic group such as phenylethyl improved potency.<sup>24</sup> Also, protection of the primary amine as a carbobenzoxy (cbz) group improved potency (**compound 1**).



**Figure 13:** Development of potent proteasome inhibitors using belactosin A analog, compounds assayed for inhibition of ChT-L activity of proteasome and for cancer cell growth.

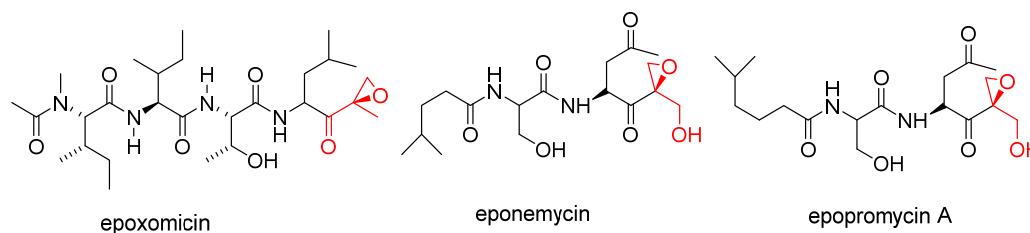
## 1.5 Peptide epoxyketones

Epoxyketones are very specific, effective irreversible proteasome inhibitors. They have a different mechanism to act on proteasome active sites. A very stable six membered morpholine adduct (**Figure 14**) is formed by reacting both Thr1 hydroxy and amino group of the proteasome with the epoxyketone site.<sup>8,25</sup>



**Figure 14:** General mechanism of peptide epoxyketone proteasome inhibition.

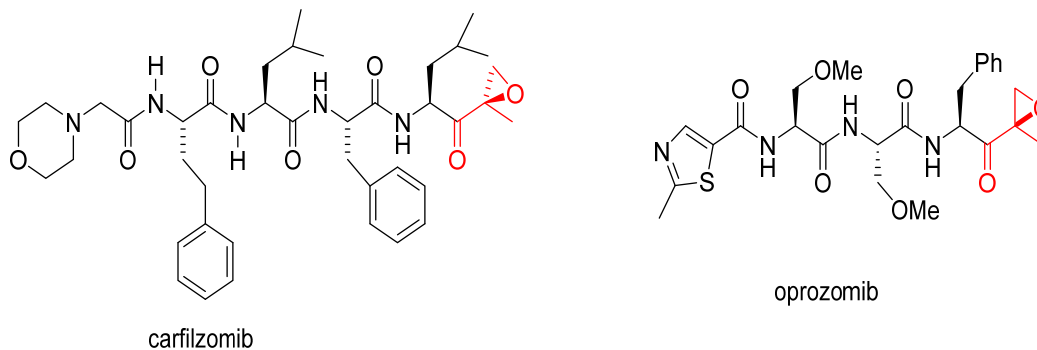
The first epoxyketone reported as a proteasome inhibitor was epoxomicin (Figure 15), which was isolated from the actinomycete strain Q996-17. It is a  $\alpha,\beta$ -epoxyketone peptide and primarily inhibits CT-L activity irreversibly. It has shown high *in vivo* antitumor activity against solid B16 melanoma tumors.<sup>7,8,14</sup>



**Figure 15:** Naturally occurring epoxyketone epoxomicin (a serine trap in red)

Other natural epoxyketone derivatives like eponemycin and epopromycins A (**Figure 15**) have *in vivo* antitumor activity.<sup>8,14</sup> After the clinical success of bortezomib and to enhance

effect of epoxomicin, a second-generation proteasome inhibitor carfilzomib (**Figure 16**) was developed. It is a tetrapeptide epoxyketone and an analog of epoxomicin.



**Figure 16:** Carfilzomib and oprozomib (a serine trap in red)

Carfilzomib was approved by FDA in 2012 for the treatment of multiple myeloma, non-Hodgkin's lymphoma and solid tumors. In phase I and phase II clinical trials, carfilzomib exhibited higher selectivity ( $IC_{50}$  = 6 nM, 3600 nM, 2400 nM for CT-L, T-L, PA activity respectively), equal potency and less peripheral neuropathy with respect to bortezomib and salinosporamide.<sup>7,27,28</sup>

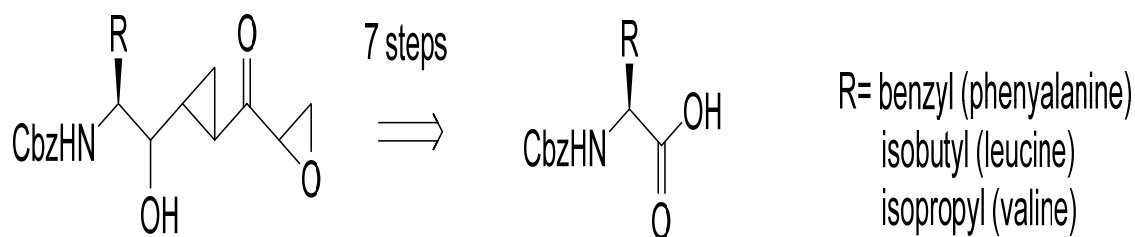
To make treatment easier, and to remove adverse effects of intravenously administered agents like carfilzomib and bortezomib, recently the next generation oprozomib (**Figure 16**) was developed.<sup>29</sup> Oprozomib is orally bioactive and irreversible proteasome inhibitor, and selectively inhibits the CT-L activity.

Since the discovery of the first proteasome inhibitors, much effort has been put into developing these compounds into drugs. While a number of proteasome inhibitors have been developed, there are still only a few on the market.



## Goals of project

The primary goal of this project is to synthesize a hybrid analog of belactosin A and carfilzomib. This would combine the semi-rigid cyclopropyl backbone of belactosin A with the epoxyketone serine trap of carfilzomib. The first objective is to synthesize the cyclopropyl backbone of belactosin A from the protected amino acids phenylalanine, leucine and valine, which will give three different hydrophobic replacements for the acid. The second objective is to incorporate the epoxyketone of carfilzomib in the analogs. The final objective is to study biological activity assays on proteasome inhibition and on cancer cells. The retrosynthetic (**Figure 17**) approach to synthesize the final product is a seven-step process.



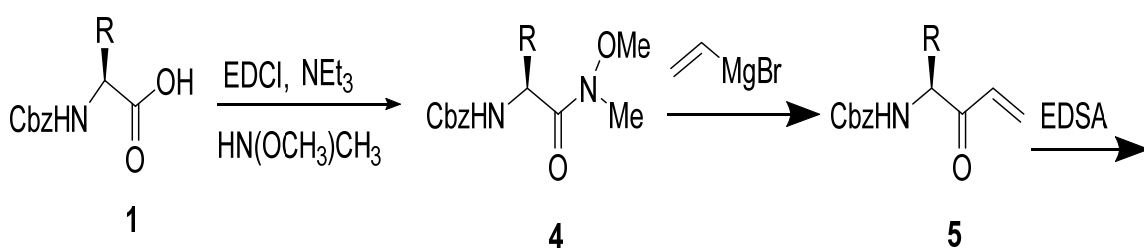
**Figure 17:** Retro-synthetic approach to proposed analogs

## Synthesis of cyclopropyl esters

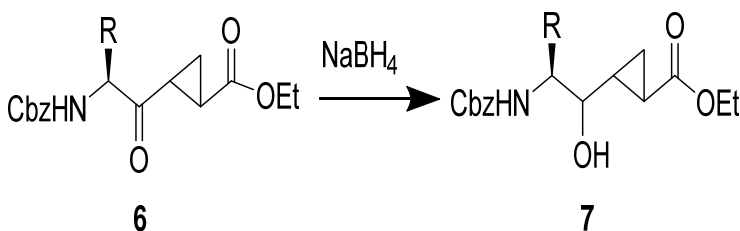
Dr. Dunlap's group has developed a new route to the cyclopropyl backbone which is outlined in **Scheme 1**. The synthesis starts with protected amino acids peptidomimetic and affords cyclopropyl esters. The protected amino acids are converted to Weinreb

intermediate enones (**5** in **Scheme 1**). The enones, on treatment with EDSA (ethyl (dimethylsulfuranylidene) acetate, a sulfonium ylide afford cyclopropyl esters, which are then reduced to the corresponding alcohols (**7** in **Scheme 1**).<sup>30,31,32</sup>

**Scheme 1:** Synthesis of cyclopropyl esters



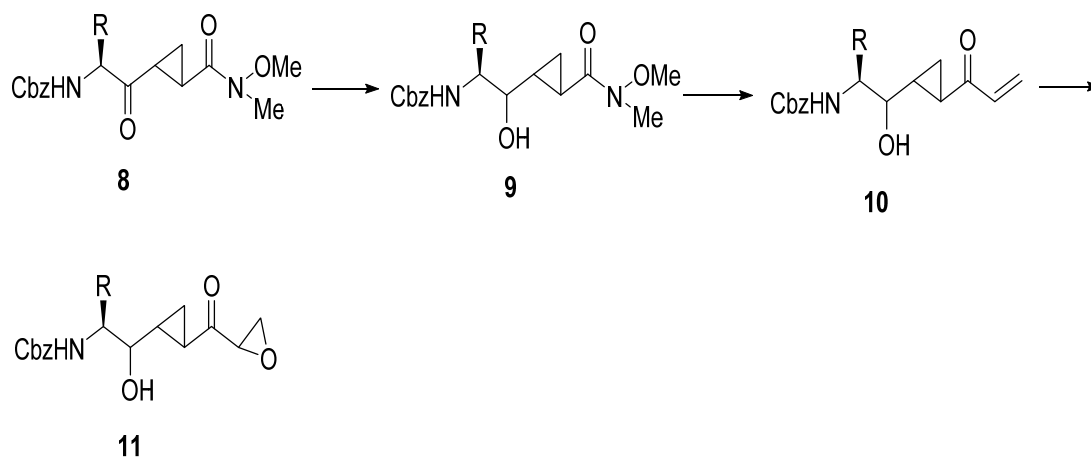
R= benzyl (phenylalanine) (a)  
 isobutyl (leucine) (b)  
 isopropyl (valine) (c)



### Proposed synthesis of Epoxides

The goal is to adapt this synthesis to give cyclopropyl epoxides as hybrid analogs. The synthesis will be modified to give a cyclopropyl Weinreb amide **8**, which will be reduced to the alcohol **9**, as outlined in **Scheme 2**.<sup>33,34</sup>

**Scheme 2:** Proposed Synthesis of epoxyketone hybrids.



Addition of vinylmagnesium bromide to either **9** or a protected alcohol derived from **9** should give the cyclopropyl enone **10**. Finally, epoxidation of the enone will give the hybrid cyclopropyl epoxyketone **11**.

## CHAPTER II: MATERIALS AND METHODS

### Instruments, materials, and reagents

The NMR data are obtained using a 500 MHz FT-NMR model ECA-500 JEOL (Peabody, MA) purchased with funding provided by National Science Foundation through the NSF-RUI program (#0321211) and a 300 MHz FT-NMR model ECA-300 JOEL (Peabody, MA). The Chemical shifts are reported in parts per million with respect to tetramethylsilane (TMS). The splitting pattern are represented by the following: s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet), dd (doublet of doublets), ddd (doublet of double doublets), dt (doublet of triplets) and br (broad signal). Coupling constant (J values) are recorded in Hz. High-resolution electrospray ionization-mass spectrometry (ESI-MS) was performed at Notre Dame University, Notre Dame, Indiana.

Thin layer chromatography (TLC) was performed on glass plates coated with silica gel and UV active backing purchased from Fisher Scientific, Pittsburg, PA. The TLC plates were analyzed with a short wavelength (254 nm) UV light and followed by staining with either phosphomolybdic acid (PMA), ninhydrin, or anisaldehyde (reagent grade, Aldrich, Milwaukee, WI). Flash column chromatography was performed with silica gel, 32-63-micron ASTM (reagent grade, Fisher Scientific, Pittsburg, PA), and flash chromatography was performed on an ISCO CombiFlash R<sub>f</sub> 200 (Teledyne ISCO, Lincoln, NE) using a Teledyne ISCO cartridge preloaded with 5g of normal phase silica and a Teledyne ISCO preloaded 12g flash column.

Dichloromethane (DCM), methanol (MeOH), acetone, ethyl acetate (EA), ethanol (EtOH), and hexanes (Hex) were purchased from Fisher Scientific, Pittsburg, PA. Chloroform was purchased reagent grade from Acros Organic, New Jersey, USA. Anhydrous tetrahydrofuran (THF) was obtained using a Pure Solv solvent purification system (Model PS-MD-3, Innovative Technology, Amesbury, MA). deuterio-chloroform ( $\text{CDCl}_3$ ), deuterio-methanol (Methanol- $d_4$ ), and Deutero-acetone (Acetone- $d_6$ ) were purchased from Aldrich, Milwaukee, WI. Solvent extractions were performed using EA or DCM where indicated and washed with either distilled water, 1M hydrochloric acid (HCl), saturated sodium bicarbonate ( $\text{NaHCO}_3$ ), and brine (reagent grade, Fisher Scientific, Pittsburg, PA). The organic layer was dried with magnesium sulfate (Fisher Scientific, Pittsburgh, PA) and filtered. Evaporation of solvents was achieved using a Heidolph rotary evaporator (Model G3, Schwabach, Germany).

Triethylamine ( $\text{NEt}_3$ ) was obtained from Fisher Scientific, Pittsburgh, PA. Anhydrous reagent grade vinylmagnesium bromide and tetrahydrofuran (THF) were purchased from Aldrich, Milwaukee, WI. Other reagents including vinylmagnesium bromide solution (1M), N-carbobenzyloxy-L-phenylalanine, Z-L-phenylalanine, N-(3-dimethylaminopropyl)-N<sup>1</sup>-ethylcarbodiimide hydrochloride, bromoacetyl bromide, dimethyl sulphide, carbobenzyloxy-valine, *tert*-butylhydroperoxide solution, potassium *tert*-butoxide, trifluoroacetic anhydride (TFAA), N,O-dimethyl hydroxylamine hydrochloride, N-(3-dimethylaminopropyl)-N-ethylcarbodiimide hydrochloride (EDCI), sodium borohydride, and ammonium formate were obtained from Aldrich, Milwaukee, WI.

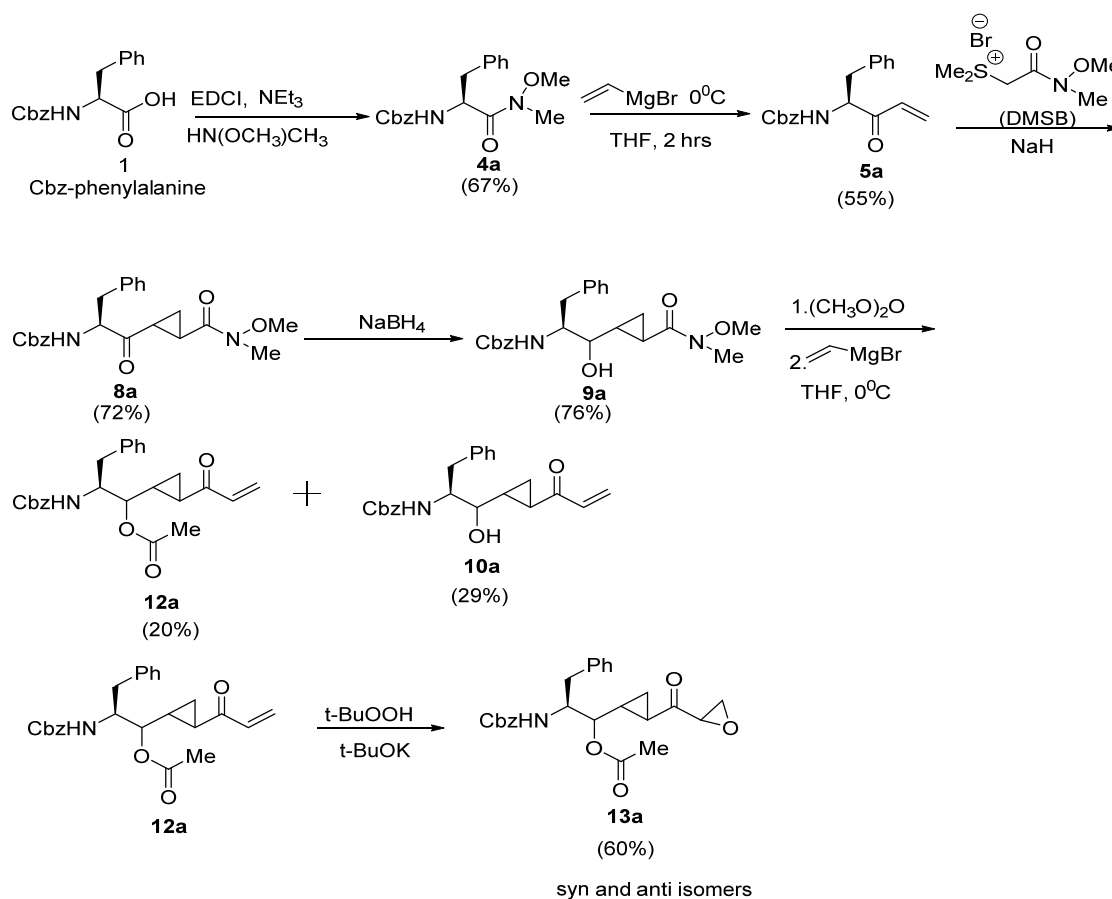
## Synthetic methods

The specific synthesis of three of the analogs is described here.

### 2.1 Synthesis of hybrid analog from Cbz-phenylalanine

The Cbz protected phenylalanine was converted to a Weinreb amide **4a**, followed by treatment with vinylmagnesium bromide to afford an enone **5a**. The enone was treated with DMSB salt and a base to form mixture of cyclopropyl ketone isomers **8a**. There were reduced by sodium borohydride to form mixture of cyclopropyl alcohols **9a**.

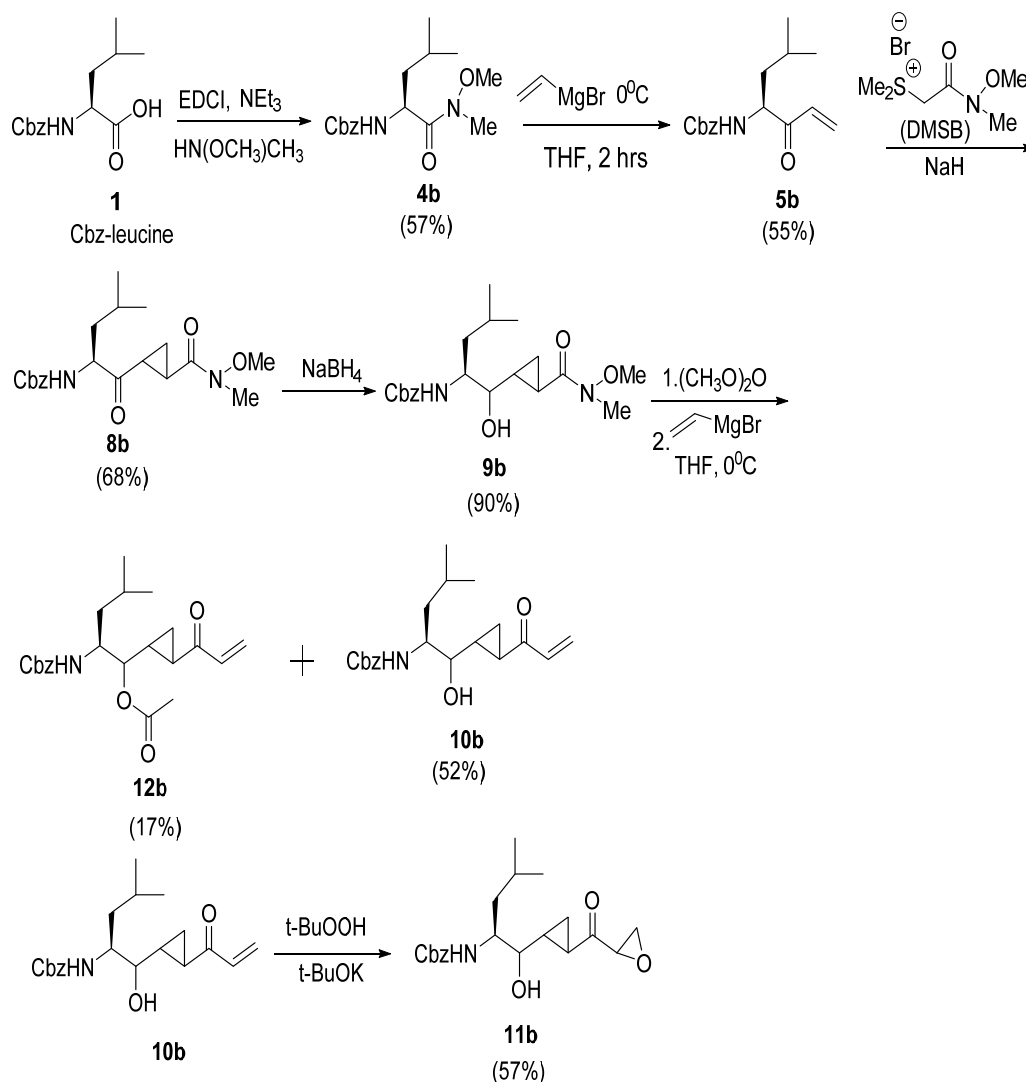
**Scheme 3:** Synthesis of hybrid analog of belactosin A and carfilzomib from Cbz-phenylalanine



The alcohol group was protected by acylation then treated with vinylmagnesium bromide to afford enone **12a**, as well as deacylated alcohol **10a**. The enone was treated with *t*-butylperoxide in presence of base to afford a hybrid analog of belactosin A and carfilozomib **13a**. The route is outlined in **Scheme 3**.

## 2.2 Synthesis of hybrid analog from Cbz-leucine

The Cbz protected leucine was converted to a Weinreb amide **4b**, followed by treatment with vinylmagnesium bromide to afford an enone **5b**. The enone was treated with DMSB salt and a base to form mixture of cyclopropyl ketone isomers **8b**. These were reduced by sodium borohydride to form mixture of cyclopropyl alcohols **9b**. The alcohol group was protected by acylation then treated with vinylmagnesium bromide to afford enone **12b**, as well as deacylated alcohol **10b**. The enone was treated with *t*-butylperoxide in presence of base to afford a hybrid analog of belactosin A and carfilozomib **11b**. The route is outlined in **Scheme 4**.

**Scheme 4:** Synthesis of hybrid analog of belactosin A and carfilzomib from Cbz-leucine

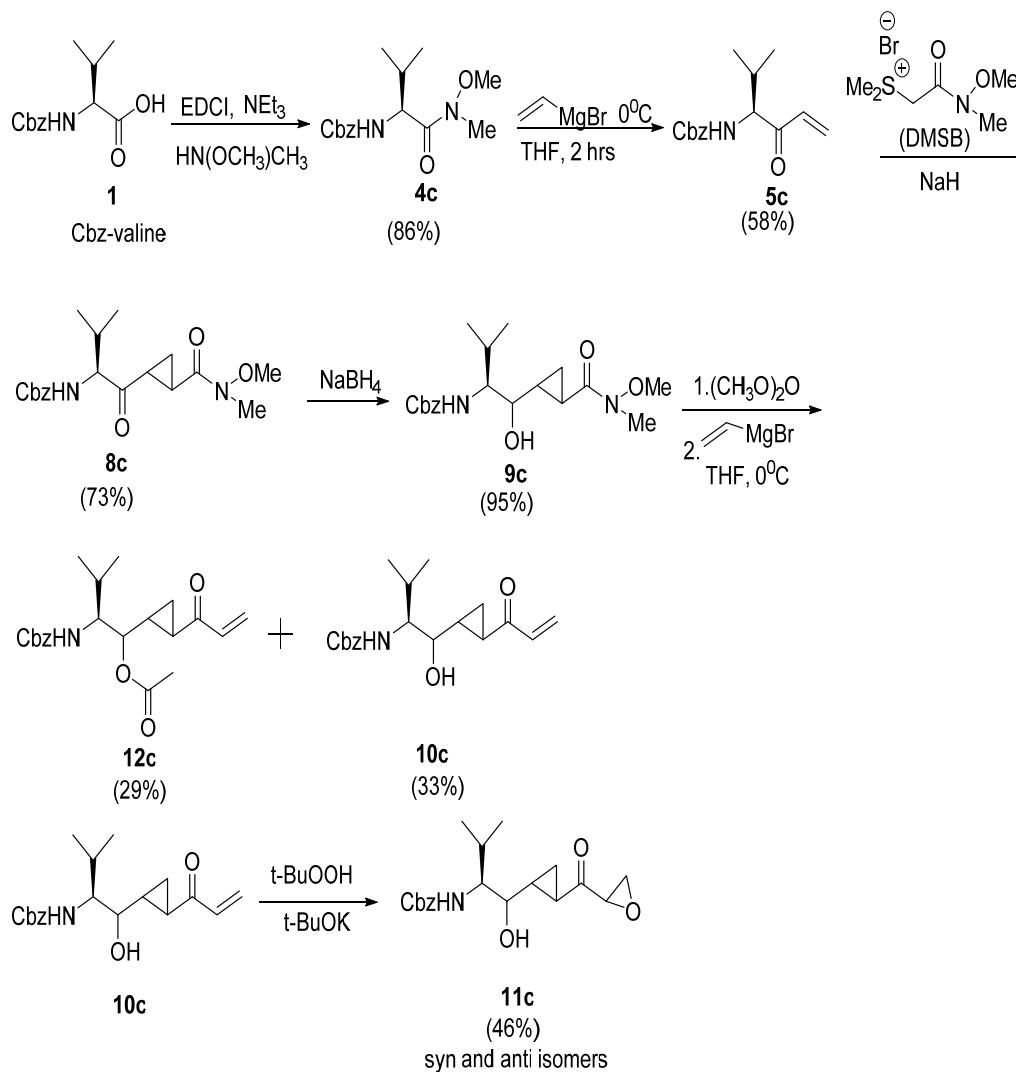
### 2.3 Synthesis of hybrid analog from Cbz-valine

The Cbz protected phenylalanine was converted to a Weinreb amide **4c**, followed by treatment with vinylmagnesium bromide to afford an enone **5c**. The enone was treated with DMSB salt and a base to form mixture of cyclopropyl ketone isomers **8c**. There were reduced by sodium borohydride to form mixture of cyclopropyl alcohols **9c**. The alcohol group was protected by acylation then treated with vinylmagnesium bromide to afford



enone **12c**, as well as deacylated alcohol **10c**. The enone was treated with t-butylperoxide in presence of base to afford a hybrid analog of belactosin A and carfilozomib **11c**. The route is outlined in **Scheme 5**.

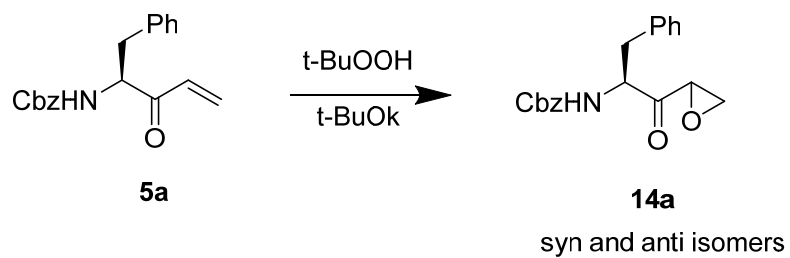
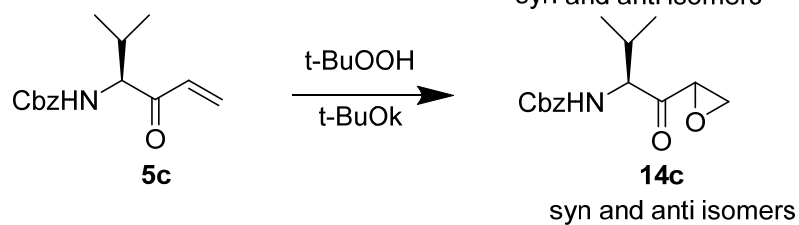
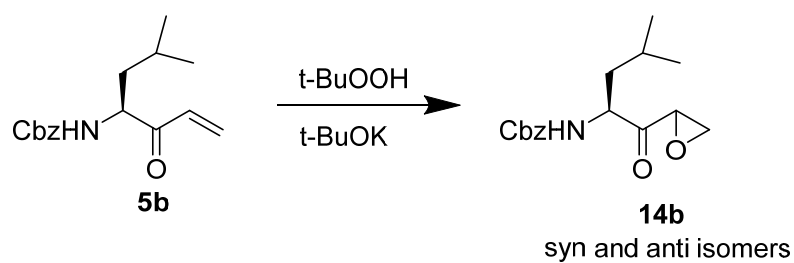
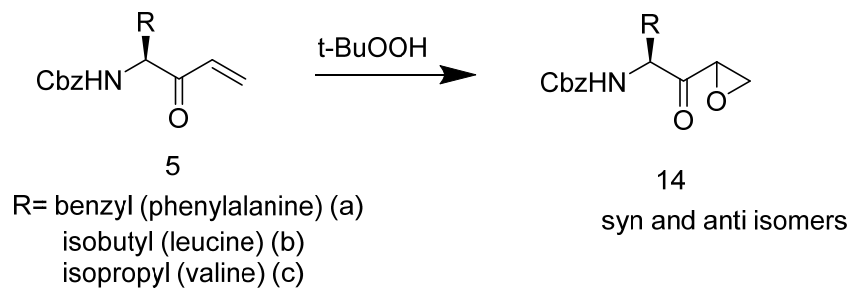
**Scheme 5:** Synthesis of hybrid analog of belactosin A and carfilzomib from Cbz- valine



### Simple epoxides

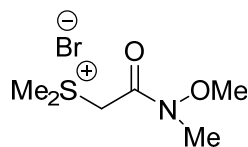
The synthesis of simple epoxides was accomplished by treating Cbz-protected enones **5a-c** with t-butyl hydrogen peroxide in the presence of the strong base potassium

tert-butoxide, These were prepared as model systems for the cyclopropyl epoxyketones hybrid analogs.



## Experimental procedures

### (2-(methoxy(methyl)amino)-2-oxoethyl)dimethylsulfonium bromide (DMSB) (3):

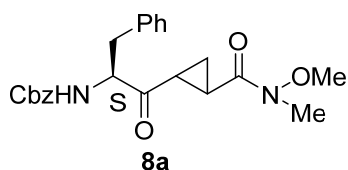


**Formation of 2-bromo-N-methoxy-N-methylamide:** To a stirred suspension of N,O-dimethylhydroxylamine hydrochloride (2.1 g, 21.5 mmol), and  $K_2CO_3$  (14.86 g, 107.70 mmol) in MeCN (80 mL), bromoacetyl bromide (4.32 mL, 49.54 mmol) was added dropwise at room temperature. After 80 minutes, the reaction mixture was poured into water and extracted twice with  $CH_2Cl_2$ . The organic layer was dried over anhydrous magnesium sulfate, filtered and the solvent was evaporated to afford 3.2 g (81.75 %) of the Weinreb amide. H-NMR (500 MHz,  $CDCl_3$ ):  $\delta$  3.70 (2H, s), 3.44 (3H, s), 2.86 (3H, s)

To a solution of bromo-N-methoxy-N-methylamide (3.2 g, 17.6 mmol) in acetone, dimethyl sulfide (1.93 mL, 26.41 mmol) was added, and stirred for 18 hours. The mixture was filtered and washed with acetone, the solid was dried under vacuum for 20 minutes to afford 3.22 g (75.01%) (3), mp 107-109 °C.

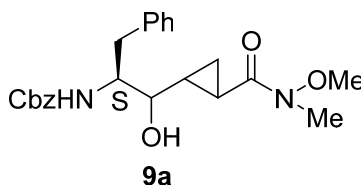
## 2.4 Cbz-phenylalanine series

### Benzyl(1-(2-(methoxy(methyl)carbamoyl)cyclopropyl)-1-oxo-3-Phenylpropan-2-yl)carbamate (8a):



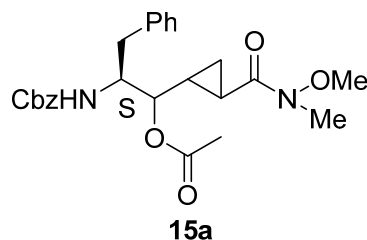
To a suspension of NaH (60%, in mineral oil, 370.0 mg, 9.32 mmol) in DMSO (11 mL), (2-(methoxy(methyl)amino)-2-oxoethyl) dimethylsulfonium bromide (2.27 g, 9.32 mmol) was added at room temperature, and was stirred for 25 minutes. The solution was cooled to 0°C, a solution of the (S)-enone **5a** (1.31 g, 4.23 mmol) in THF (11 ml) was added, and the mixture was stirred for 10 minutes. The creamy semisolid was dissolved in methylene chloride, and then poured into 1M HCl and extracted with methylene chloride thrice. The organic layer was washed with brine, dried with anhydrous magnesium sulfate and the solvent was evaporated. The crude product was purified by CombiFlash and evaporated to afford 1.25 g (72.1%) (**8a**) as a mixture of syn and antiisomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.34-7.08 (m, 10H, aryl), 5.71, 5.68 (2d, J=7.45, 7.45 Hz, 1H, NH), 5.06 (m, 2H, CbzCH<sub>2</sub>), 4.81 (m, 1H, CHN), 3.69, 3.67 (s, 3H, OCH<sub>3</sub>), 3.18, 3.17 (s, 3H, NCH<sub>3</sub>), 3.07 (m, 2H, CH<sub>2</sub>Ph), 2.8-2.69 (d, 1H, CHc<sub>pyl</sub>), 2.58 (m, 1H, CHc<sub>pyl</sub>), 1.5, 1.41 (m, 1H, CH<sub>2</sub>c<sub>pyl</sub>); <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ 206.04, 205.91 (C=O, ketones), 171.0 (C=O, amide), 156.73, 155.65 (CbzCH<sub>2</sub>), 136.5, 135.7(4°aryls), 129.51, 129.46, 128.62, 128.27, 128.16, 127.25 (10 aryl), 66.98 (CbzCH<sub>2</sub>), 61.92 (OCH<sub>3</sub>), 61.70, 61.58 (CHN'S), 37.35, 37.25 (CH<sub>2</sub>Ph), 32.63 (NCH<sub>3</sub>), 27.34, 27.19 (CHc<sub>pyl</sub>), 23.05, 22.90 (CHc<sub>pyl</sub>), 18.48, 18.03 (CH<sub>2</sub>c<sub>pyl</sub>); mass spectrum (ESI-MS) m/z (C<sub>23</sub>H<sub>27</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M+H) 411.1917, found 411.1914.

**Benzyl(1-hydroxy-1-(2-methoxy(methyl)carbamoyl)cyclopropyl)-3-phenylpropan-  
an-2-yl)carbamate (9a):**



To a solution of the ketone (**8a**) (1.18 g, 2.87 mmol) in methanol, NaBH<sub>4</sub> (200 mg, 5.23 mmol) was added and the mixture was stirred for 170 minutes at room temperature. The solution was poured in to 1M HCl and extracted with methylene chloride twice. The solution was washed with NaHCO<sub>3</sub> and brine, and was dried with anhydrous magnesium sulfate, filtered and the solvent evaporated. The crude was purified on CombiFlash and evaporated to afford 904.9 mg (76%) (**9a**) as a mixture of isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.3-7.18 (m, 10H, aryl), 5.83, 5.75, 5.71 (3d, J= 9.15, 9.15, 8.6 Hz, 1H, NH), 5.42, 5.38 (d, J=9.74 Hz, NH), 4.99 (m, 2H, CH<sub>2</sub>Cbz), 4.1, 3.9 (m, 1H, CHN), 3.69, 3.65 (s, 3H, OCH<sub>3</sub>), 3.52, 3.35 (m, 1H, CHO), 3.16, 3.14, 3.12 (s, 1H, NCH<sub>3</sub>), 2.99, 2.94, 2.79 (m, 2H, CH<sub>2</sub>Ph), 2.25, 2.15 (brs, 1H, OH), 1.64, 1.25, 1.17 (m, 1H, CH<sub>2</sub>cypyl), 1.03, 0.9, 0.65 (m, 2H, CH<sub>2</sub>cypyl); <sup>13</sup>C (125 MHz, CDCl<sub>3</sub>): δ 172.82 (C=O, amide), 155.99, 155.86, 155.67 (C=O, Cbz), 137.44, 137.3, 135.62, 135.52 (4° aryl C's), 128.18, 127.46, 126.99, 126.89, 126.83, 125.37 (aryl C's), 74.18, 73.23, 72.38, 71.48 (CHO), 65.62 (CH<sub>2</sub>Cbz), 60.70 (OCH<sub>3</sub>), 56.43, 56.15 (CHN), 36.90, 36.71, 34.82, 34.55 (CH<sub>2</sub>Ph), 31.56 (NCH<sub>3</sub>), 24.27, 23.92, 23.12, 22.56 (C<sub>2</sub>cypyl), 14.82, 14.12, 13.55 (C<sub>1</sub>cypyl), 11.37, 11.04 (CH<sub>2</sub>cypyl); Mass spectrum (ESI-MS) m/z (C<sub>23</sub>H<sub>29</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M+H) 413.2071, found 413.2071.

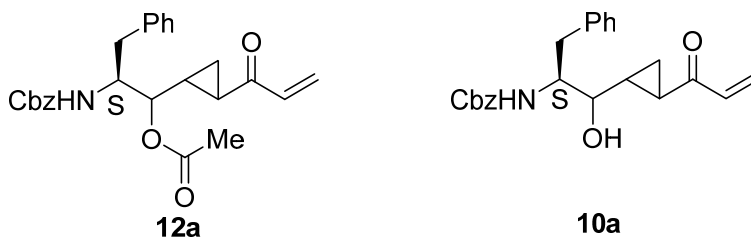
**Formation of (2S)-2-(((benzyloxy)carbonyl)amino)-1-(2-(methoxy(methyl)carbamoyl)cyclopropyl)-3-phenylpropyl acetate (**15a**):**



To a solution of (**9a**) (254 mg, 0.614 mmol) in 5 mL CH<sub>2</sub>Cl<sub>2</sub>, acetic anhydride (0.48 mL, 4.30 mmol) and pyridine (0.35 mL, 4.3 mmol) were added. The resulting solution was stirred for 36 hours, and then poured into 1M HCl and extracted with ethyl acetate twice. The organic layer was washed with aqueous NaHCO<sub>3</sub> and dried over magnesium sulfate, filtered and the solvent was evaporated. The crude product was purified on flash column chromatography on silica gel eluting with 1:3, 1:1, and 2:1 ethyl acetate-hexane to afford 248.6 mg (88.38%) (**15a**) as a mixture. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.28-7.17 (m, 10H, aryl), 5.64, 5.42, 5.32, 5.29 (4d, J= 9.15, 9.7, 10.3, 10.3 Hz, 18.3 Hz, 12.6Hz, 1H, NH), 5.06-4.83 (m, 2H, CH<sub>2</sub>Cbz), 4.67, 4.5, 4.34 (m, 1H, CHO), 4.28, 4.19 (m, 1H, CHN), 3.69, 3.65, 3.62 (s, 3H, OCH<sub>3</sub>), 3.15, 3.14, 3.11, 3.1 (s, 3H, NCH<sub>3</sub>), 2.96, 2.75 (m, 2H, CH<sub>2</sub>Ph), 2.28-2.2 (d, 1H, CHc<sub>pyl</sub>), 2.01, 1.99 (s, 3H, CH<sub>3</sub>), 1.72 (m, 1H, CHc<sub>pyl</sub>), 1.29, 1.2, 1.01, 0.91, 0.71 (m, 2H, CH<sub>2</sub>c<sub>pyl</sub>); <sup>13</sup>C NMR (125Hz, CDCl<sub>3</sub>): δ 173.19, 172.95 (C=O, ester), 170.71, 170.68, 170.31 (C=O, amide), 156.29, 156.11, 156.06 (C=O, Cbz), 137.60, 137.4, 137.22, 136.73, 136.54 (4° aryl), 129.27, 129.09, 128.54, 128.19, 127.99, 127.91, 127.81, 126.75, 126.62 (6 aryl C's), 76.7, 76.41, 75.54 (CHO), 66.85, 66.49 (CH<sub>2</sub>Cbz), 61.79, 61.73, 61.65 (OCH<sub>3</sub>), 55.55, 55.41, 54.74, 54.46 (CHN), 38.71, 38.39, 36.44, 36.20

(CH<sub>2</sub>Ph), 32.56 (NCH<sub>3</sub>), 23.34, 22.85, 21.74, 21.11 (CHc<sub>pyl</sub>), 21.05 (CH<sub>3</sub>CO), 15.79, 15.58 (CHc<sub>pyl</sub>), 12.92, 12.66, 12.59 (CH<sub>2</sub>c<sub>pyl</sub>) Mass spectrum (ESI-MS) m/z (C<sub>25</sub>H<sub>31</sub>N<sub>2</sub>O<sub>6</sub>) calculated for (M+H) 455.2178, found 455.2177.

**Formation of benzyl ((2S)-1-(2-acryloylcyclopropyl)-1-hydroxy-3-phenylpropan-2-yl) carbamate (10a) and ((2S)-1-(2-acryloylcyclopropyl)-2-(((benzyloxy)carbonyl)amino)-3-phenylpropyl acetate, (12a):**



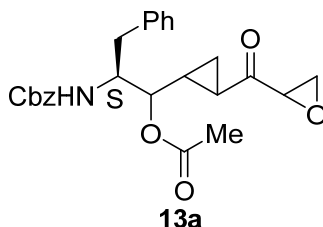
To a solution of (**15a**) (110 mg, 0.24 mmol) in THF (1.5 mL), at 0°C under argon, vinylmagnesium bromide (0.61 ml of 1M solution) was added, and stirred for 1 hour, and again the same amount of vinylmagnesium bromide was added. After 3 hours, the resulting solution was poured into 1M HCl, and extracted with ethyl acetate thrice. The organic layer was washed with NaHCO<sub>3</sub>, dried over anhydrous magnesium sulfate, filtered and the solvent evaporated. The crude product was purified by flash column chromatography on silica gel eluting with 1:10, 1:5, 1:3, and 1:2 ethyl acetate-hexane to afford 31.49 mg (28.5%) (**10a**) and 19.2 mg (19.2%) (**12a**) as a mixture of isomers.

**COMPOUND (10a):**  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.33-7.18 (m, 10H, aryl), 6.44 (m, 1H, CH alkene), 6.3 (m, 2H,  $\text{CH}_2$  trans alkene), 5.85 (m, 2H,  $\text{CH}_2$  cis alkene), 5.00 (m, 3H, NH and Cbz  $\text{CH}_2$ ) alkene), 4.03 (s, 1H, CHN), 3.63, 3.45 (s, 1H, CHO), 3.18, 3.09 (s, 1H, OH), 2.96, 2.81 (m, 2H,  $\text{CH}_2\text{Ph}$ ), 2.34 (m, 1H, CHcypyl), 1.68 (m, 1H, CHcypyl), 1.28, 1.12, 1.02, 0.88 (m, 2H,  $\text{CH}_2\text{cypyl}$ );  $^{13}\text{C}$  (125 Mz,  $\text{CDCl}_3$ ):  $\delta$  200.41 (C=O, ketone), 157.23 (C=O,Cbz), 137.66, 136.65, 136.15 (CH alkene and  $4^\circ$  aryl C's), 129.12, 128.11, 128.5 (CH alkene and aryl C's), 74.90, 74.4 (CHO), 66.86 ( $\text{CH}_2\text{Cbz}$ ), 57.83, 57.32 (CHN), 35.74, 29.71 ( $\text{CH}_2\text{Ph}$ ), 27.09, 26.7 (CHcypyl), 23.19, 23.18 (CHcypyl), 15.16, 14.65 ( $\text{CH}_2\text{cypyl}$ ); Mass spectrum (ESI-MS)  $m/z$  ( $\text{C}_{23}\text{H}_{26}\text{NO}_4$ ) calculated for (M+H) 380.1849, found 380.1856.

**COMPOUND (12a):**  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.34-7.14 (m, 10H, aryl), 6.42 (m, 1H, CH alkene), 6.24 (m, 2H,  $\text{CH}_2$  trans alkene), 5.8 (m, 2H,  $\text{CH}_2$  cis alkene) 5.43, 5.3 (m, 1H, NH), 4.98 (m, 2H,  $\text{CH}_2\text{Cbz}$ ), 4.66, 4.53, 4.35 (m, 1H, CHO), 4.26, 4.19 (m, 1H, CHN), 2.95, 2.71 (m, 2H,  $\text{CH}_2\text{Ph}$ ), 2.42, 2.25 (m, 1H, CHcypyl), 2.02, 1.99 (s, 3H,  $\text{CH}_3$ ), 1.76 (m, 1H, CHcypyl), 1.25, 1.1, 1.02, 0.88, 0.81 (m, 2H,  $\text{CH}_2\text{cypyl}$ );  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ ):  $\delta$  198.91, 198.69, 198.27, 198.15 (C=O, ketone), 170.35, 170.30, 169.98, 169.88 (C=O, ester), 155.67 (C=O, Cbz), 136.99, 136.51, 136.07 (CH alkene,  $4^\circ$  aryl C's), 128.8, 128.67, 128.31, 128.23, 127.73, 127.62, 127.48, 127.42, 126.38, 126.26 ( $\text{CH}_2$  alkene, 6 aryl C's), 76.26, 75.83, 75.24, 74.97, 73.78, 72.1 (CHO), 66.46, 66.4, 66.13 ( $\text{CH}_2\text{Cbz}$ ), 54.93, 54.35, 54.15 (CHN), 38.29, 37.95, 35.84, 35.78 ( $\text{CH}_2\text{Ph}$ ), 32.53, 31.30, 30.53, 29.93, 29.33 (CHcypyl), 26.45, 25.46, 25.29, 24.38, 23.8 (CHcypyl), 23.05, 22.83, 20.68, 20.54 (acetate  $\text{CH}_3$ ), 15.63, 15.40, 15.56, 15.56 ( $\text{CH}_2\text{cypyl}$ ); Mass spectrum (ESI-MS)  $m/z$  ( $\text{C}_{25}\text{H}_{28}\text{NO}_5$ ) calculated for (M+H) 422.1950, found 422.1962.



**Formation of (2S)-2-(((benzyloxy)carbonyl)amino)-1-(2-(oxirane-2-carbonyl)cyclopropyl)-3-phenylpropyl acetate (**13a**):**

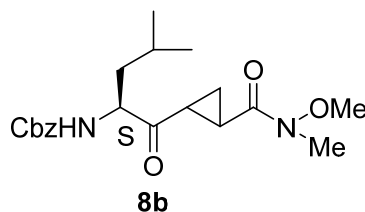


A solution of potassium tert-butoxide in tetrahydrofuran (1.0 M, 0.73  $\mu$ L, 0.806 mmol) was added to a stirred solution of (**12a**) (50 mg, 0.12 mmol) and tert-butyl hydroperoxide in hexane (26.46  $\mu$ L, 0.237 mmol) in tetrahydrofuran (1.64 mL) at 0°C. To the reacting mixture the same amount of tert-butyl hydroperoxide was added after one and three hours, and the same amount of potassium tert-butoxide was added every 1 hour over 8 h. The reacting mixture was maintained at 0°C for another 13 h, and solid then sodium sulfite (75 mg, 0.6 mmol) was added. The resulting mixture was stirred at room temperature for an hour. The crude product was dissolved in ethyl acetate (15 mL), and filtered, evaporated, then purified by flash column chromatography on silica gel eluting with 1:10, 1:5 and 1:3 ethyl acetate-hexane to afford 31 mg (59.61%) (**13a**) as a mixture of isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.36-7.20 (m, 10H, aryl) 5.06-4.96 (m, 2H, CbzCH<sub>2</sub>), 4.88, 4.78 (2d, J= 10, 10 Hz, 1H, NH), 4.71, 4.57, 4.35 (m, 1H, CHO), 4.26, 4.16, 4.21 (m, 1H, CHN), 3.41 (m, CH epoxide), 3.01 (m, 2H, CH<sub>2</sub> epoxide), 2.75 (m, 2H, CH<sub>2</sub>Ph), 2.07, 2.06, 2.05, 2.04 (s, 3H, acetate CH<sub>3</sub>), 1.88, 1.74, 1.66, 1.25 (m, 1H, CHc<sub>pyl</sub>), 0.86 (m, 2H, CH<sub>2</sub>c<sub>pyl</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  204.91, 204.73, 204.57 (C=O, ketone), 169.52, 169.12 (C=O, ester), 154.93, 154.79 (C=O, Cbz), 135.96, 135.4, 135.29 (4° aryl C's), 128.06, 127.68,

127.57, 127.14, 126.3 (aryl C's), 75.36, 75.07, 74.43, 74.32 (CHO), 66.04, 65.85 (CH<sub>2</sub>Cbz), 54.3, 54.1, 53.83, 53.32, 52.78, 52.66 (CH epoxide, CHN), 44.87, 44.63 (CH<sub>2</sub> epoxide), 37.9, 37.47, 35.36, 35.22 (CH<sub>2</sub>Ph), 26.12, 25.57, 25.25, 24.94, 24.78, 24.35, 23.99 (CHcpyl), 20.79, 19.98, 19.8, 19.26, 18.75, 18.03 (CHcpyl and acetate CH<sub>3</sub>), 16.01, 15.08, 14.92, 14.24, 13.88 (CH<sub>2</sub>cpyl); mass spectrum (ESI-MS) m/z (C<sub>25</sub>H<sub>28</sub>NO<sub>6</sub>) calculated for (M+H) 438.1907, found 438.1911.

## 2.5 Cbz-leucine series

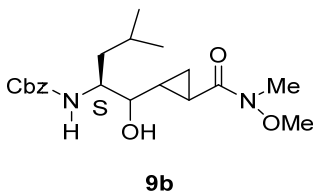
**Formation of benzyl ((2S)-1-(2-(methoxy(methyl)carbamoyl)cyclopropyl)-4-methyl-1-oxopentan-2-yl) carbamate (8b):**



To a suspension of NaH (60% in mineral oil, 52.0 mg, 1.28 mmol) in DMSO (1.5 mL), DMSB (310.0 mg, 1.27 mmol) was added at room temperature, and was stirred for 25 minutes. The solution was cooled to 0°C, a solution of (**5b**) (160 mg, 0.58 mmol) in THF (1.6 ml) was added, and the mixture was stirred for 20 minutes. The creamy semisolid was dissolved in methylene chloride, and then poured in to 1M HCl and extracted with methylene chloride thrice. The organic layer was washed with brine, dried with anhydrous magnesium sulfate and the solvent was evaporated. The crude product was purified by CombiFlash and evaporated to afford 148 mg (67.62%) (**8b**) as a mixture of syn and anti-isomers. <sup>1</sup>H-NMR (500 Mhz, CDCl<sub>3</sub>): δ 7.37-7.35 (m, 5H, aryl isomers), 5.29 (d, J=8 Hz,

1H, NH isomers), 5.1, 5.09 (s, 2H, CbzCH<sub>2</sub> isomers), 4.61 (d, J=14.9 Hz, 1H, CHN isomers), 3.72 (s, 3H, OCH<sub>3</sub> isomers), 3.21 (s, 3H, CH<sub>3</sub> isomers), 2.56, 2.50 (m, 1H, CHc<sub>pyl</sub> isomers), 1.76-1.72 (m, 2H, isobutyl CH and one H of isobutyl CH<sub>2</sub> isomers), 1.54, 1.50 (m, 2H, CHc<sub>pyl</sub> and one H of CH<sub>2</sub>c<sub>pyl</sub> isomers), 1.43-1.39 (m, 2H, one H of isobutyl CH<sub>2</sub> and one H of CH<sub>2</sub> c<sub>pyl</sub>), 1.01 (t, J=4.6 Hz, 3H, isobutyl CH<sub>3</sub>), 0.93, 0.92 (d, J=6.3, 6.85 Hz, 3H, isobutyl CH<sub>3</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 207.48 (C=O, ketone), 171.12 (C=O, amide), 156.13, 155.6 (C=O, Cbz), 136.40, 136.17 (4°aryl C), 128.57, 128.18, 128.09 (aryl C's), 66.95 (CH<sub>2</sub>Cbz), 61.87, 61.82 (OCH<sub>3</sub>), 59.49, 59.32 (CHN), 40.61, 40.51 (isobutyl CH<sub>2</sub>), 32.57 (NCH<sub>3</sub>), 26.94, 26.55 (CHc<sub>pyl</sub>), 24.95, 24.91 (isobutyl CH), 23.42, 23.35 (CH<sub>3</sub>), 22.42 (CHc<sub>pyl</sub>), 21.75 (CH<sub>3</sub>), 17.84, 17.72 (CH<sub>2</sub> c<sub>pyl</sub>); mass spectrum (ESI-MS) m/z (C<sub>20</sub>H<sub>29</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M+H) 377.2053, found 377.2071.

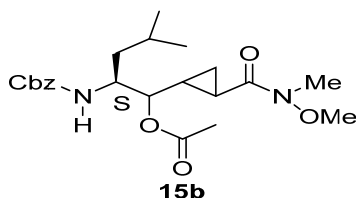
**Formation of benzyl ((2S)-1-hydroxy-1-(-2-(methoxy(methyl)carbamoyl)cyclopropyl)-4-methylpentan-2-yl)carbamate (9b):**



To a solution of (**8b**) (129 mg, 0.34 mmol) in methanol (5 mL), NaBH<sub>4</sub> (26 mg, 0.68 mmol) added and the mixture was stirred for 120 minutes at room temperature. The solution was poured in to 1M HCl and extracted with ethyl acetate twice. The solution was washed with NaHCO<sub>3</sub> and brine, and was dried with anhydrous magnesium sulfate, filtered and the solvent evaporated. The crude was purified on CombiFlash and the solvent evaporated to afford 117.4 mg (90%) (**9b**) as a mixture of isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): <sup>1</sup>H-

NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.34-7.27 (m, 5H, aryl isomers), 5.33, 5.28, 5.24 (d, J=8.6, 4.6, 12.95 Hz, 1H, NH isomers), 5.12-5.4 (m, 2H, Cbz CH<sub>2</sub> isomers), 3.89 (m, 1H, CHN isomers), 3.74, 3.72 (d, J=3.6, 12.05 Hz, 3H, OCH<sub>3</sub> isomers), 3.45, 3.29 (d, J=4.0, 5.15 Hz, 1H, CHO isomers), 3.17, 3.16 (s, 3H, NCH<sub>3</sub> isomers), 2.20, 2.15 (br s, 1H, OH isomers), 1.65-1.63 (m, 1H, isobutyl CH isomers), 1.5-1.43 (m, 2H, CHcypyl and one H of isobutyl CH<sub>2</sub>), 1.34-1.29 (m, 1H, isobutyl CH isomers), 1.2-1.18 (m, 2H, CH<sub>2</sub>cypyl isomers), 1.18-1.14 (m, 1H, CHcypyl isomers), 0.9, 0.86 (s, 6H, isobutyl CH<sub>3</sub> isomers); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  173.88 (C=O, amide), 157.36, 157.31 (C=O, Cbz), 136.6, 136.44 (4° aryl C's), 128.58, 128.21, 128.08 (aryl C's), 76.93, 75.26 (CHO), 66.98, 66.94 (CH<sub>2</sub>Cbz), 61.67, (OCH<sub>3</sub>), 54.45, 54.36, 53.92 (CHN), 39.98, 38.67 (isobutyl CH<sub>2</sub>), 32.05 (CH<sub>3</sub>N), 25.28, 24.83, 24.67 (isobutyl CH), 23.69, 23.61, 23.30, 23.13 (CHcypyl), 22.13, 21.75, 21.65 (CH<sub>3</sub>), 14.26, 14.17 (CHcypyl), 11.86, 11.56 (CH<sub>2</sub>cypyl); mass spectrum (ESI-MS) m/z (C<sub>20</sub>H<sub>31</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M+H) 379.2233, found 379.2227.

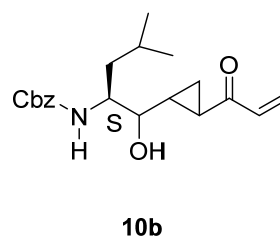
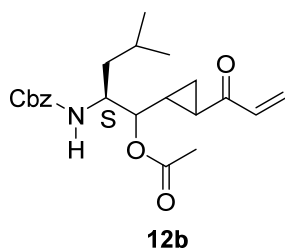
**Formation of (2S)-2-((benzyloxy)carbonyl)amino)-(2(methoxy(methyl) carbamoyl)cyclopropyl)-4-methylpentyl acetate (15b):**



To a solution of (**9b**) (104 mg, 0.275 mmol) in 3 mL CH<sub>2</sub>Cl<sub>2</sub>, acetic anhydride (0.182 mL, 1.92 mmol) and pyridine (0.166 mL, 2.061 mmol) were added. The resulting solution was stirred for 48 hours at room temperature, and then poured into 1M HCl and extracted with ethyl acetate twice. The organic layer was washed with aqueous NaHCO<sub>3</sub> and dried over

magnesium sulfate, filtered and the solvent was evaporated. The crude product was purified on flash column chromatography on silica gel eluting with 1:20, 1:10, 1:5, 1:2, 1:1, 2:1 and 3:1 ethyl acetate-hexane to afford 248.6 mg (94.89%) (**15b**). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.35-7.33 (m, 5H, aryl isomers), 5.10-5.06 (m, 2H, Cbz CH<sub>2</sub>), 4.95, 4.89 (d, J=9.7, 10.0 Hz, 1H, NH isomers), 4.55, 4.49, 4.44, 4.36 (dd, J=11.45 Hz, 1H, CHO isomers), 4.06-4.02 (m, 1H, CHN isomers), 3.75, 3.68 (s, 3H, OCH<sub>3</sub> isomers), 3.18 (t, J=15.45 Hz, 3H, NCH<sub>3</sub> isomers), 2.03, 2.00 (s, 3H, acetate CH<sub>3</sub> isomers), 1.68-1.64 (m, 1H, isobutyl CH isomers), 1.61-1.57 (m, 1H, CHcypyl isomers), 1.41-1.36 (m, 2H, isobutyl CH<sub>2</sub> isomers), 1.26, 1.19 (m, 1H, CHcypyl isomers), 1.01 (m, 2H, CH<sub>2</sub>cypyl isomers), 0.9 (m, 6H, isobutyl CH<sub>3</sub> isomers): <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 173.24, 173.09 (C=O, ester), 170.71, 170.65, 170.5, 170.41 (C=O, amide), 156.16 (C=O, Cbz), 136.64, 136.55 (4° aryl C's), 128.6, 128.54, 128.27, 128.22, 128.1, 128.98 (aryl C's), 76.96, 76.27 (CHO), 66.89, 66.71 (CH<sub>2</sub>Cbz), 61.78, 61.62 (OCH<sub>3</sub>), 52.19, 51.87, 55.77 (CHN), 41.52, 39.13, 39.02 (isopropyl CH<sub>2</sub>), 32.58 (CH<sub>3</sub>N), 24.76, 24.64 (CHcypyl), 23.58, 23.54 (CH<sub>3</sub>), 23.21, 22.39 (CHcypyl), 21.63, 20.89 (acetate CH<sub>3</sub>), 21.54, 21.09 (CHcypyl) 15.45 (isobutyl CH), 12.73, 12.33 (CH<sub>2</sub>cypyl); mass spectrum (ESI-MS) m/z (C<sub>22</sub>H<sub>32</sub>N<sub>2</sub>O<sub>6</sub>) calculated for (M) 420.2260.

**Formation of (2S)-1-(2-acryloylcyclopropyl)-2-(((benzyloxy)carbonyl)amino)-4-methylpentyl acetate (12b) and benzyl ((2S)-1-(2-acryloylcyclopropyl)-1-hydroxy-4-methylpentan-2-yl) carbamate (10b):**

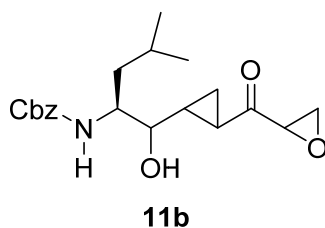


To a solution of (**15b**) (151 mg, 0.36 mmol) in THF (2.0 mL), at 0°C under argon, vinylmagnesium bromide (0.9 mL of 1M solution) was added, and stirred for 1 hour and gradually warmed to RT and again same amount of vinylmagnesium bromide was added after 2 and 4.0 hrs. After 5 hours, the resulting solution was poured into 1M HCl, and extracted twice with ethyl acetate. The organic layer was washed with NaHCO<sub>3</sub>, dried over anhydrous magnesium sulfate, filtered and the solvent evaporated. The crude product was purified by flash column chromatography on silica gel eluting with 1:10, 1:5, 1:2, and 1:1 ethyl acetate-hexane to afford and 64.3 mg (51.85%) (**10b**) and 24 mg (17.15%) (**12b**) as a mixture of isomers.

**Compound (10b)** (alcohol) : <sup>1</sup>H-NMR (500 MHz, CD<sub>3</sub>OD): δ 7.33-7.26 (m, 5H, aryl isomers), 6.47-6.4 (m, 1H, vinyl CH isomers), 6.35, 6.34, 6.30 (d, J=1.15 Hz, 1H, vinyl CH<sub>2</sub> trans isomers), 5.88-5.82 (m, 1H, vinyl CH<sub>2</sub> trans isomers), 5.08-4.94 (m, 2H, Cbz CH<sub>2</sub> isomers and 1H, NH isomers), 3.68-3.65 (m, 1H, CHN isomers), 3.06-3.04 (m, 1H, CHO isomers), 2.35-2.31 (m, 1H, CHc<sub>pyl</sub> isomers), 1.66-1.58 (m, 2H, CHc<sub>pyl</sub> isomers and isobutyl CH isomers), 1.45-1.38 (m, 2H, isobutyl CH<sub>2</sub> isomers), 1.2-1.17 (m, 2H, CH<sub>2</sub> c<sub>pyl</sub> isomers), 1.05-1.02 (m, 2H, CH<sub>2</sub>c<sub>pyl</sub> isomers), 0.9-0.82 (m, 6H, isobutyl CH<sub>3</sub> isomers); <sup>13</sup>C-NMR (125 MHz, CD<sub>3</sub>OD): δ 200.45, 199.87 (C=O, ketone), 156.86, 156.81 (C=O, Cbz), 135.95, (alkene, CH), 135.85 (4°aryl C), 127.55, 126.79 (aryl C's), 126.91 (alkene CH<sub>2</sub>), 75.30, 75.18 (CHO), 65.44 (CH<sub>2</sub>Cbz), 53.56 (CHN), 38.47, 38.39 (isobutane CH<sub>2</sub>), 28.32, 27.75 (isobutyl CH), 24.03, 24.01 (CHc<sub>pyl</sub>), 22.74, 23.37, 22.28 (CHc<sub>pyl</sub>), 19.96 (CH<sub>3</sub>), 14.06, 13.88 (CH<sub>2</sub>c<sub>pyl</sub>); mass spectrum (ESI-MS) m/z (C<sub>20</sub>H<sub>28</sub>NO<sub>4</sub>) calculated for (M+H) 346.2009, found 346.2013.

**Compound (12b):**  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.37-7.33 (m, 5H, aryl isomers), 6.48-6.34 (m, 1H, CH alkene isomers), 6.30-6.24 (m, 2H,  $\text{CH}_2$  alkene trans isomers), 5.85 (m, 2H,  $\text{CH}_2$  alkene cis isomers), 5.07, 4.70 (m, 3H, one H of NH and two H of Cbz isomers), 4.56, 4.53, 4.38 (dd,  $J=3.45, 2.85$  Hz, 1H, CHO isomers), 4.0 (m, 1H, CHN isomers), 2.44, 2.25, 2.16 (m, 1H, CHcypyl isomers), 2.03 (m, 3H, acetate  $\text{CH}_3$  isomers), 1.72, 1.64 (m, 2H, one H of isobutyl CH and one of H of CHcypyl isomers), 1.29 (m, 2H, isobutyl  $\text{CH}_2$ ), 1.16, 1.12 (m, 2H,  $\text{CH}_2$  cpyl), 0.9-0.85 (m, 6H, isobutyl  $\text{CH}_3$ );  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ ):  $\delta$  197.72, 197.38, 197.19 (C=O, ketone), 169.39, 169.16 (C=O, ester), 154.83, 154.78 (C=O, amide), 135.33 (CH alkene), 135.14 ( $4^\circ$  aryl), 127.30 ( $\text{CH}_2$  alkene), 126.99, 126.95, 126.83, 126.76 (aryl C's), 77.02, 76.50 (CHO), 65.63, 65.58 ( $\text{CH}_2$  Cbz), 50.87, 50.60 (CHN), 40.31, 39.96, 37.59 ( $\text{CH}_2$  isobutyl), 28.45 (CH isobutyl), 25.38, 24.39, 24.11, 23.3, 22.64, 22.26, 21.77 (CHcypyl), 20.77, 20.24 ( $\text{CH}_3$  acetate), 19.80, 19.60 ( $\text{CH}_3$  isobutyl), 14.52, 14.39 ( $\text{CH}_2$  cpyl); mass spectrum (ESI-MS)  $m/z$  ( $\text{C}_{22}\text{H}_{29}\text{NO}_5$ ) calculated for (M) 387.2046.

**Formation of benzyl ((2S)-1-hydroxy-4-methyl-1-(2-(oxirane-2-carbonyl)cyclopropyl pentan-2-yl)carbamate (11b):**



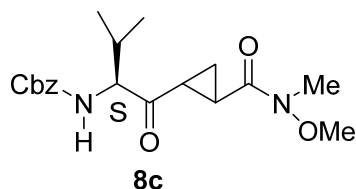
A solution of potassium tert-butoxide in tetrahydrofuran (1.0 M, 65  $\mu\text{L}$ , 0.52 mmol) was added to a stirred solution of (**10b**) (60 mg, 0.17 mmol) and tert-butyl hydroperoxide in hexane (67  $\mu\text{L}$ , 0.87 mmol) in tetrahydrofuran (2.0 mL) at  $0^\circ\text{C}$ . To the reacting mixture,

the same amount of tert-butyl hydroperoxide and the same amount of potassium tert-butoxide were added in every 1 hr for another 6 hrs, and the reacting mixture was maintained at 0°C for another 14 hrs. The resulting mixture was stirred at room temperature by adding solid sodium sulfite (220 mg, 1.74 mmol) for 30 minutes. The crude product was dissolved in ethyl acetate (15 mL), and filtered, evaporated, then purified by flash column chromatography on silica gel eluting with 1:10, 1:5, 1:4, and 1:2 ethyl acetate-hexane to afford 35.8 mg (57.02%) (**11b**) as a mixture of isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.36-7.34 (m, 5H, aryl isomers), 5.10-5.08 (m, 2H, Cbz CH<sub>2</sub> isomers), 4.89 (t apparently (real dd, J=8.6, 8.0 Hz), 1H, NH isomers), 3.87-3.85 (m, 1H, CHN isomers), 3.48 (m, 1H, CHO isomers), 3.47-3.45 (m, 1H, CH epoxide isomers), 3.02, 2.96 (m, 2H, CH<sub>2</sub> epoxide isomers), 2.15, 2.14 (1H, OH), 2.09-2.02 (m, 1H, isobutyl CH isomers), 1.65, 1.59 (m, 2H, CHcpyl isomers), 1.40 (m, 2H, one H of isobutyl CH<sub>2</sub> and one H of CH<sub>2</sub>cpyl isomers), 1.28 (m, 2H, one H of isobutyl CH<sub>2</sub> and one H of CH<sub>2</sub>cpyl isomers), 0.97-0.84 (m, 6H, isobutyl CH<sub>3</sub> isomers); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 206.87, 206.42 (C=O, ketone), 157.53, 157.37 (C=O, Cbz), 136.52, 136.22 (4° aryl C), 128.64, 128.36, 128.16 (aryl C's), 74.58, 74.13 (CHO), 67.24, 67.08 (CH<sub>2</sub>Cbz), 54.66, 54.52, 54.18 (CHN), 53.85, 53.68 (CHepoxide) 46.13, 45.92 (CH<sub>2</sub>epoxide), 39.18, 38.73 (isobutyl CH<sub>2</sub>), 24.93 (CHcpyl), 23.54 (CHcpyl), 21.64 (CH<sub>3</sub>), 19.79, 18.90 (isobutyl CH), 15.03, 14.20 (CH<sub>2</sub>cpyl); mass spectrum (ESI-MS) m/z (C.H.N<sub>2</sub>O<sub>5</sub>) calculated for (M) 403.1995.



## 2.6 Cbz-valine series

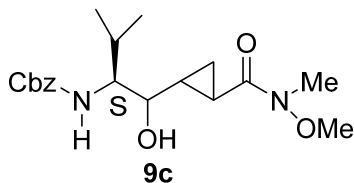
### Formation of benzyl ((2S)-1-(2-(methoxy(methyl)carbamoyl)cyclopropyl)-3-methyl-1-oxobutan-2-yl)carbamate (**8c**):



To a suspension of NaH (60% in mineral oil, 157.0 mg, 3.92 mmol) in DMSO (5.0 mL), DMSB (957.0 mg, 3.92 mmol) was added at room temperature, and was stirred for 25 minutes. The solution was cooled to 0°C, a solution of (**5c**) (465.9 mg, 1.78 mmol) in THF (5.0 ml) was added, and the mixture was stirred for 15 minutes. The creamy semisolid was dissolved in ethyl acetate and pure into 1M HCl and extracted with ethyl acetate. The organic layer was washed with NaHCO<sub>3</sub>, dried with anhydrous magnesium sulfate and the solvent was evaporated. The crude product was purified by CombiFlash and the solvent evaporated to afford 472.6 mg (73.16%) (**8c**). <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.37-7.29 (m, 5H, aryl isomers), 5.54-5.50 (t apparently (real d, J= 8.6, 8.6 Hz), 1H, NH), 5.09 (s, 2H, CbzCH<sub>2</sub>), 4.59, 4.58 (dd, J= 4.0, 4.0 Hz, 1H, CHN isomers), 3.70, 3.69 (s, 3H, OCH<sub>3</sub> isomers), 3.2 (s, 3H, NCH<sub>3</sub> isomers), 2.59-2.54 (m, 1H, CHcpyl isomers), 2.39, 2.33 (m, 1H, isopropyl CH isomers), 1.50 (m, 1H, CHcpyl), 1.40 (m, 2H, CH<sub>2</sub>cpyl isomers), 1.04 (t, J= 14.35 Hz, 3H, isopropyl CH<sub>3</sub>), 0.78 (t, J= 16.5, 3H, isopropyl CH<sub>3</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 206.87 (C=O, ketone), 171.02 (C=O, amide), 156.50, 155.8 (C=O, Cbz), 136.49 (4° aryl C), 128.53, 128.09 (aryl C's), 66.84, 66.41 (CH<sub>2</sub>Cbz), 65.74, 65.63 (CHN), 61.74 (OCH<sub>3</sub>), 31.49 (NCH<sub>3</sub>), 30.01 (isopropyl CH), 27.52, 27.18 (CHcpyl), 22.66, 22.22

(CHcpyl), 19.8 (CH<sub>3</sub>), 17.6, 17.27 (CH<sub>2</sub> cpyl), 16.9 (CH<sub>3</sub>); mass spectrum (ESI-MS) m/z (C<sub>19</sub>H<sub>27</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M+H) 363.1906, found 363.1914.

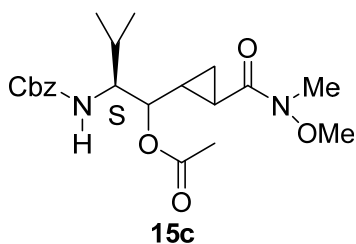
**Formation of benzyl ((2S)-1-hydroxy-1-(2-(methoxy(methyl)carbamoyl)cyclopropyl)-3-methylbutan-2-yl)-3-methylbutan-2-yl)carbamate (9c):**



To a solution of (**8c**) (391 mg, 1.08 mmol) in methanol (13 mL), NaBH<sub>4</sub> (75 mg, 0.075 mmol) added and the mixture was stirred for 2 hrs at room temperature. The solution was poured in to 1M HCl and extracted twice with ethyl acetate. The organic layer was washed with NaHCO<sub>3</sub> and brine, and was dried with anhydrous magnesium sulfate, filtered and the solvent evaporated. The crude was purified on CombiFlash and the solvent evaporated to afford 374.1 mg (95.19 %) (**9c**) as a mixture of isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.25-7.19 (m, 5H, aryl isomers), 5.51 (t (apparently), real 2d, J= 9.15, 9.15 Hz 1H, NH), 5.03-4.94 (m, 2H, CH<sub>2</sub>Cbz isomers), 3.61 (s, 3H, CH<sub>3</sub> isomers), 3.57- 3.52 (m, 1H, CHN and CHO isomers), 3.38 (m, 1H, CHO isomers), 3.06, 3.04 (s, 3H, NCH<sub>3</sub> isomers), 2.16, 2.01 (br s, 1H, OH), 1.91 (m, 1H, isopropyl CH isomers), 1.49, 1.44 (m, 1H, CHcpyl isomers), 1.08, 1.02 (m, 1H, CHcpyl isomers), 0.94 (m, CH<sub>2</sub>cpyl isomers), 0.87 (d, J= 6.85 Hz, 3H, isopropyl CH<sub>3</sub>), 0.84 (t, J= 10.9 Hz, 3H, isopropyl CH<sub>3</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 174.02, 173.38, 171.24 (C=O, amide), 157.6, 157.19, 157.15 (C=O, Cbz), 136.82, 136.64 (4° aryl C's), 128.48, 128.06, 127.97, 127.85 (aryl C's), 72.54, 72.04 (CHO), 66.75 (CH<sub>2</sub>Cbz), 61.58, 60.42 (OCH<sub>3</sub>), 61.39 (CHN), 32.52 (CH<sub>3</sub>N), 28.66

(isopropyl CH), 23.94 (CHcpyl), 23.14 (CHcpyl), 20.69, 20.55 (CH<sub>3</sub>), 18.49, 18.38 (CH<sub>3</sub>), 14.20 (CHcpyl), 11.66, 11.34 (CH<sub>2</sub>cpyl); mass spectrum (ESI-MS) m/z (C<sub>19</sub>H<sub>28</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M) 364.1998.

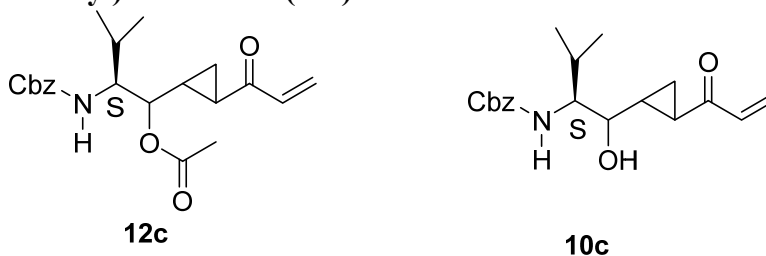
**Formation of (2S)-2-(((benzyloxy)carbonyl)amino)-1-(2-(methoxy(methyl)carbamoyl) cyclopropyl)-3-methylbutyl acetate (15c):**



To a solution of (**9c**) (332.1 mg, 0.91 mmol) in 5.0 mL CH<sub>2</sub>Cl<sub>2</sub>, acetic anhydride (0.6 mL, 6.38 mmol) and pyridine (0.52 mL, 6.38 mmol) were added. The resulting solution was stirred for 52 hours at room temperature, and then poured into 1M HCl and extracted twice with ethyl acetate. The organic layer was washed with aqueous NaHCO<sub>3</sub> and dried over magnesium sulfate, filtered and the solvent was evaporated. The crude product was purified on CombiFlash ethyl acetate-hexane (55:45) to afford 326.9 mg (88.35%) (**15c**) as a mixture of isomers. H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.30-7.24 (m, 5H, aryl isomers), 5.07, 5.05 (d, J= 6.85, 5.15 Hz, 1H, NH isomers), 5.03 (m, 2H, CH<sub>2</sub> Cbz isomers), 4.69, 4.61 (dd, J= 5.2, 5.15 Hz, 1H, CHO isomers), 3.76 (m, 1H, CHN isomers), 3.7, 3.64, 3.62 (s, 3H, OCH<sub>3</sub> isomers), 3.11, 3.1 (s, 3H, NCH<sub>3</sub> isomers), 1.98, 1.96 (s, 3H, acetate CH<sub>3</sub> isomers), 1.83 (m, 1H, isopropyl CH isomers), 1.67, 1.6, 1.54 (m, 1H, CHcpyl isomers), 1.12 (m, 2H, CH<sub>2</sub>cpyl isomers), 0.95 (d, J= 6.85 Hz, 3H, isopropyl CH<sub>3</sub> isomers), 0.85 (d, J= 6.85 Hz, 3H, CH<sub>3</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 173.28, 172.96 (C=O, ester), 170.55, 170.46 (C=O, amide), 156.62, 156.59 (C=O, Cbz), 136.72 (4°aryl C's), 128.53,

128.12, 128.08, 127.98 (aryl C's), 75.05, 74.58 (CHO), 66.75 (CH<sub>2</sub>Cbz), 61.68, 61.61, 60.42 (OCH<sub>3</sub>), 58.67 (CHN), 32.56 (CH<sub>3</sub>N), 28.85, 28.77 (isopropyl CH), 21.66, 21.08, 20.89 (CHcpyl), 20.39, 20.34(acetate CH<sub>3</sub>) 18.22 (CH<sub>3</sub>), 15.45, 14.25 (CHcpyl), 12.09, 11.89 (CH<sub>2</sub>cpyl); mass spectrum (ESI-MS) m/z (C<sub>21</sub>H<sub>30</sub>N<sub>2</sub>O<sub>6</sub>) calculated for (M) 406.2104.

**Formation of (2S)-1-(2-acryloylcyclopropyl)-2-(((benzyloxy)carbonyl)amino)-methylbutyl acetate (12c) and benzyl ((2S)-1-(2-acryloylcyclopropyl)-1-hydroxy-3-methylbutan-2-yl) carbamate (10c):**



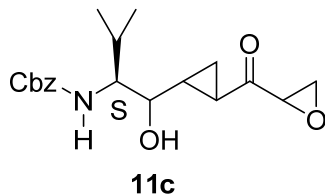
To a solution of (**15c**) (151.3 mg, 0.36 mmol) in THF (2.0 mL), at 0°C under argon, vinylmagnesium bromide (0.93 ml of 1M solution) was added, and stirred for 1 hour on melting ice bath, and again same amount of vinylmagnesium bromide was added in every 1 hour for two more time at 0°C under argon. After 4 hours, the resulting solution was poured into 1M HCl, and extracted twice with ethyl acetate. The organic layer was washed with NaHCO<sub>3</sub>, dried over anhydrous magnesium sulfate, filtered and the solvent evaporated. The crude product was purified by flash column chromatography on silica gel eluting with 1:10, 1:5, and 1:2 ethyl acetate-hexane to afford and 41.1 mg (33.31%) (**10c**) and 39.8 mg (28.63%) (**12c**) as a mixture of isomers.

**Compound 10c (alcohol group):**<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.35-7.31 (m, 5H, aryl isomers), 6.47-6.39 (m, 1H, vinyl CH isomers), 6.28 (d, J= 17.75 Hz, 2H, vinyl CH<sub>2</sub> trans isomers), 5.81, 5.77 (dd, J= 10.9, 10.85 Hz, 2H, vinyl CH<sub>2</sub> cis isomers), 5.11-5.05 (m, 2H,

CH<sub>2</sub>Cbz), 5.0 (m, 1H, NH isomers), 3.69 (m, 1H, CHO isomers), 3.63 (m, 1H, CHN isomers), 3.48 (m, 1H, CHO isomers), 3.01, 2.96 (1H, OH isomers), 2.32, 2.26 (m, 1H, CHcypyl isomers), 1.94 (m, 1H, isopropyl CH isomers), 1.61 (m, 1H, CHcypyl isomers), 1.24, 1.12 (m, 2H, CH<sub>2</sub>cypyl isomers), 0.97 (t, J= 12.6 Hz, 3H, isobutyl CH<sub>3</sub> isomers), 0.92 (t, J=12.6 Hz, 3H, isobutyl CH<sub>3</sub> isomers); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 201.06, 200.05 (C=O, ketone), 157.73, 157.65 (C=O, Cbz), 136.77, 136.73 (alkene, CH), 136.32 (4° aryl C), 128.63, 128.3, 128.14 (aryl C's), 128.22 (alkene CH<sub>2</sub>), 73.32, 71.86 (CHO), 67.17 (CH<sub>2</sub>Cbz), 61.75, 61.6 (CHN), 28.89, 28.65 (isopropyl CH), 27.33, 26.32 (CHcypyl), 22.46, 22.12 (CHcypyl), 20.59, 20.51, 18.97, 18.49 (CH<sub>3</sub>), 15.00, 14.22 (CH<sub>2</sub>cypyl); mass spectrum (ESI-MS) m/z (C<sub>19</sub>H<sub>26</sub>N<sub>2</sub>O<sub>4</sub>) calculated for (M+H) 332.1857, found 332.1856.

**Compound (12c):** <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>): δ 7.38-7.2 (m, 5H, aryl isomers), 6.43 (m, 1H, vinyl CH isomers), 6.28, 6.25 (dd, J= 6.85, 6.85 Hz, 2H, vinyl CH<sub>2</sub> trans isomers), 5.84 (m, 2H, vinyl CH<sub>2</sub> cis isomers), 5.07 (m, 2H, CH<sub>2</sub> Cbz isomers), 4.79 (m, 1H, NH isomers), 4.70, 4.55 (m, 1H, CHO isomers), 3.82, 3.66 (m, 1H, CHN isomers), 2.49, 2.39, 2.20 (m, 1H, CHcypyl isomers), 2.06, 2.03, 2.02, 1.99 (s, 3H, acetate CH<sub>3</sub> isomers), 1.86 (m, 1H, isopropyl CH isomers), 1.74, 1.69 (m, 1H, CHcypyl isomers), 1.39, 1.35, 1.28, 1.12 (m, 2H, CH<sub>2</sub>cypyl isomers), 0.99, 0.93, 0.89 (ddd, J=2.85, 1.7, 4.6 Hz, 6H isopropyl CH<sub>3</sub> isomers); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>): δ 199.05, 198.65 (C=O, ketone), 170.68, 170.61 (C=O, ester), 156.62, 156.53 (C=O, Cbz), 137.03 (4° aryl C), 136.57, 136.48 (alkene, CH), 128.64, 128.42, 128.27, 128.23, 128.15 (aryl C's), 128.62 (alkene CH<sub>2</sub>), 75.78, 74.59 (CHO), 67.03, 66.98 (CH<sub>2</sub>Cbz), 58.59 (CHN), 28.61 (isopropyl CH), 25.80 (CHcypyl), 23.53 (CHcypyl), 21.16, 20.5 (acetate CH<sub>3</sub>), 17.79 (CH<sub>3</sub>), 15.25, 14.94 (CH<sub>2</sub>cypyl); mass spectrum (ESI-MS) m/z (C<sub>21</sub>H<sub>27</sub>NO<sub>5</sub>) calculated for (M) 373.1889.

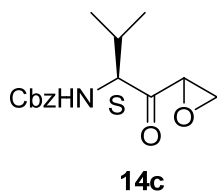
**Formation of benzyl ((2S)-1-hydroxy-3-methyl-1-(2-(oxirane-2-carbonyl)cyclopropyl) butan-2-yl)carbamate (**11c**):**



A solution of potassium tert-butoxide in tetrahydrofuran (1.0 M, 43  $\mu$ L, 0.348 mmol) was added to a stirred solution of (**10c**) (38.5 mg, 0.116 mmol) and tert-butyl hydroperoxide in hexane (65  $\mu$ L, 0.58 mmol) in tetrahydrofuran (1.0 mL) at 0°C. To the reacting mixture, the same amount of tert-butyl hydroperoxide and the same amount of potassium tert-butoxide were added in every 1 hr for another 5 hrs, and the reacting mixture was maintained at 0°C for another 17 hrs. The resulting mixture was stirred at room temperature by adding solid sodium sulfite (146 mg, 1.16 mmol) for 30 minutes. The crude product was dissolved in ethyl acetate (15 mL), and filtered, the solvent evaporated. The crude product was purified by flash column chromatography on silica gel eluting with 1:10, 1:5 and 1:2 ethyl acetate-hexane to afford 18.6 mg (46.1%) (**11c**) as a mixture of isomers.  $^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ ):  $\delta$  7.36-7.31 (m, 5H, aryl isomers), 5.10, 5.06 (dd,  $J=13.75, 3.4$  Hz, 2H,  $\text{CH}_2$  Cbz isomers), 4.84 (d,  $J= 8.05$  Hz, 1H, NH), 3.7 (m, 1H, CHO isomers), 3.61 (m, 1H, CHN isomers), 3.46, 3.43 (q,  $J=6.85$ Hz, 1H,  $\text{CHepoxide}$  isomers), 3.35 (m, 1H, CHO isomers), 3.12, 2.96, 2.90 (dd,  $J= 10.0, 10.5, 8.6$  Hz,  $\text{CH}_2\text{epoxide}$  isomers), 2.78 (m, 1H,  $\text{CHcpyl}$  isomers), 2.35, 2.31 (1H, OH isomers), 2.14, 2.09, 1.93, 1.87 (m, 1H, isopropyl CH isomers), 1.30, 1.21 (m, 1H,  $\text{CH}_2\text{cpyl}$  isomers), 0.98-0.9 (m, 6H, isopropyl  $\text{CH}_3$  isomers);  $^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ ):  $\delta$  205.66, 205.5 (C=O, ketone), 156.44, 156.27

(C=O, Cbz), 134.94 (4° aryl C), 127.40, 127.12, 126.96, 126.80 (aryl C's), 72.48, 70.91, 70.5, 70.35, 70.07 (CHO), 66.03 (CH<sub>2</sub>Cbz), 60.44, 60.27 (CHN), 52.67, 52.36 (CHepoxide) 44.88, 44.72 (CH<sub>2</sub>epoxide), 28.5, 27.69, 27.54, 27.24, 27.14 (CHisopropyl), 25.52 (CHcpyl), 19.26, 19.11 (CHcpyl), 17.75, 17.52 (CH<sub>3</sub>), 13.34 (CH<sub>2</sub>cpyl); mass spectrum (ESI-MS) m/z (C<sub>19</sub>H<sub>26</sub>N<sub>2</sub>O<sub>5</sub>) calculated for (M+H) 348.1802, found 348.1805.

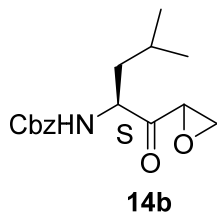
**benzyl ((2S)-3-methyl-1-(oxiran-2-yl)-1-oxobutan-2-yl)carbamate (14c):**



A solution of potassium tert-butoxide in THF (1.0 M, 1.3  $\mu$ L, 0.096 mmol) was added to a stirred solution of (**5c**) (50 mg, 0.19 mmol) and tert-butyl hydroperoxide in hexane (43  $\mu$ L, 0.38 mmol) in THF (1.5 mL) at 0°C and stirred. In every one-hour same amount of tert-butyl hydroperoxide and potassium tert-butoxide were added over 4 h, then the resulting solution was kept at 0°C for another 13 h, then solid sodium sulfite (121 mg, 0.96 mmol) was added and stirred at room temperature for 1 h. The crude compound was dissolved in ethyl acetate (15 mL) and filtered, the solvent evaporated, and then purified by flash column chromatography on silica gel eluting with 1:20, 1:10 and 1:6 ethyl acetate-hexane to afford 25.1 mg (47.36%) (**14c**) as a mixture of syn and anti-isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.36 (m, 5H, aryl), 5.38 (d, J=8.6 Hz, 1H, NH), 5.1 (m, 2H, CH<sub>2</sub>Cbz), 4.59 (m, 1H, CHN), 3.67 (s, 1H, CH epoxide), 2.99, 2.92 (d, 2H, CH<sub>2</sub> epoxide), 2.28 (m, 1H, CH(CH<sub>3</sub>)<sub>2</sub>), 1.06, 0.88 (s, 3H, CH<sub>3</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  204.84 (C=O, ketone), 156.39 (C=O, Cbz), 136.17 (4° aryl C's), 128.67, 128.37, 128.22 (3 aryl C's),

67.29 (CH<sub>2</sub>Cbz), 63.37 (CHN), 51.12 (CH epoxide), 48.02 (CH<sub>2</sub> epoxide), 30.19 (CH(CH<sub>3</sub>)<sub>2</sub>), 19.85, 17.01 (CH<sub>3</sub>); Mass spectrum (ESI-MS) m/z (C<sub>15</sub>H<sub>20</sub>NO<sub>4</sub>) calculated for (M+H) 278.1370, found 278.1387.

**benzyl ((2S)-4-methyl-1-(oxiran-2-yl)-1-oxopentan-2-yl)carbamate (14b):**



A solution of potassium tert-butoxide in THF (1.0 M, 1.3  $\mu$ L, 0.011 mmol) was added to a stirred solution of (**5b**) (60 mg, 0.21 mmol) and tert-butyl hydroperoxide in hexane (48.64  $\mu$ L, 0.44 mmol) in THF (2.0 mL) at 0°C and stirred. In every 1 h, the same amount of tert-butyl hydroperoxide and the same amount of potassium tert-butoxide were added over 3 h, then the resulting solution was kept at 0°C for another 12 h, then solid sodium sulfite (137 mg, 1.1 mmol) was added and stirred at room temperature for 1 h. The crude compound was dissolved in ethyl acetate (20 mL) and filtered, the solvent evaporated, and then purified by flash column chromatography on silica gel eluting with 1:20, 1:10 and 1:6 ethyl acetate-hexane to afford 34.6 mg (56.72%) (**14b**) as a mixture of syn and anti-isomers. <sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>):  $\delta$  7.36-7.33 (m, 5H, aryl), 5.16 (d, j=8.6, 1H, NH), 5.07 (m, 2H, CH<sub>2</sub>Cbz), 4.38 (m, 1H, CHN), 3.51 (s, 1H, CH epoxide), 3.14, 3.06 (d, 2H, CH<sub>2</sub> epoxide), 1.73 (m, 1H, CH(CH<sub>3</sub>)<sub>2</sub>), 1.57, 1.28 (m, 2H, CH<sub>2</sub>), 0.97, 0.94 (s, 3H, CH<sub>3</sub>); <sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>):  $\delta$  207.23 (C=O, ketone), 156.19 (C=O, CH<sub>2</sub>Cbz), 136.17 (4° aryl C's), 128.63, 128.3, 128.01 (3 aryl C's), 67.11 (CH<sub>2</sub>Cbz), 52.73 (CHN), 52.61 (CH epoxide),



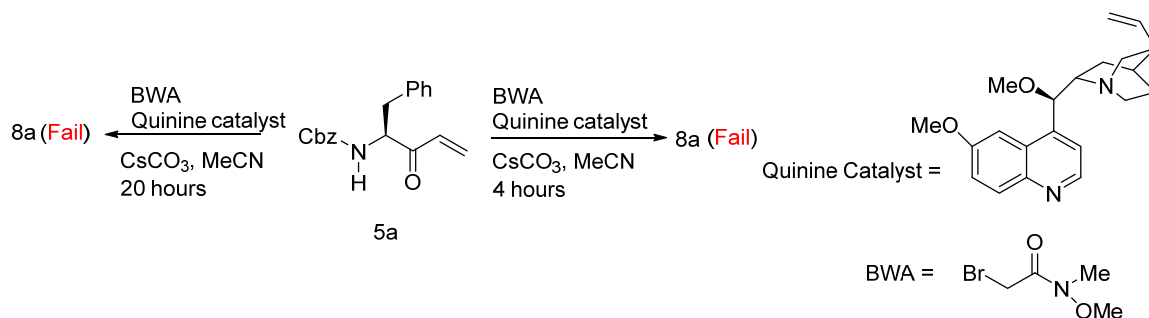
46.39 (CH<sub>2</sub> epoxide), 40.26 ( CH<sub>2</sub>), 25.04 (CH(CH<sub>3</sub>)<sub>2</sub>), 23.42, 21.41 (CH<sub>3</sub>); mass spectrum (ESI-MS) m/z (C<sub>16</sub>H<sub>22</sub>NO<sub>4</sub>) calculated for (M+H) 292.1529, found 292.1543.

## CHAPTER III: RESULTS AND DISCUSSION

### Synthesis of cyclopropyl Weinreb amide

In order to synthesize the proposed cyclopropyl epoxy ketones, the starting point would be our existing synthesis of cyclopropyl esters. The amino acid-derived enones need to be converted to cyclopropyl Weinreb amides. These could then be converted to enones which could be epoxidized to give the epoxyketones. Hoping to get a stereoselective cyclopropyl Weinreb amide as reported by Gaunt, the bromo Weinreb amide of bromoacetic acid (BWA) was treated with **5a** in presence of the quinine catalyst and CsCO<sub>3</sub> as seen in **Scheme 6**.<sup>35</sup> But, it was found that the starting compound **5a** was decomposed at the end of the reaction despite trying different reaction times.

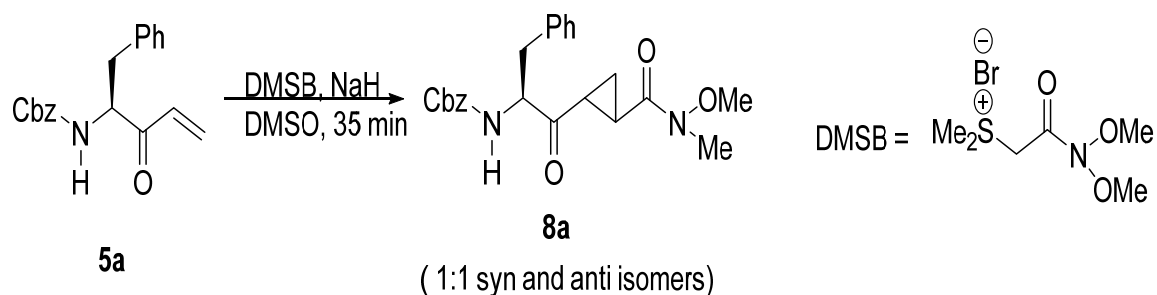
**Scheme 6:** Attempted synthesis of cyclopropyl Weinreb amide



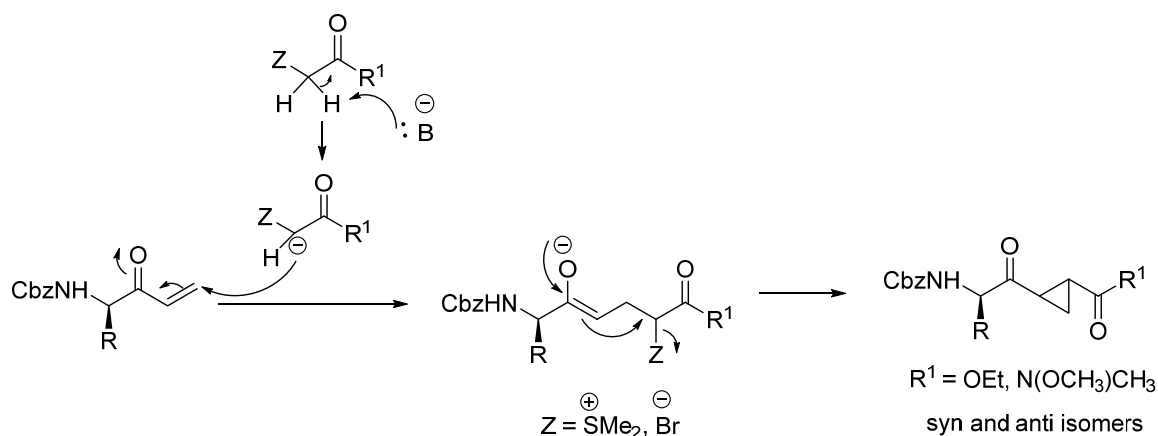
Although Gaunt had reported successful stereoselective cyclopropanations of other enones using BWA, this did not work with our substrate. After the failure with BWA, a salt, (2-(methoxy(amino)-2-oxoethyl)dimethylsulfonium bromide (DMSB), was synthesized by displacing the bromide of BWA with dimethyl sulfide. This salt was a precursor to the sulfonium ylide, an analog of EDSA, used in the earlier synthesis. The

salt was treated with NaH in DMSO followed by **5a** to afford the cyclopropyl Weinreb amide **8a**, as a 1:1 mixture of diastereomers (Scheme 7). During this reaction, it was observed that the reaction time was very crucial, that is, the reactants were decomposed completely when they were allowed to react for 1 hour.

**Scheme 7:** Successful synthesis of cyclopropyl Weinreb amide



The general mechanism for Michael induced ring closure to form a cyclopropyl ester or Weinreb amide is shown in **Figure 18**. The strong base NaH removes an acidic hydrogen from BWA or DMSB *in situ* to form an ylide that can react with **5a** to form an intermediate enolate, which displaces the leaving group and forms a cyclopropane.

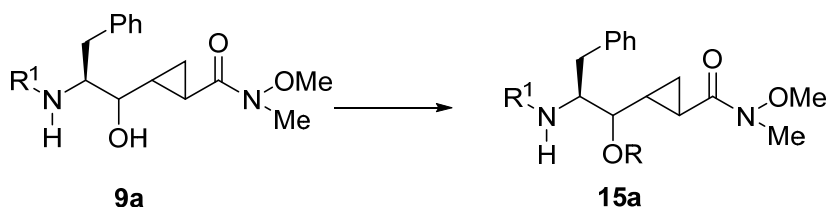


**Figure 18:** General cyclopropanation mechanism

## Protection of alcohol

The ketone of compound **8a** was reduced to the alcohol **9a** with NaBH<sub>4</sub> without difficulty, but with the alcohol group, a vinylmagnesium bromide addition to the Weinreb amide did not work in presence of the acidic hydrogen.<sup>30,31</sup> So, several protecting groups were investigated to protect this hindered alcohol.<sup>36</sup> The first attempts were to add silyl protecting groups using either t-BDMSCl or TMSCl. These were attempted several times using different conditions, but they gave only the starting materials. Also, addition of a benzyl ether protecting group failed with the hindered alcohol.

**Table 1:** Attempts at protection of the alcohol **9a**



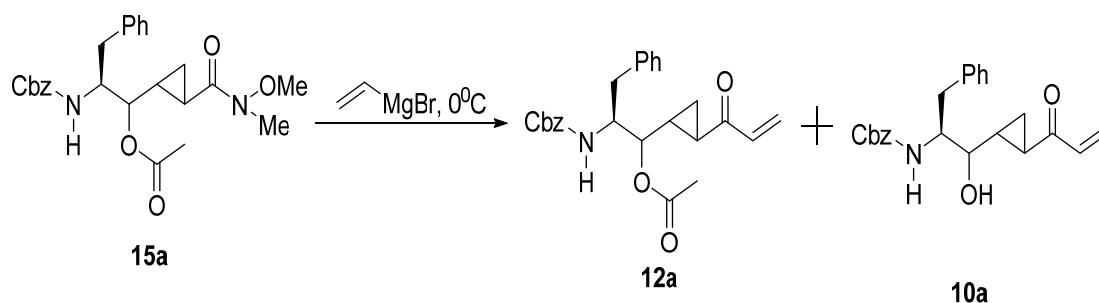
<b>9a</b> $\longrightarrow$ <b>15a</b>			
R	R'	Conditions	Results
t-Bu(CH <sub>3</sub> ) <sub>2</sub> Si	Cbz	t-Bu(CH <sub>3</sub> ) <sub>2</sub> SiCl , Et <sub>3</sub> N, CH <sub>2</sub> Cl <sub>2</sub> , RT, 24 hrs	No reaction
(CH <sub>3</sub> ) <sub>3</sub> Si	Cbz	(Me <sub>3</sub> ) <sub>3</sub> SiCl , NEt <sub>3</sub> , CH <sub>2</sub> Cl <sub>2</sub> , RT, 8 hrs	No reaction
Bn	Cbz	BnBr, NaH, THF, 24 hrs	decomposition
Bn	PhCH <sub>2</sub> CO	BnBr, NaH, THF	decomposition
Ac	Cbz	(CH <sub>3</sub> O) <sub>2</sub> O, Pyridine, CH <sub>2</sub> Cl <sub>2</sub> , 3 days	88% yield

These protecting groups did not work because of their bulky size towards the hindered alcohol. In this case, with both the Cbz analog **9a** or its corresponding amide, we saw decomposition. So, the final attempt was to protect the alcohol group as the acetate, using acetic anhydride, which successfully gave the acetate **15a**.<sup>33</sup> The protection results are summarized in Table 1.

### Synthesis of the cyclopropyl enone

After the successful protection of the hindered alcohol as an acetate group, compound **15a** was treated with vinylmagnesium bromide to convert it into the enone **12a**. During the reaction, we also observed a cleavage of the acetate, affording the alcohol **10a** as a byproduct (**Scheme 9**)<sup>30,31</sup>. This was an acceptable result since both series could be tested for bioactivity.

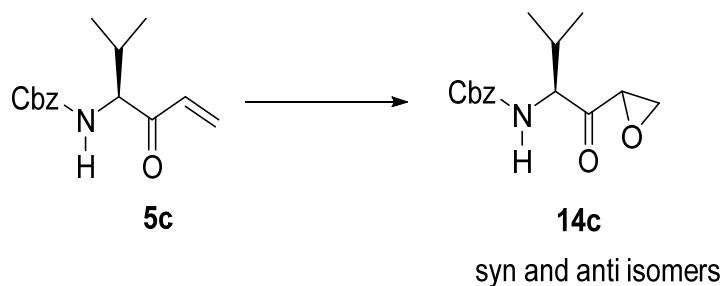
### Scheme 8: Synthesis of the cyclopropyl enone



## Synthesis of the epoxide

Formation of the epoxide proved to be another challenging task. Three different procedures were attempted multiple times. Reactions were investigated first using the simple enone **5c**.<sup>34,37,38</sup> The first attempts used mCPBA. However no reaction was observed even after 6 days. A second attempt with oxone also gave no reaction. We concluded that this electron-deficient alkene needed a different approach. Finally, the Weitz-Scheffer epoxidation in basic conditions was the successful route to synthesize **14c** from **5c** (Table 2).

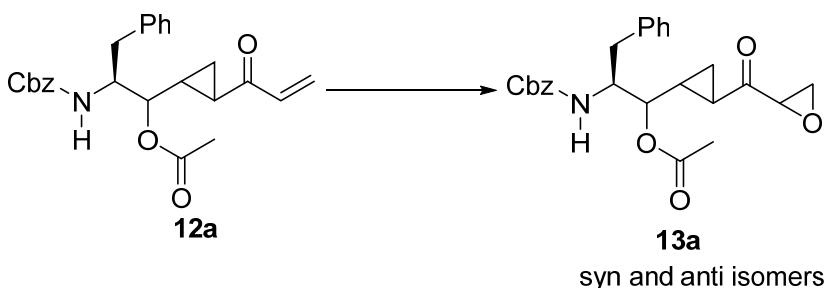
**Table 2:** Summary of the attempted synthesis of the simple epoxides



<b>5c</b> <span style="margin-left: 50px;">→</span> <b>14c</b>		
Reagents	Conditions	Results
mCPBA	CH <sub>2</sub> Cl <sub>2</sub> , RT, 3 days	No reaction
mCPBA	CH <sub>2</sub> Cl <sub>2</sub> , 0°C, 19 hrs	No reaction
mCPBA	CH <sub>2</sub> Cl <sub>2</sub> , 0°C, 6 days	No reaction
Oxone, NaHCO <sub>3</sub>	acetone, 4 hrs	No reaction
t-BuOOH, t-BuOK	THF, 0°C, 4 hrs	47 yield%

After the success of the synthesis of simple epoxides by Weitz-Scheffer epoxidation, it was implemented to synthesize the cyclopropyl epoxyketone **13a** from **12a** in presence of peroxide (t-BuOOH) and a strong base (t-BuOK) at 0°C (**Table 3**).

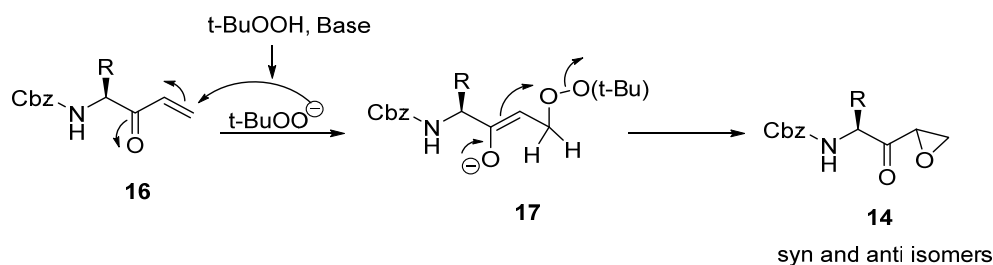
**Table 3:** Summary of the attempted synthesis of the cyclopropyl epoxyketone



<b>12a</b> <span style="margin-left: 50px;">→</span> <b>13a</b>		
Reagents	Conditions	Results
mCPBA	CH <sub>2</sub> Cl <sub>2</sub> , RT, 4 hrs	No reaction
mCPBA	CH <sub>2</sub> Cl <sub>2</sub> , RT, 18 hrs	No reaction
t-BuOOH, t-BuOK	THF, 0°C, 13 hrs	60%

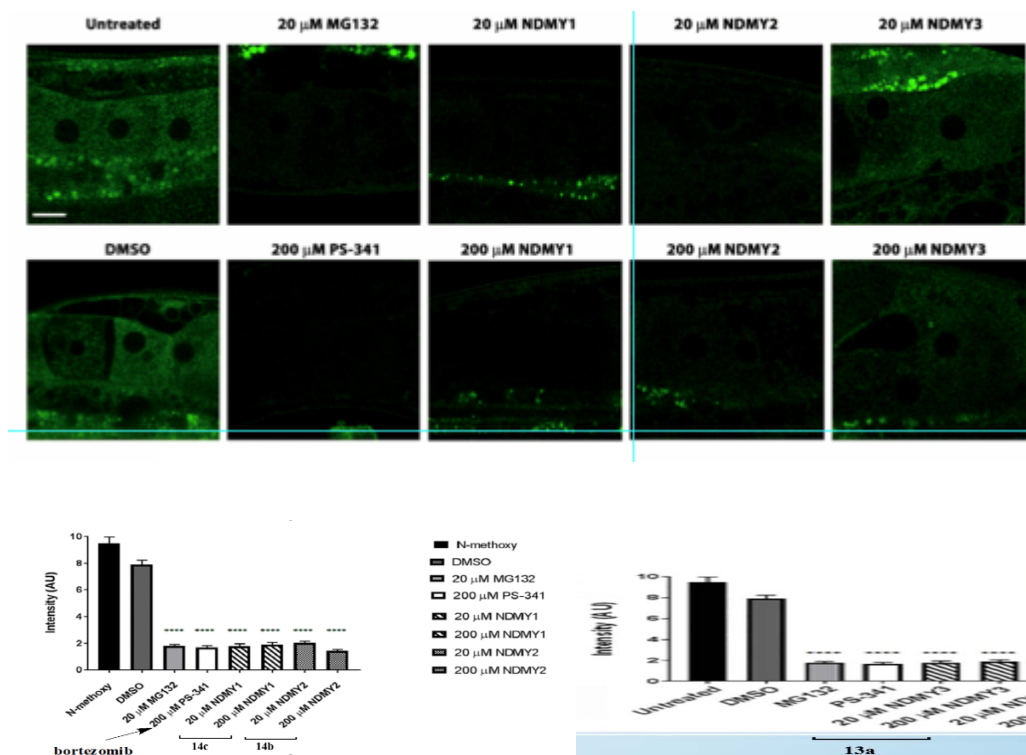
The peroxyanion formed from alkyl hydroperoxide *in situ* under basic conditions undergoes conjugate addition at the beta-position of an  $\alpha$ ,  $\beta$ -unsaturated ketone (**16**) to afford an intermediate  $\beta$ -peroxyenolate (**17**). This further undergoes intramolecular nucleophile substitution by the cleavage of O-O bond with the removal of alkoxide (RO<sup>-</sup>) to afford (**14**). The mechanism of the Weitz-Scheffer epoxidation is shown in **Scheme 9**.

**Scheme 9:** The two-step mechanism of the Weitz-Scheffer epoxidation



**Results of testing in phenotypic assay**

Several of the simple epoxides and also compound 13a were tested in a phenotypic assay in the nematode *C. elegans* as seen in Figure 19.<sup>39</sup> Simple epoxides **14b** and **14c** show similar in potency to bortezomib at 20 and 200  $\mu$ M and hybrid analog **13a** less potent than **14b** and **14c** at 20 and 200  $\mu$ M.



**Figure 19:** Phenotypic assay for simple epoxides and analog



## Conclusions

Belactosin A is a naturally occurring proteasome inhibitor with potent anti-tumor activity that has inhibitory effects on the 20S core particle of the proteasome. Several analogs have been proposed and synthesized. Carfilzomib is a tetrapeptide epoxyketone, and an analog of epoxomicin with improved properties. Carfilzomib was approved by FDA for the treatments of multiple myeloma, non-Hodgkin's lymphoma and solid tumors. The primary goal of this project was to synthesize hybrid analogs of belactosin A and carfilzomib, with the cyclopropyl backbone and epoxyketone serine trap.

Several problems were encountered during the synthesis. The first one was a protection of a hindered alcohol. While silyl ethers and benzyl ether were not obtained, it was finally protected as an acetate. Another issue was encountered during epoxidation of the cyclopropyl enone. Traditional epoxidation with electrophilic reagents was not successful. However a Weitz-Scheffer epoxidation led to two different series of epoxyketones.

In summary a successful synthesis of the hybrid analogs of belactosin A and carfilzomib has been completed. Starting from phenylalanine, leucine and valine, three series, giving both the acetate analogs **13a-c** and alcohol analogs **11a-c** have been made. In addition, the "simple" epoxyketones **14a-c** have been prepared. Initial testing of several epoxyketones in a phenotypic assay indicates that both the simple epoxides and hybrid analogs are effective proteasome inhibitors.<sup>39</sup>

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- Tripeptide Proteasome inhibitor Belactosin A and Its Highly Potent Cis-cyclopropane Stereoisomer” *Organic Letters*, **2008**, 10, 3571-3574.
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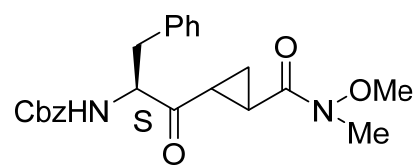
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**APPENDICES**



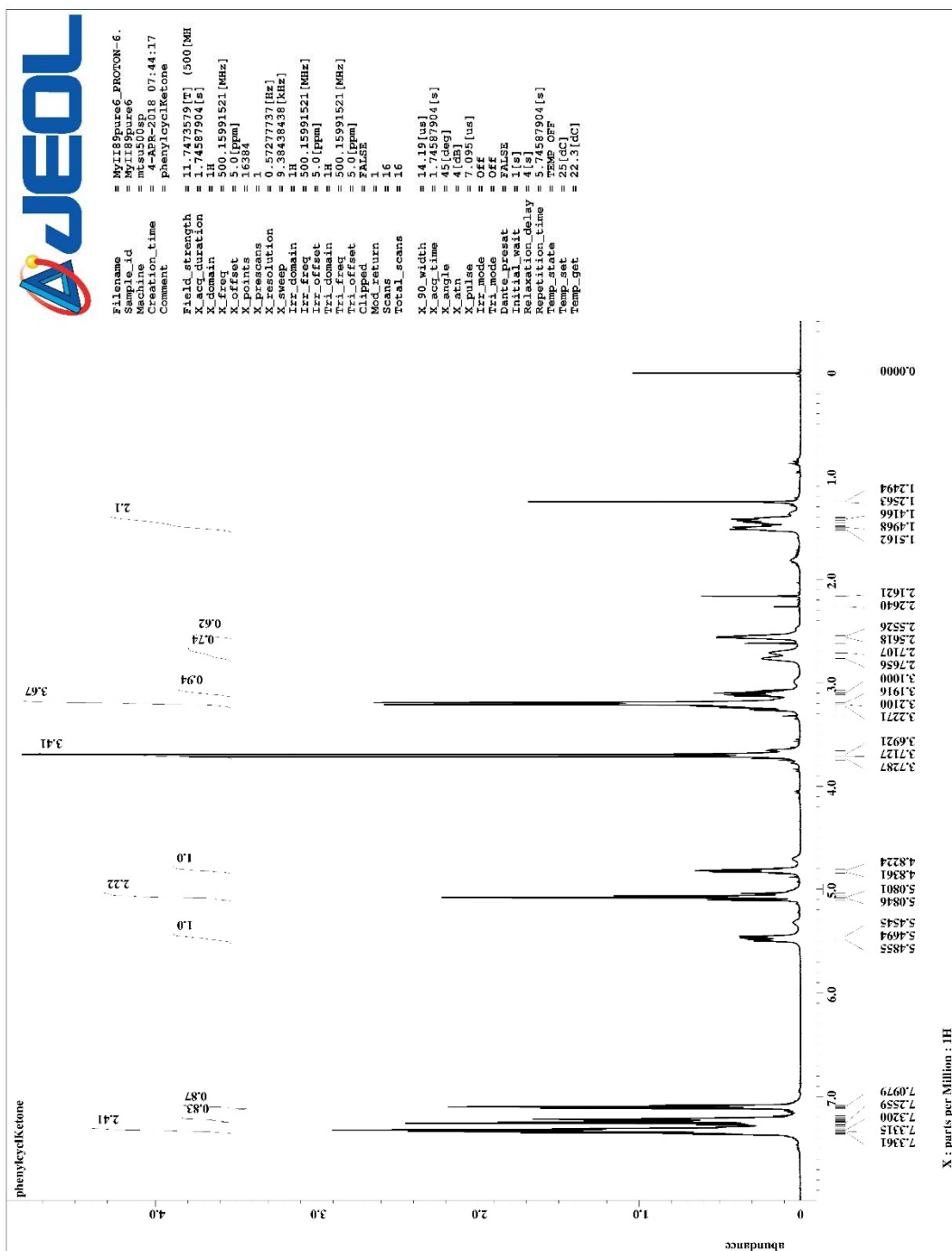
**APPENDIX A: NMR SPECTRA**



<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

HRMS



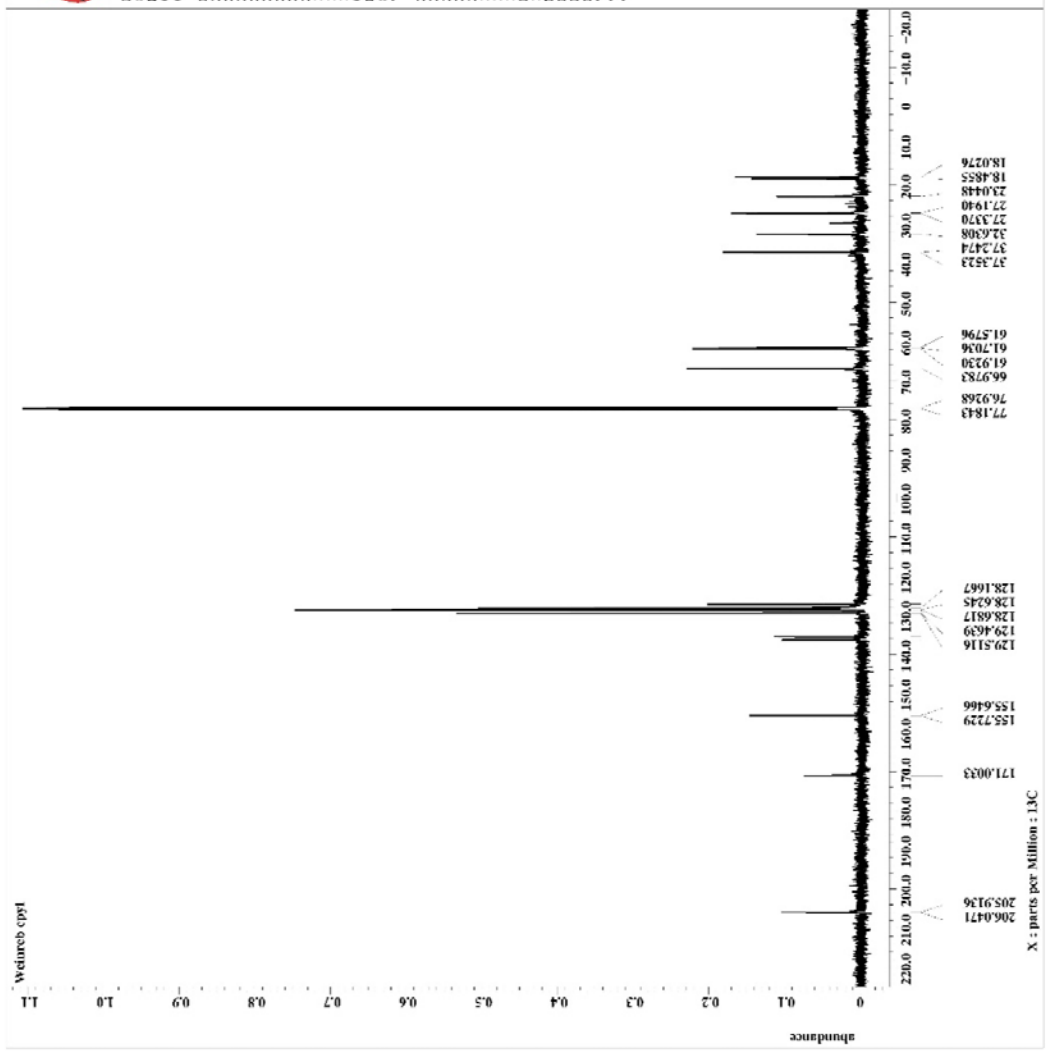


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Noe_wait = 2 [s]
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```



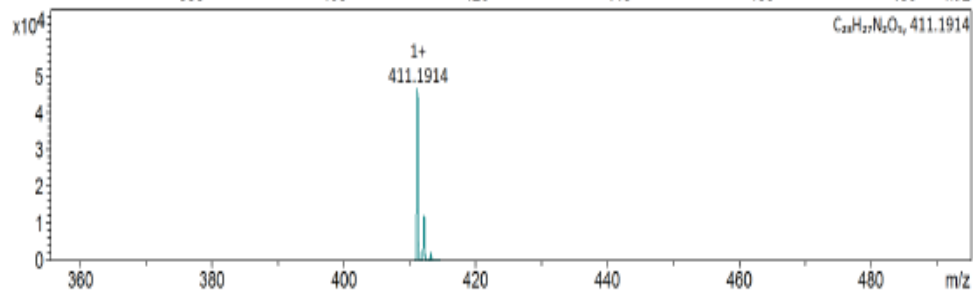
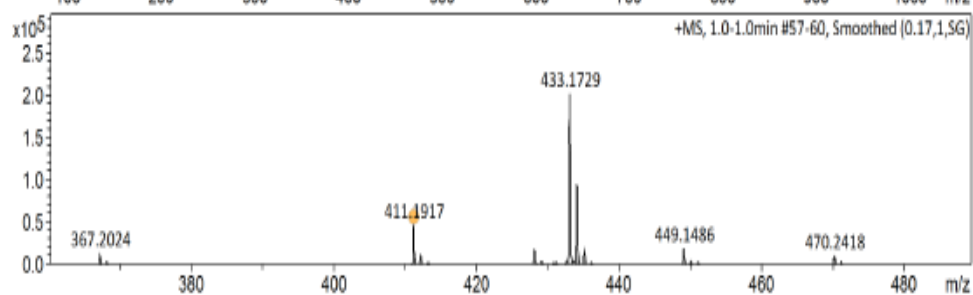
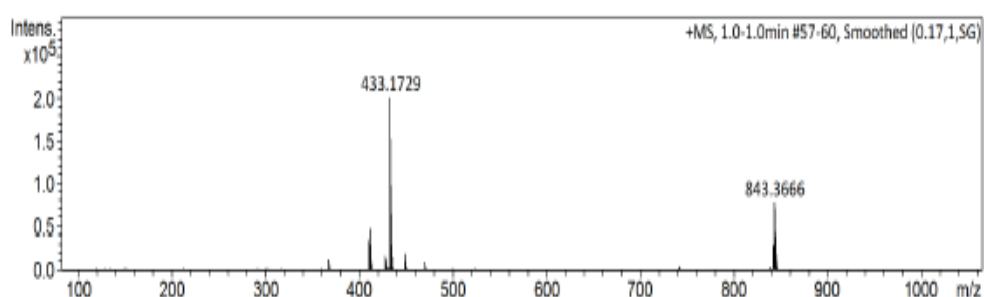
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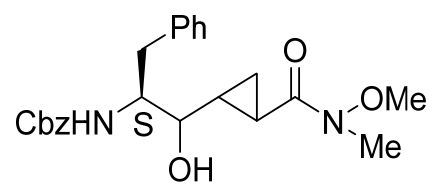
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 Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
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Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdB	N-Rule	e <sup>-</sup> Conf
411.191727	1	C <sub>23</sub> H <sub>27</sub> N <sub>2</sub> O <sub>5</sub>	411.191448	-0.7	-2.8	11.5	ok	even



<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

HRMS

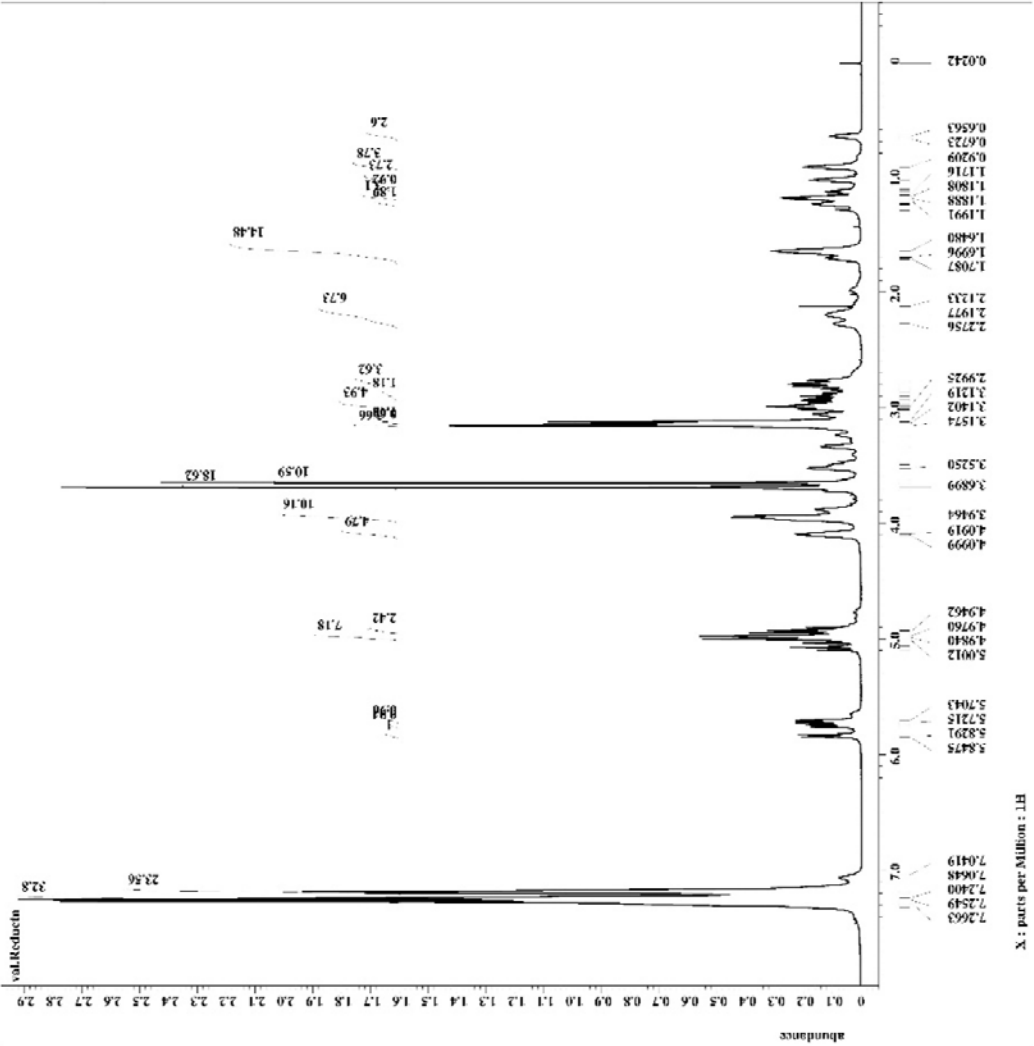


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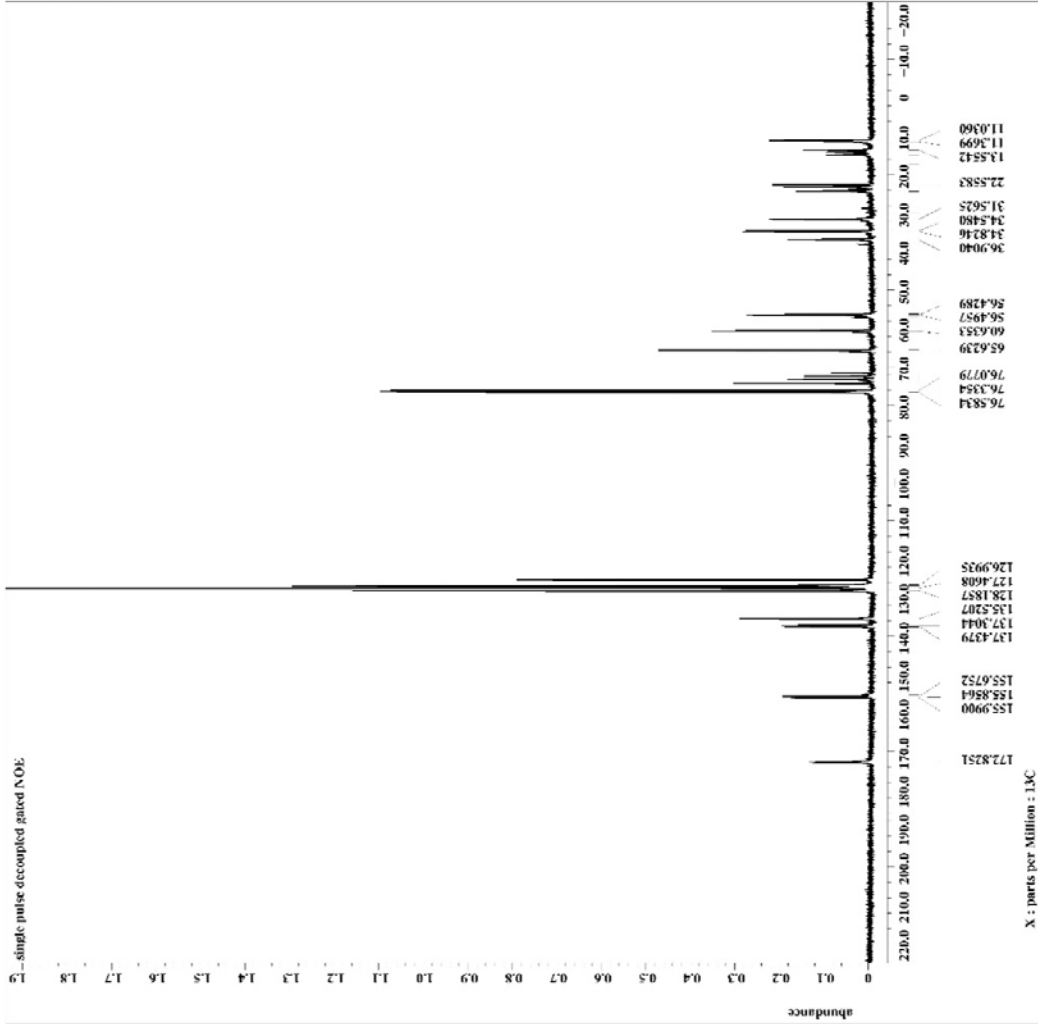
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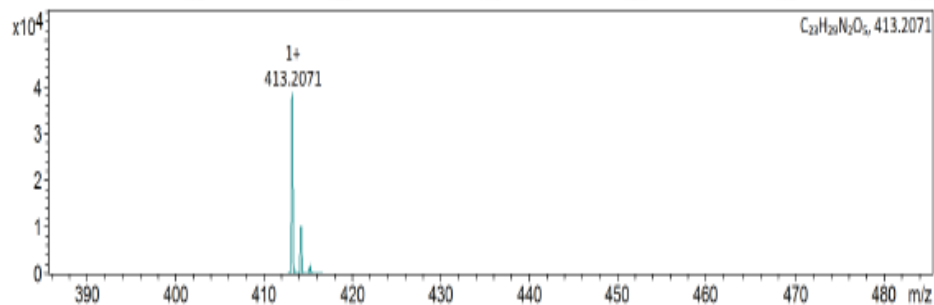
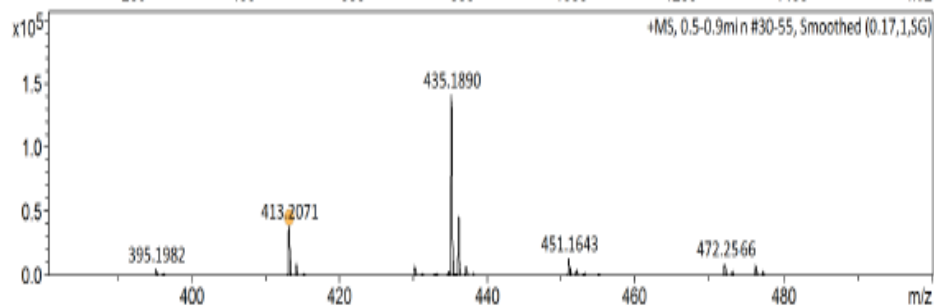
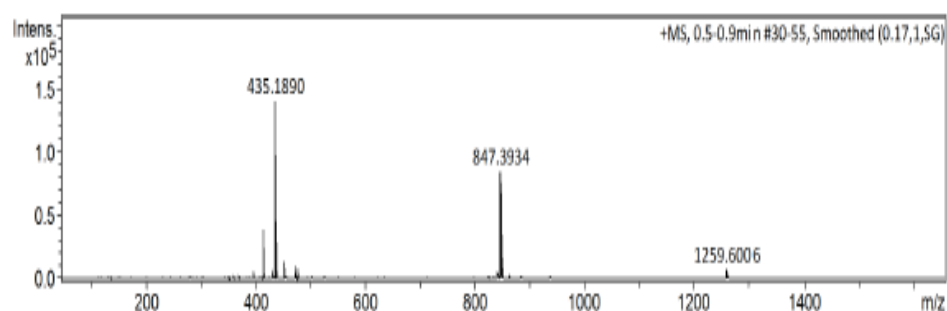
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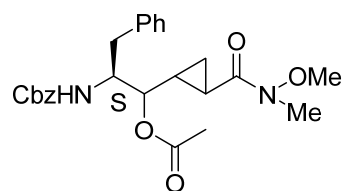
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$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

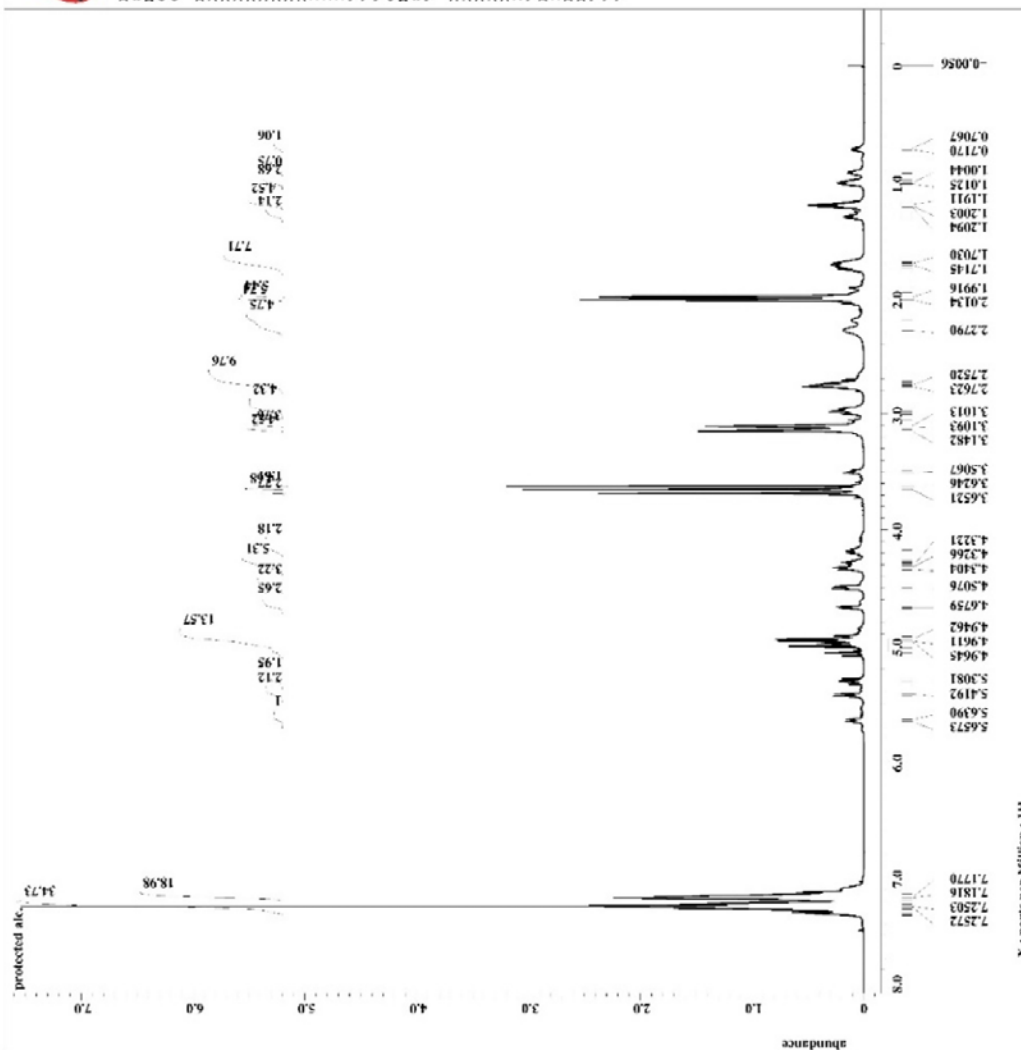
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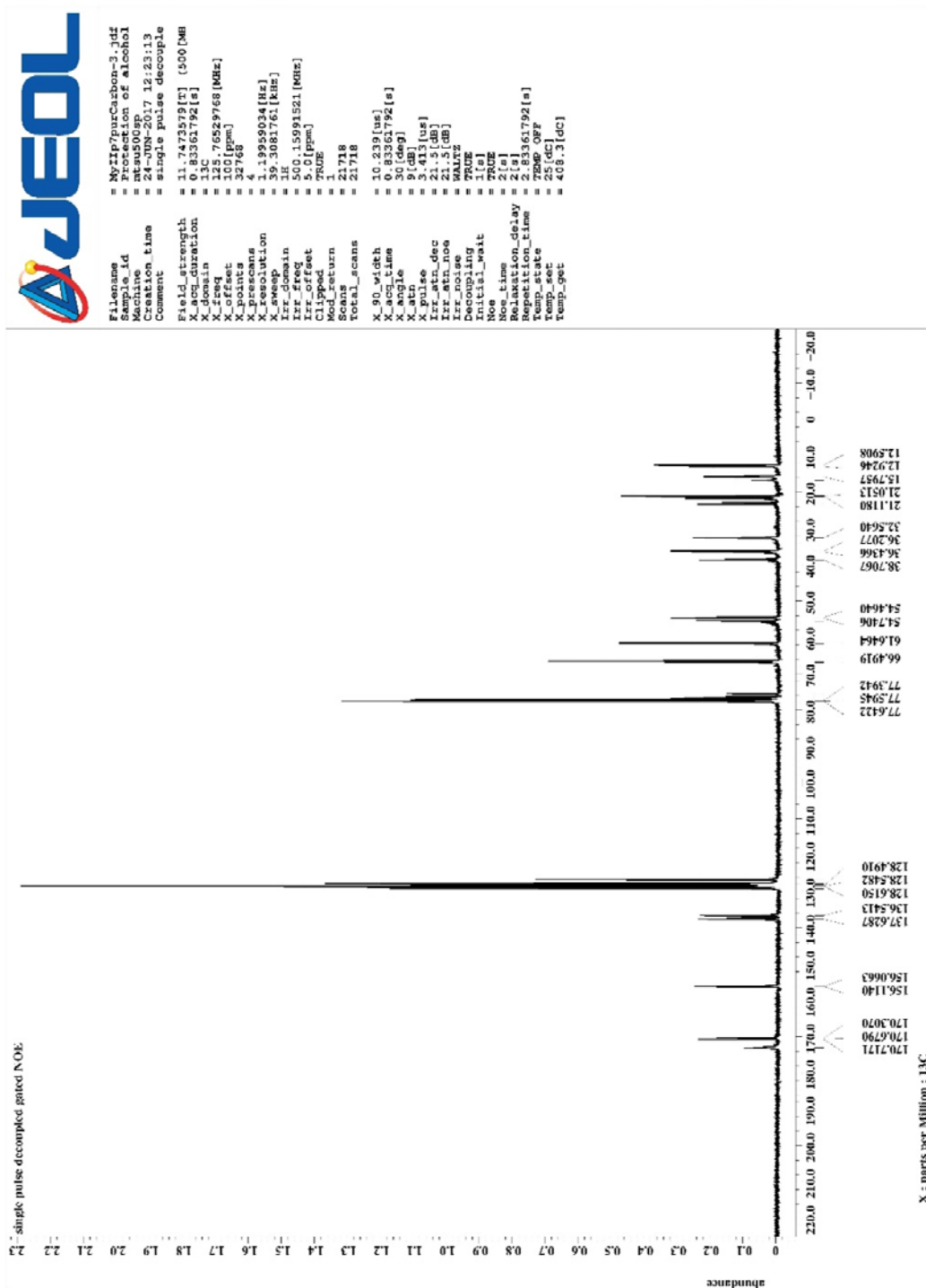
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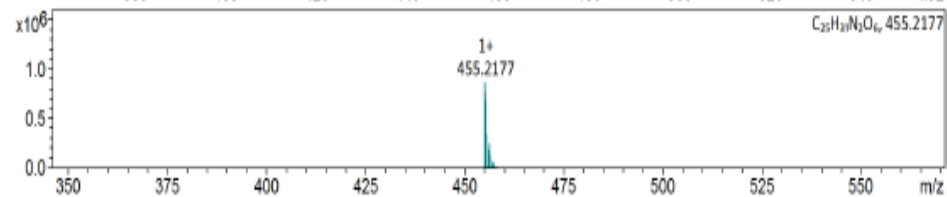
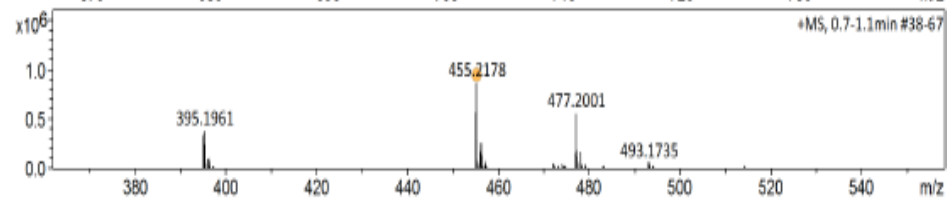
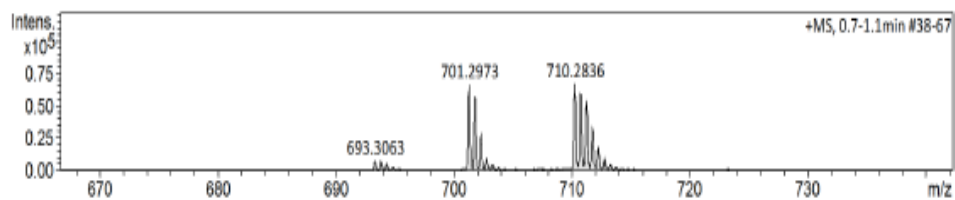
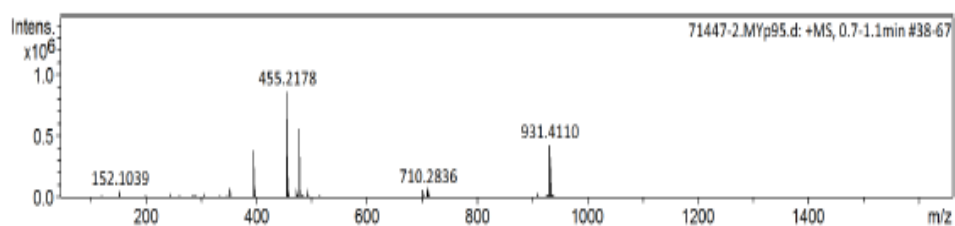
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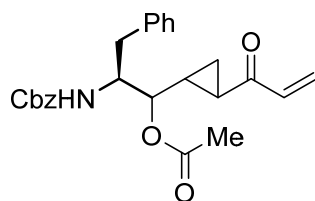
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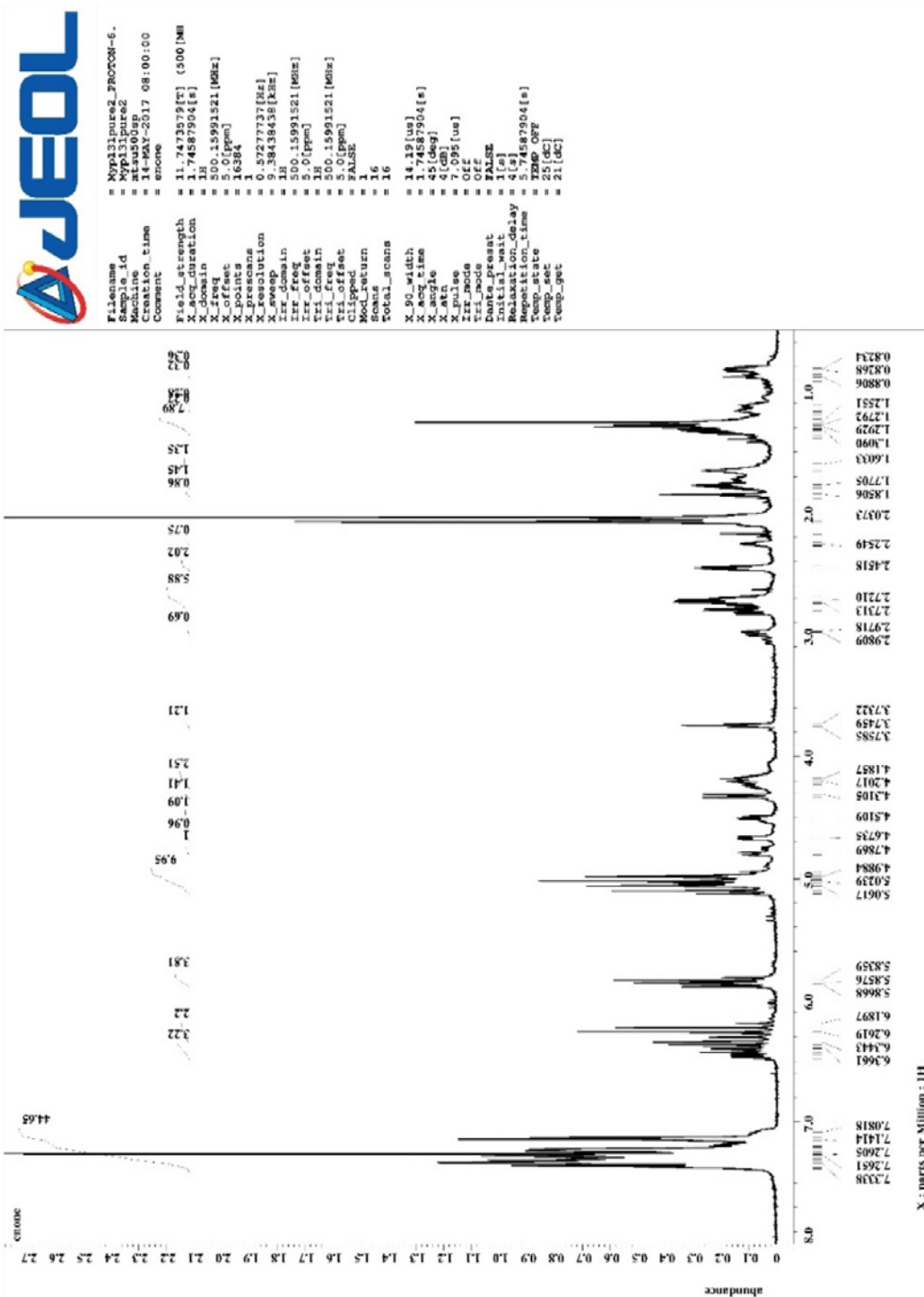
Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
455.217818	1	C <sub>25</sub> H <sub>31</sub> N <sub>2</sub> O <sub>6</sub>	455.217883	-0.3	1.2	11.5	ok	even



$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

HRMS



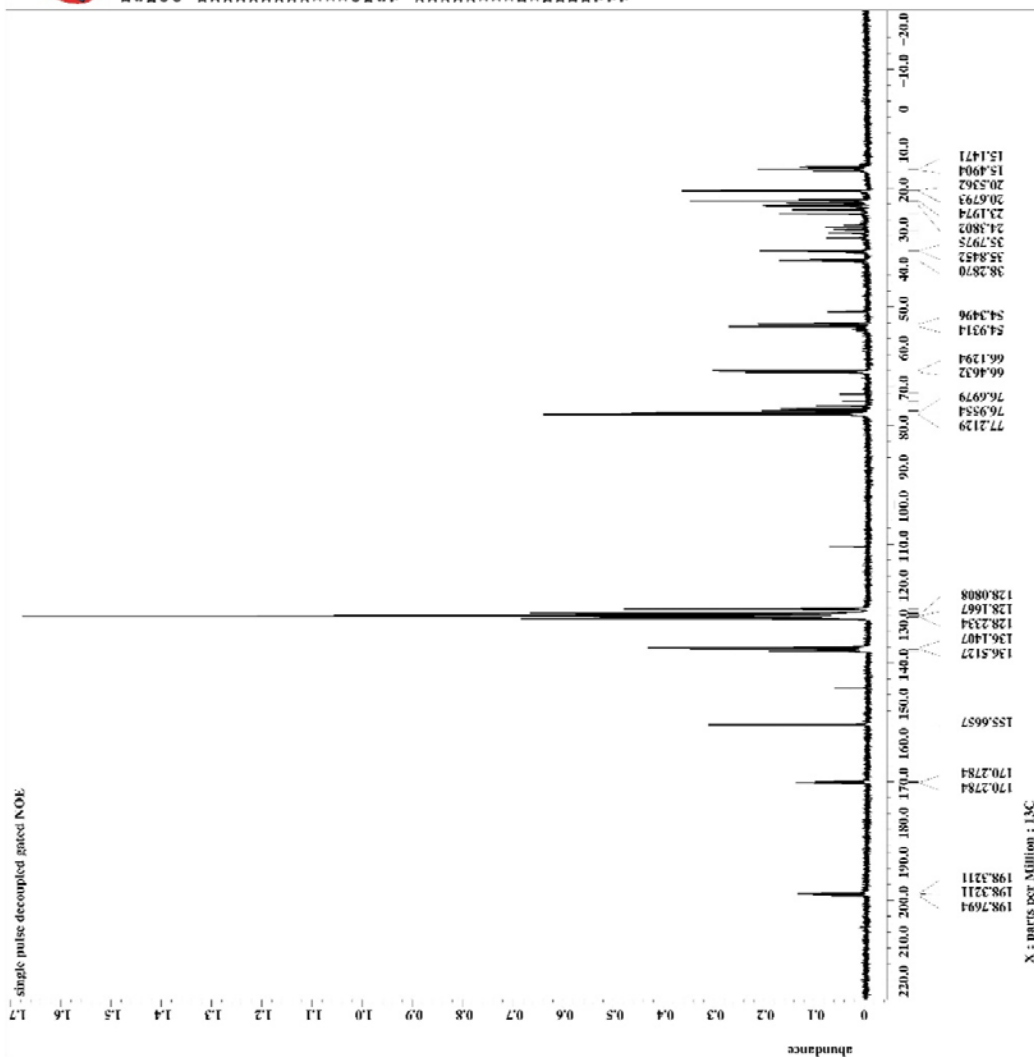


```

Filename = MyIpliprSprA-7.jde
Sample_id = Phenyl_enone
Machine = mtaub009p
Creation_time = 2010-08-20 7:07:06:17
Comment = single pulse decouple

Field_strength = 11.7472579 [T] (500 [MH]
X_acq_duration = 0.83361792 [s]
X_domain = 125.76529768 [MHz]
X_offset = 100 [ppm]
X_points = 32768
X_prescans = 4
X_resolution = 1.9959034 [Hz]
X_sweep_time = 39.3081761 [kHz]
X_domain = 119.3081761 [kHz]
X_resolution = 500.15991521 [MHz]
X_offset = 5.0 [ppm]
X_pulse_width = 7.0 [us]
Modulation = WALTZ
Modulation_delay = 1 [s]
Total_scans = 14846
X_90_width = 10.239 [us]
X_acq_time = 0.83361792 [s]
X_angle = 30 [deg]
X_atn = 2 [dB]
X_pulse_dec = 2.41 [us]
X_pulse_width = 21.5 [us]
X_atn_dec = 21.5 [dB]
X_atn_noise = WALTZ
Decoupling = TRUE
Initial_wait = 1 [s]
Noe_time = 2 [s]
Relaxation_delay = 2 [s]
Repetition_time = 2.83361792 [s]
Temp_state = TEMP OFF
Temp_set = 29.5 [degC]
Temp_get = 21.3 [degC]

```





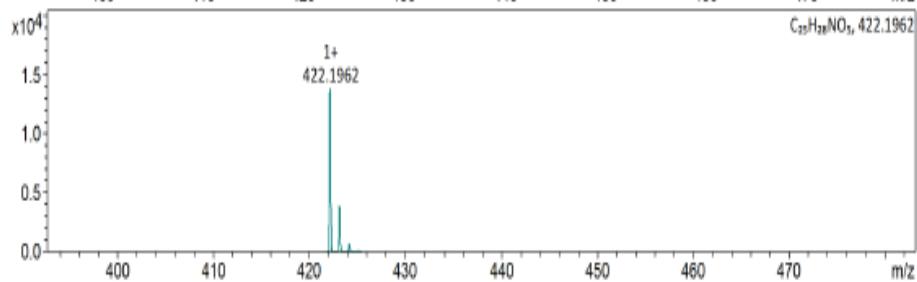
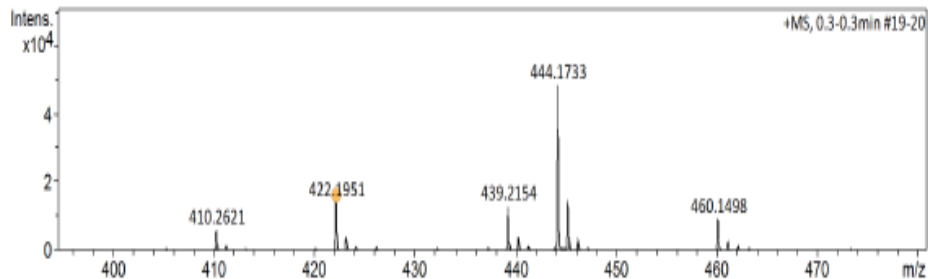
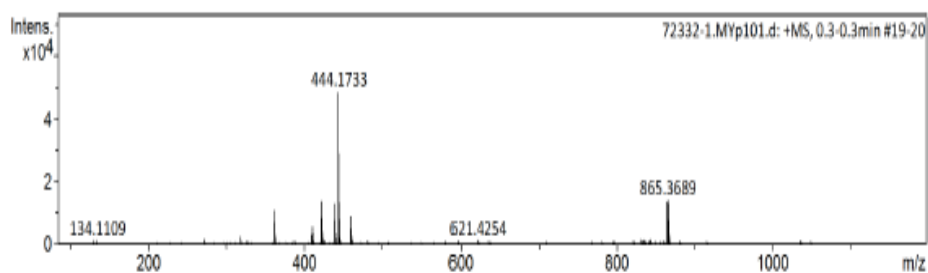
## Analysis Info

Analysis Name D:\Data\0502\72332-1.Myp101.d  
 Method 050918\_ns\_tune\_low\_pos.m  
 Sample Name 1.Myp101  
 Comment direct infusion

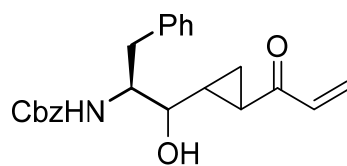
Acquisition Date 5/2/2017 4:17:55 PM  
 Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.3 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



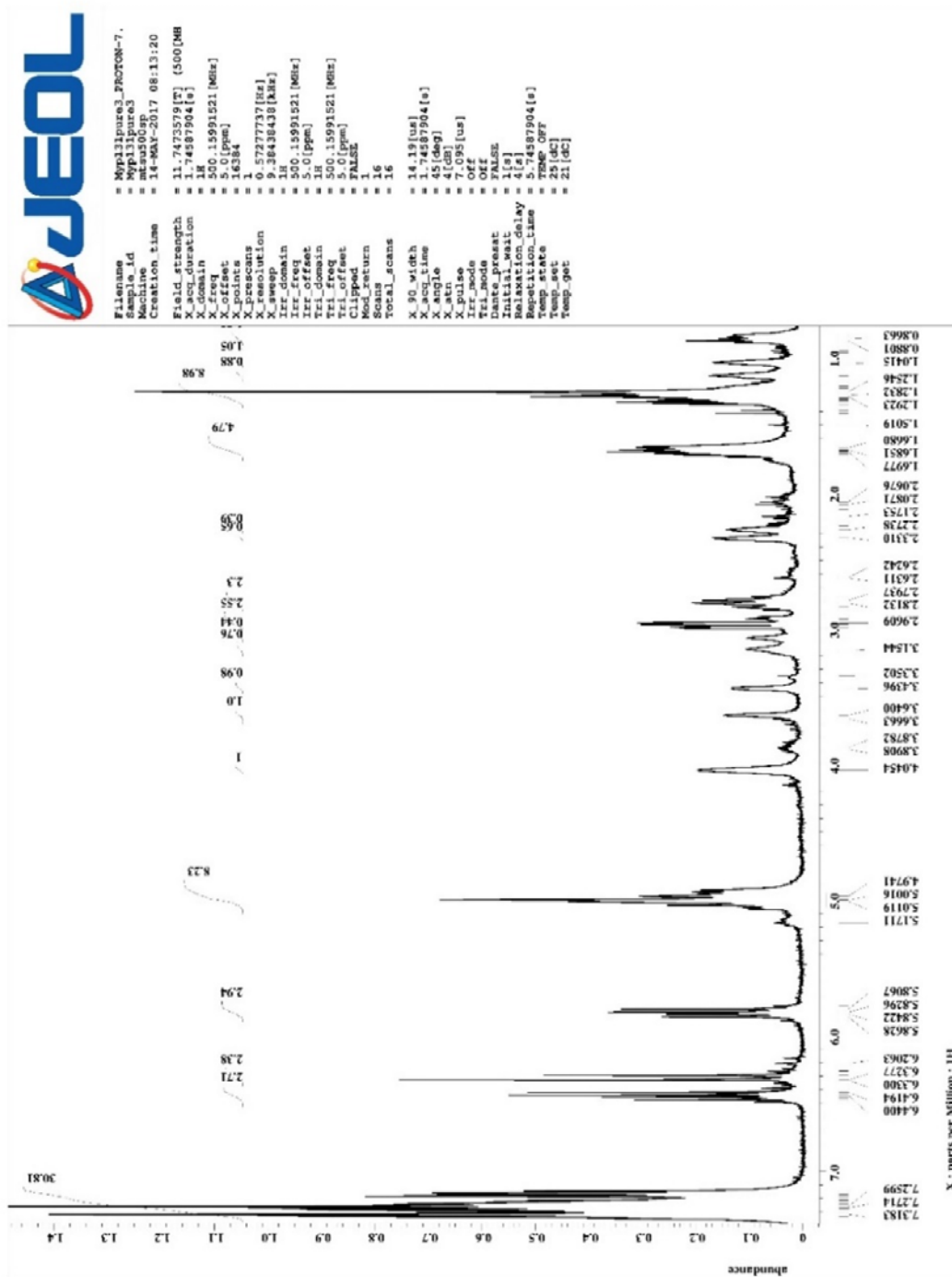
Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdl	N-Rule	e <sup>-</sup> Conf
422.195075	1	C <sub>15</sub> H <sub>28</sub> NO <sub>5</sub>	422.196199	2.7	4.5	12.5	ok	even



<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

HRMS

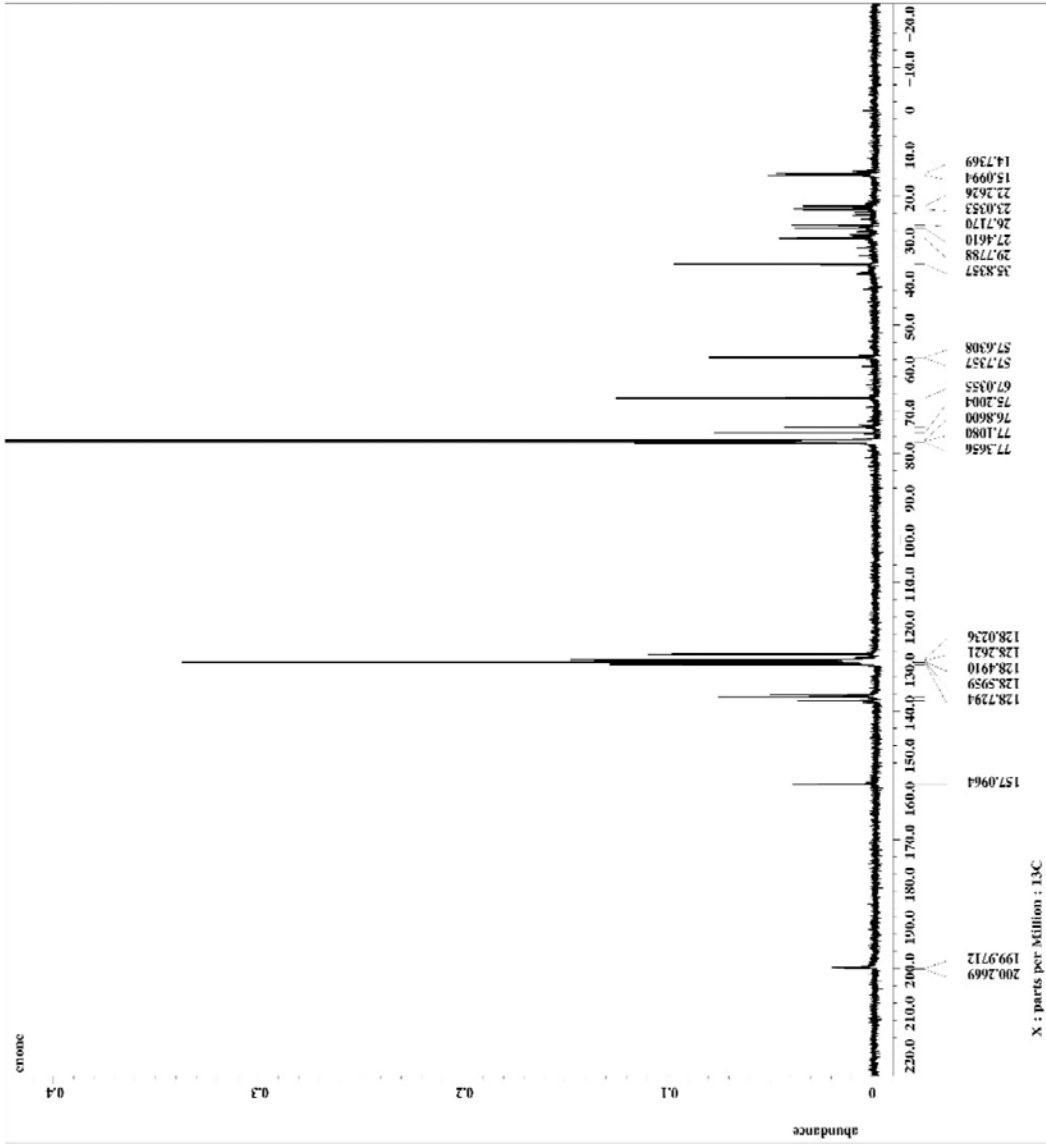




```
File: single_pulse_dec-24.3
Sample_id: Mypl31pure3crbn
Machine: mteus00sp
Creation_time: 15-MAY-2017 08:44:10
Comment: stone

Field_strength: 11.7473579[T] [500[MH
X_acq_duration: 0.83361792[s]
X_domain: 15
X_offset: 125.76529768[MHz]
X_points: 100[ppm]
X_prescans: 4
X_resolution: 1.18950024[Hz]
X_sweep_rate: 39.3081701[Hz]
Irr_domain: 1H
Irr_freq: 500.15991521[MHz]
Irr_offset: 5.0[ppm]
Clipped: FALSE
Data_return: 17254
Total_scans: 17254

X_90_width: 10.239[us]
X_acq_time: 0.83361792[s]
X_angle: 90[deg]
X_atn: 9[db]
X_pulse: 3.413[us]
Irr_atn_dec: 21.5[db]
Irr_atn_on: 21.5[db]
Irr_noise: 20[MHz]
Decoupling: WALTZ
Initial_wait: TRUE
Noe_time: TRUE
Relaxation_delay: 2[s]
Repetition_time: 2.83361792[s]
Temp_state: TEMPOFF
Temp_set: 25[dc]
Temp_get: 21.5[dc]
```



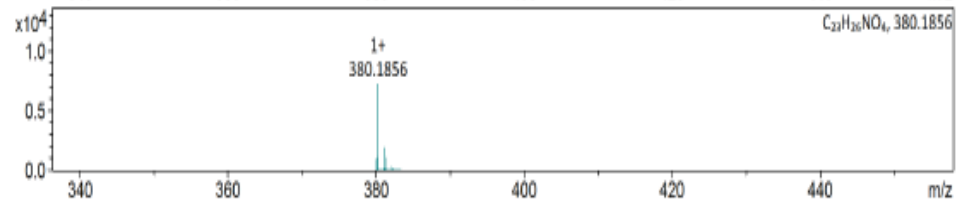
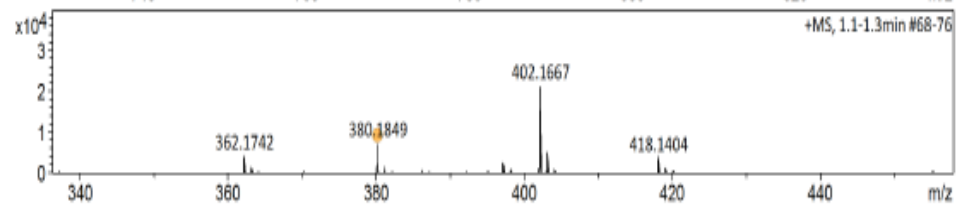
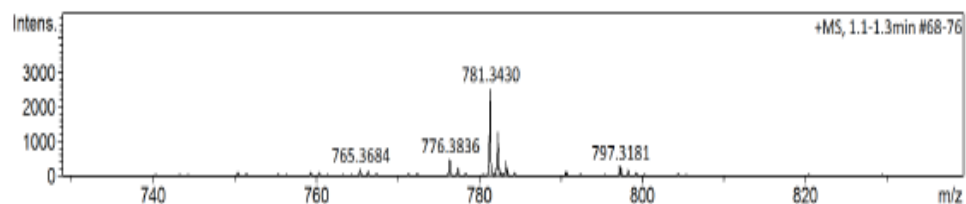
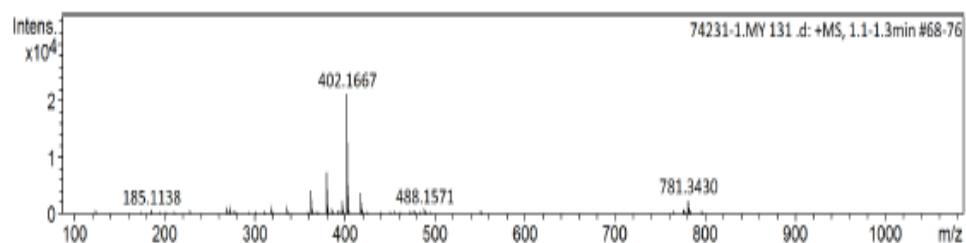
## Analysis Info

Analysis Name D:\Data\0609\74231-1.MY 131 .d  
 Method 050918\_ns\_tune\_low\_pos.m  
 Sample Name 1.MY 131  
 Comment direct infusion

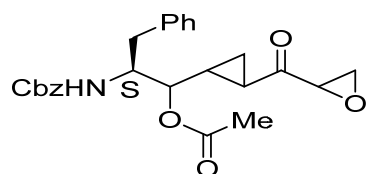
Acquisition Date 6/9/2017 3:37:57 PM  
 Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	1.5 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
380.184935	1	C <sub>23</sub> H <sub>26</sub> NO <sub>4</sub>	380.185835	1.8	2.6	11.5	ok	even



$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

HRMS

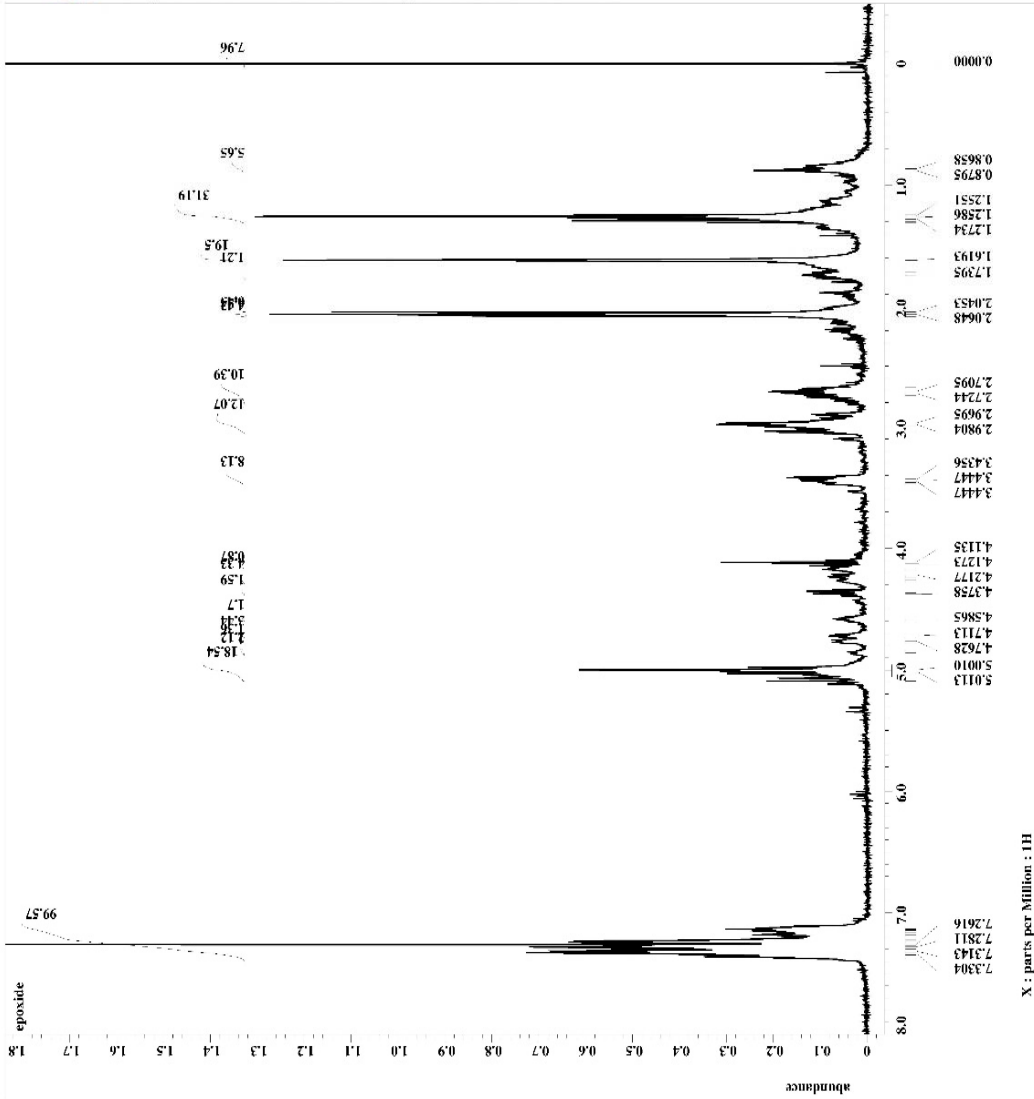


```

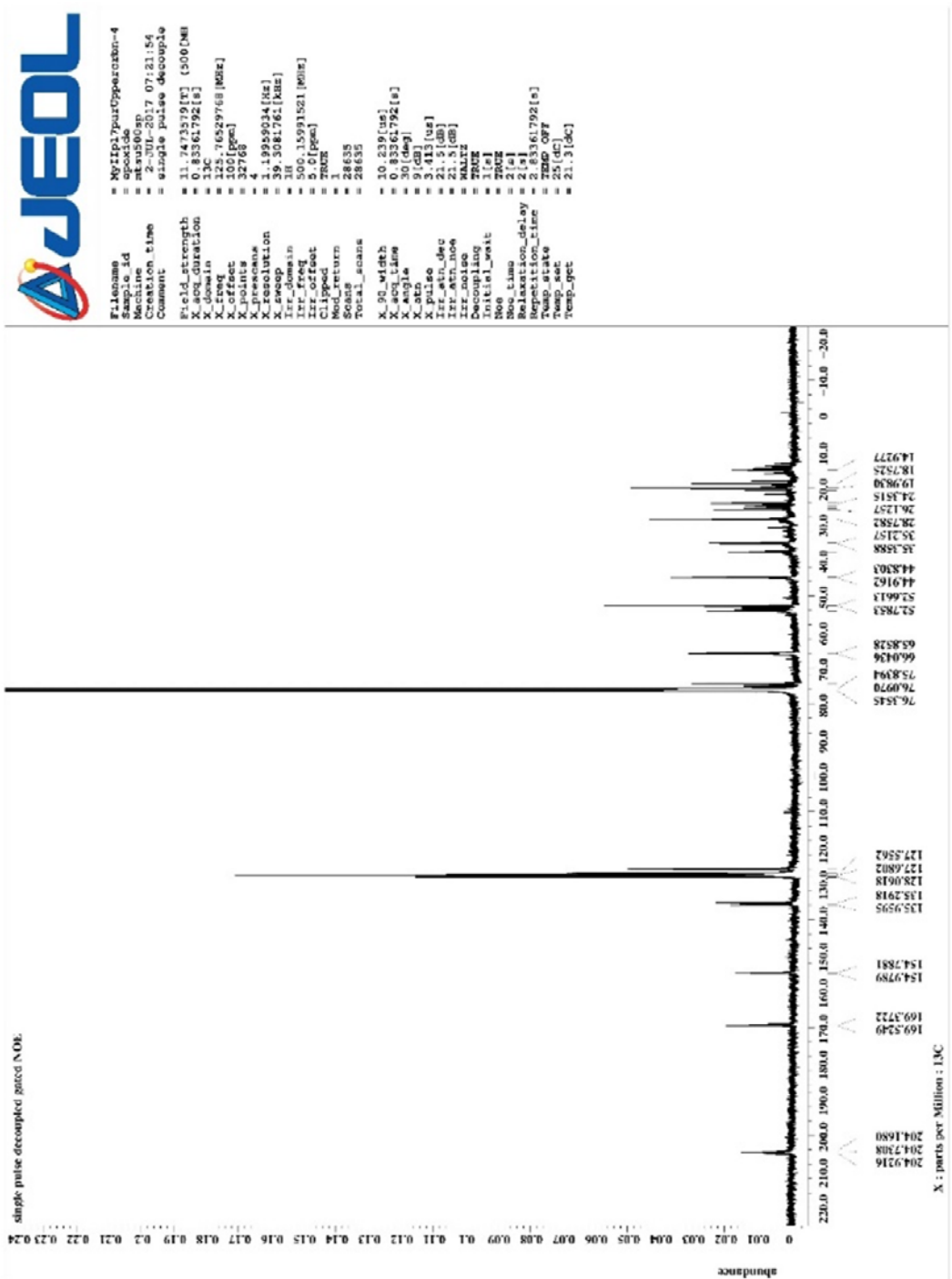
= MyJip17burUpper_PROTO
Sample_id = MyJip17burUpper
Machine = mcsu500sp
Scan_time = 2017 18:16:03
Comment = epoxide

Field_strength = 11.7473579[T] (500 MHz)
X_acq_duration = 1.74587904[s]
X_sweep = 500.15991521[MHz]
X_freq = 500.15991521[MHz]
X_offset = 5.0[ppm]
X_points = 16384
X_prescans = 1
X_resolution = 1.5727737[Hz]
X_sweep = 9.38438438[KHz]
X_sweep = 1H
Irr_domain = 500.15991521[MHz]
Irr_freq = 500.15991521[MHz]
Irr_offset = 3.0[ppm]
Irr_sweep = 1H
Tri_freq = 500.15991521[MHz]
Tri_offset = 5.0[ppm]
Clipped = FALSE
Rod_return = 1
Total_scans = 16

X_90_width = 14.19[us]
X_acq_time = 4.74587904[s]
X_points = 16384
X_snr = 4[db]
X_atn = 7.095[us]
X_pulse = Off
Irr_mode = Off
Irr_mode = Off
Irr_mode = Off
Initial_wait = 1[s]
Relaxation_delay = 4[s]
Repetition_time = 5.74587904[s]
Temp_state = 21[degC]
Temp_off = 21[degC]
Temp_get = 21.1[degC]
  
```



X : parts per Million : 1H





## Analysis Info

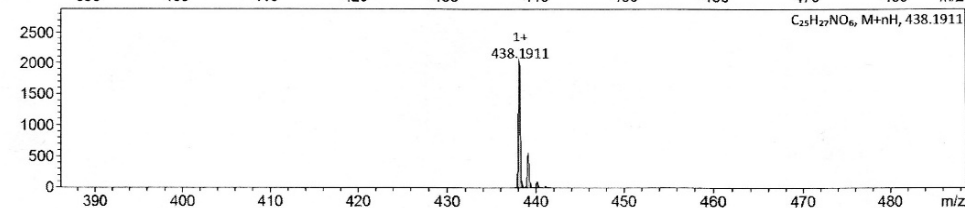
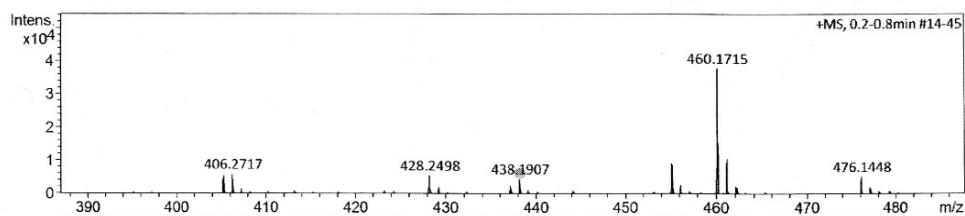
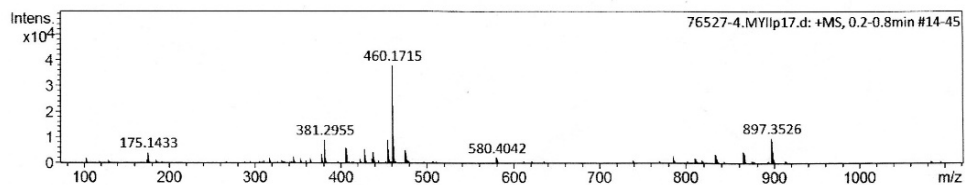
Analysis Name D:\Data\0726\76527-4.MYIIP17.d  
 Method 061517\_tune\_low\_pos\_NS.m  
 Sample Name 4.MYIIP17  
 Comment direct ifusion

Acquisition Date 7/26/2017 3:07:10 PM

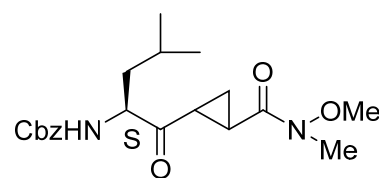
Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Waste



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
438.190724	1	C <sub>25</sub> H <sub>28</sub> NO <sub>6</sub>	438.191114	0.9	1130.7	12.5	ok	even



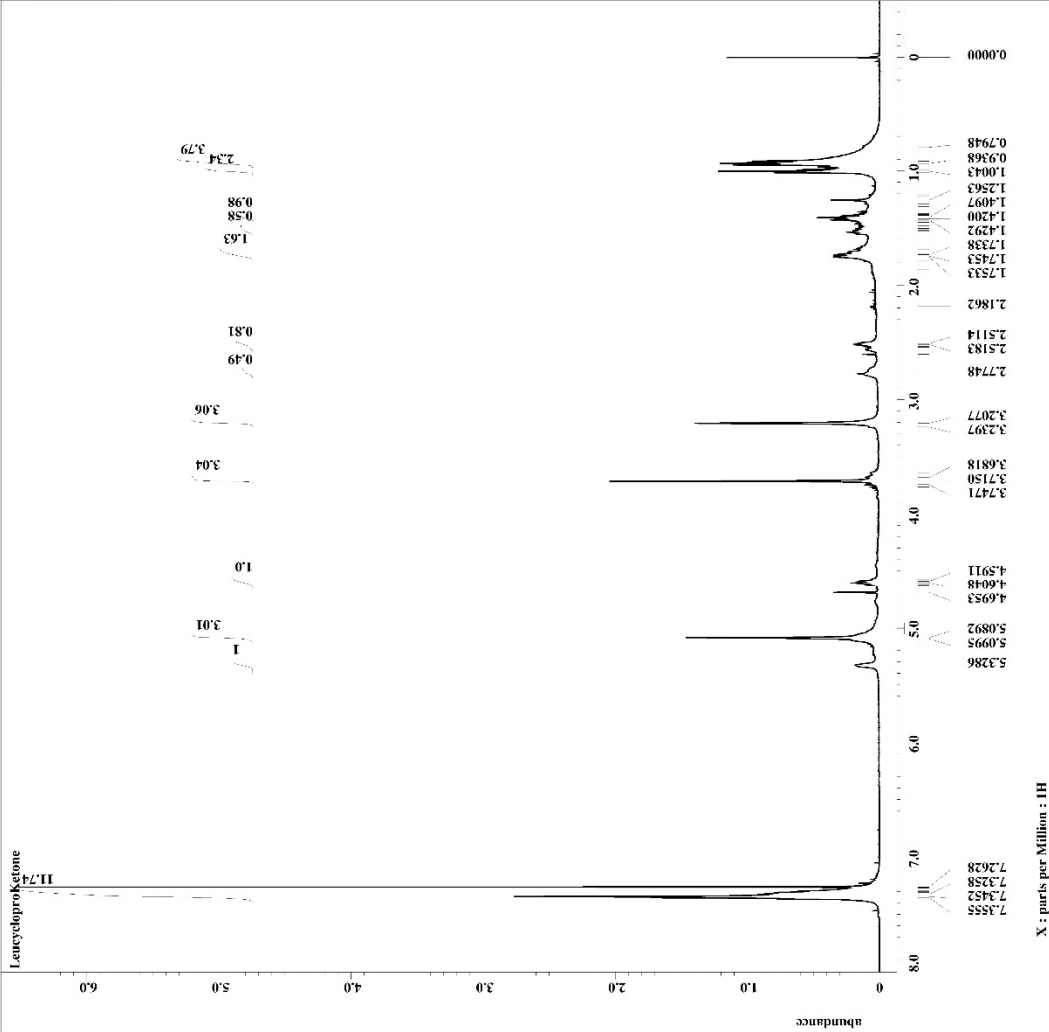
<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

HRMS



Filename = My11p4p9ureee\_PROTON-  
Sample\_id = My11p4p9ureee  
Machine = mtaus00sp  
Creation\_time = 16-APR-2018 14:04:49  
Comment = Leucycloproketone  
Field\_strength = 11.7473579[T] (500[MH  
X\_acq\_duration = 1.74587904[s]  
X\_domain = 1H  
X\_freq = 500.15991521[Mhz]  
X\_f2 = 5.0 [ppm]  
X\_points = 16384  
X\_prescans = 1  
X\_resolution = 0.5727737 [Hz]  
X\_sweep = 9.38438438 [KHz]  
X\_sweep\_rate = 5.38 [ppm/s]  
Irr\_domain = 500.15991521 [Mhz]  
Irr\_freq = 5.0 [ppm]  
Irr\_offset = 1H  
Tri\_domain = 500.15991521 [Mhz]  
Tri\_freq = 5.0 [ppm]  
Tri\_offset = FALSE  
Mod\_return = 1  
Total\_scans = 16  
X\_90\_width = 14.19 [us]  
X\_acq\_time = 1.74587904 [s]  
X\_angle = 45 [deg]  
X\_atn = 4 [dB]  
X\_pulse = 7.095 [us]  
X\_mode = OFZ  
Tri\_mode = OFZ  
Dante\_presat = FALSE  
Initial\_wait = 1 [s]  
Relaxation\_delay = 4 [s]  
Acquisition\_time = 1.74587904 [s]  
Temp\_state = TEMP\_OFF  
Temp\_set = 25 [dC]  
Temp\_get = 22.6 [dC]



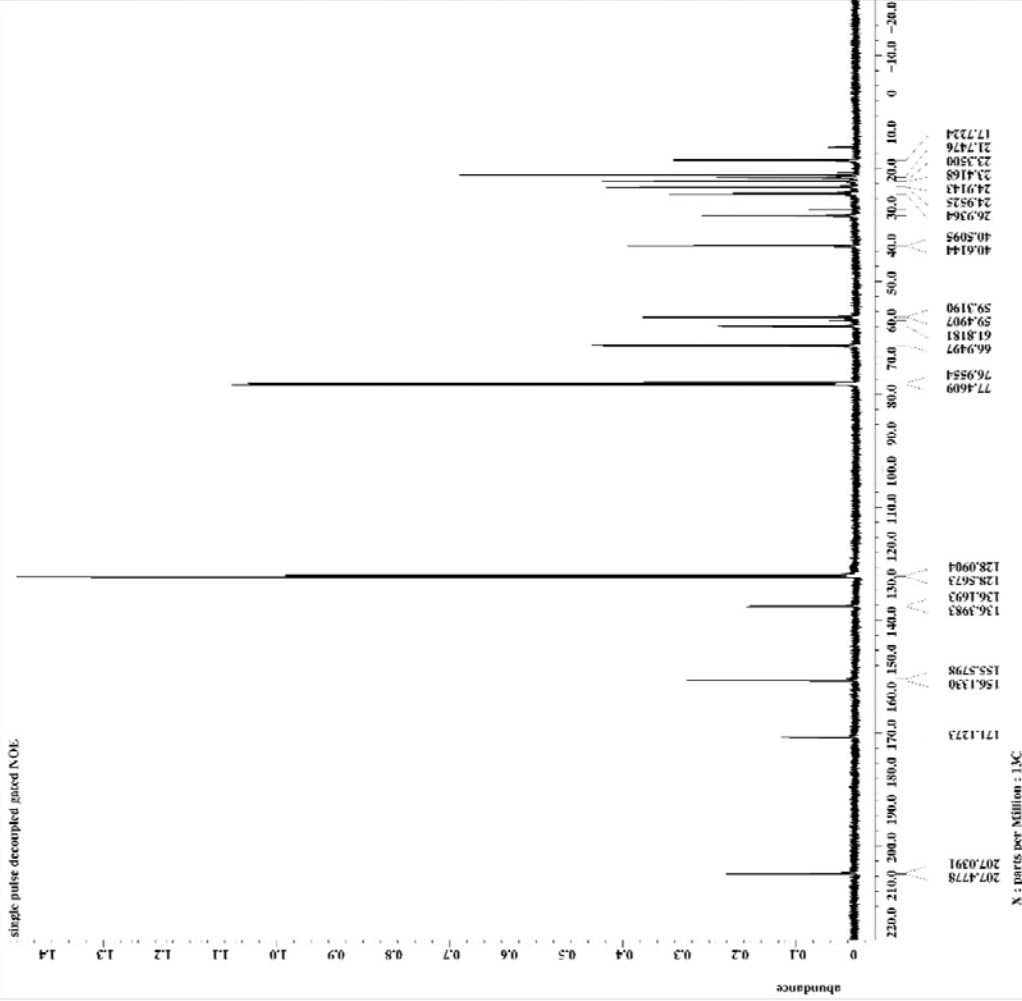
X : parts per Million : 1H



```

Filename = MyIip27Carbon-4.jdf
Sample_id = Leu.Cyclpropt
Acquire = 00:00:00
Date_time = 14-DEC-2017 07:23:59
Comment = single pulse decouple
Field_strength = 11.7473792 [T] (500[M]B
X_coordination = 135.3561792 [s]
X_freq = 125.76529768 [MHz]
X_offset = 100 [ppm]
X_points = 32768
X_resolution = 1.19959034 [Hz]
X_sweep = 39.3081761 [kHz]
Irr_domain = 1R
Irr_freq = 500.15991521 [MHz]
Irr_offset = 0 [ppm]
Mod_return = 1
Total_scans = 16214
X_90_width = 10.239 [us]
X_acq_time = 0.83361792 [s]
X_angle = 30 [deg]
X_atn = 3 [dB]
X_atn_time = 3 [us]
Irr_atn_dec = 21.5 [dB]
Irr_atn_noise = 21.5 [dB]
Irr_noise = WALTZ
Spectrum = 1R
Initial_wait = 1 [s]
Noe_time = 7200
Noe_time = 2 [s]
Relaxation_delay = 2 [s]
Spectrum_time = 61.792 [s]
Temp_set = 7200 [C]
Temp_get = 25 [C]
Temp_set = 22.9 [C]
Temp_get = 22.9 [C]

```



## Analysis Info

Analysis Name D:\Data\0208\86718-1.MY\lp27.d  
 Method 0118\_tune\_low\_pos\_NS.m  
 Sample Name 1.MY\lp27  
 Comment

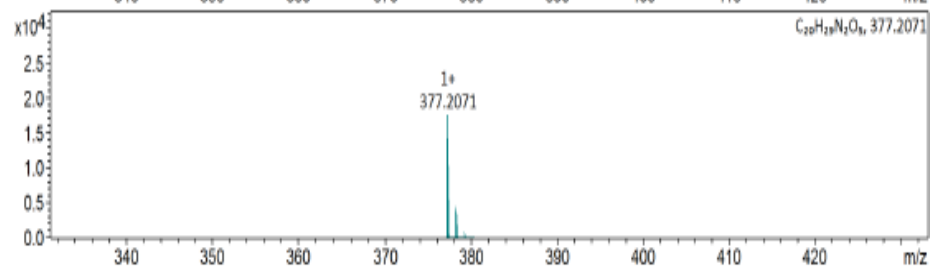
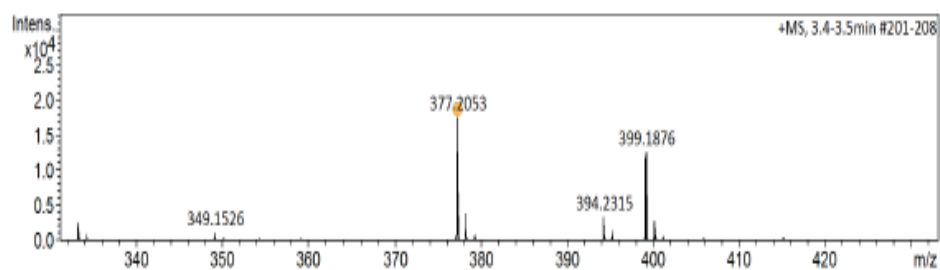
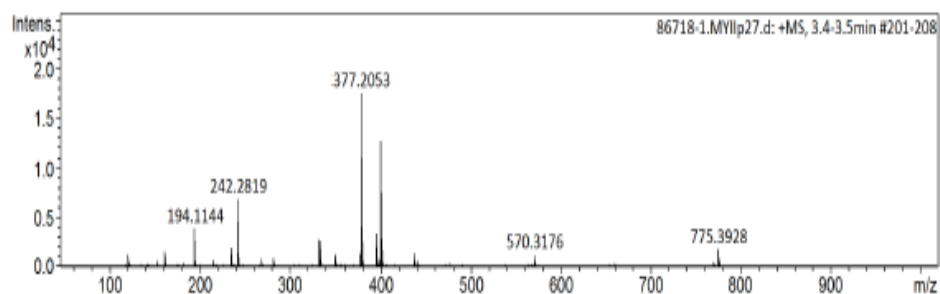
Acquisition Date 2/6/2018 2:23:15 PM

Operator BDAL@DE

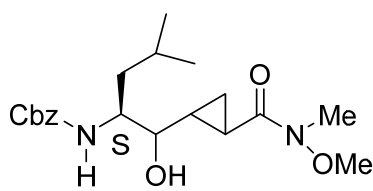
Instrument / Ser# micrOTOF II 8213750.1  
0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
377.205315	1	C <sub>20</sub> H <sub>29</sub> N <sub>2</sub> O <sub>5</sub>	377.207098	4.7	4.8	7.5	ok	even



$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

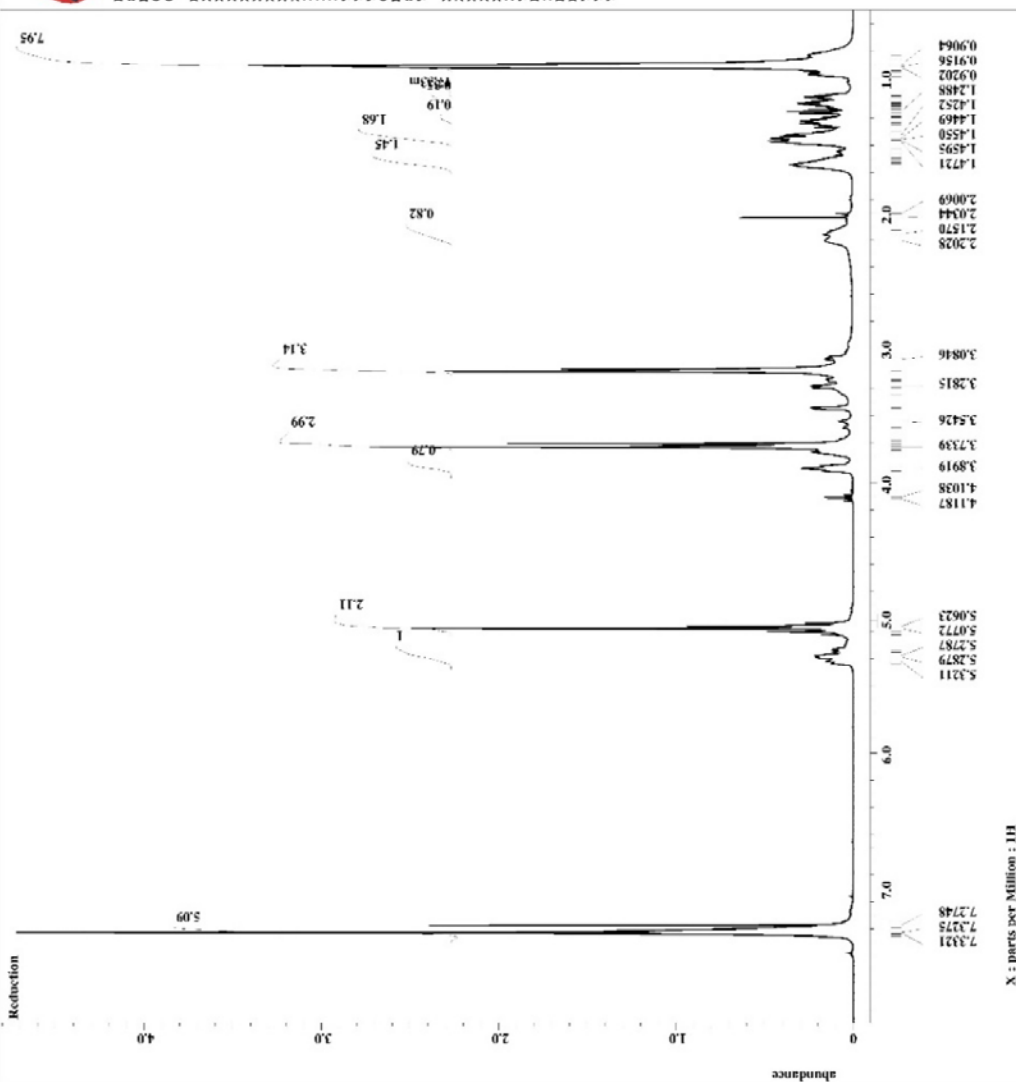
$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

HRMS



```

Filename = MyTip35pure_PROTON-6.
Sample_id = MyTip35pure
Acquire_date = 20170801
Creation_time = 11:00:20.17 16:03:52
Comment = Reduction
Field_strength = 11.7473579[T] (500 [Mm])
X_acq_direction = 1H
X_dom_freq = 1H 4587904[Hz]
X_freq = 500.15991521[MHz]
X_offset = 5.0[ppm]
X_points = 16384
X_resolution = 0.9277737[Hz]
X_sweep = 9.38438438[kHz]
Irr_domain = 1R
Irr_freq = 500.15991521[MHz]
Irr_offset = 5.0[ppm]
Tri_domain = 1R
Tri_freq = 500.15991521[MHz]
Tri_offset = 5.0[ppm]
X_ppm = 1
Mod_return = 1
Mod_start = 1
Mod_end = 16
Total_scans = 16
X_90_width = 14.19[us]
X_acq_time = 1.74587904[s]
X_angle = 45[deg]
X_atn = 4[db]
X_pulse = 0.1095[us]
Tri_mode = OFF
Dante_preset = FALSE
Initial_wait = 1[s]
Relaxation_delay = 1[s]
Temp_ctrl_time = 5.714587904[s]
Temp_state = TEMPO OFF
Temp_set = 25[degC]
Temp_get = 23[degC]
    
```



X : parts per Million : 1H

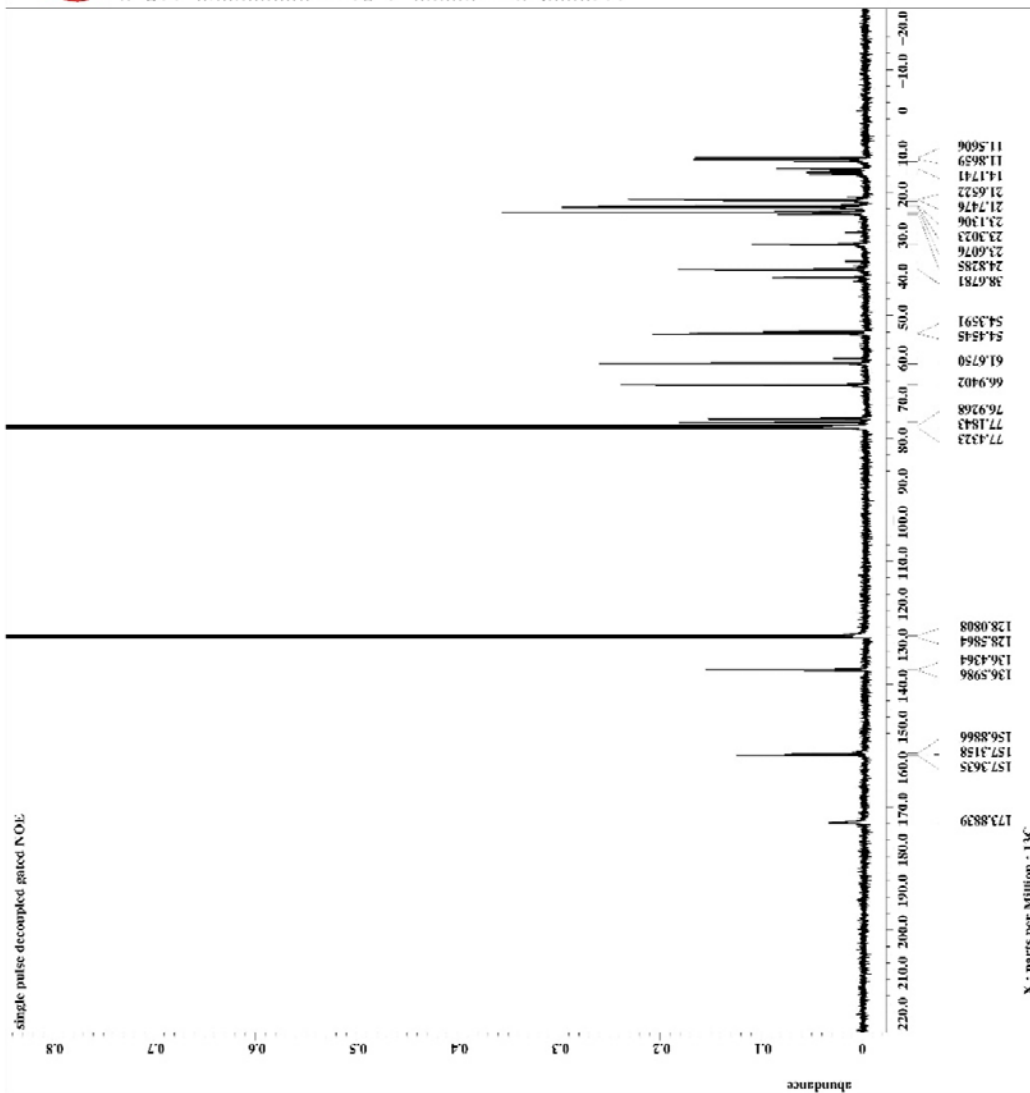


```

File Name      = Nv1p35Carbon-3.jdf
Sample ID     = Redn cycl. propmt
Machine       = msv500sp
Preamplifier = 19.4C.09
Pulse Program = 35CP-21
Comment       = single pulse decouple

Field strength = 11.747378[T] (500 MHz)
X_acq_duration = 0.83361792[s]
X_domain       = 125.76529768 [MHz]
X_freq         = 100 [ppm]
X_offset       = 32768
X_points       = 40
X_prescans     = 40
X_resolution   = 39.5919034 [Hz]
X_sweep        = 39.5081761 [kHz]
IRF_domain     = 1H
IRF_freq       = 500.15991521 [MHz]
IRF_offset     = 5.0 [ppm]
NUC1           = 13C
NUC2           = 1
NUC3           = 1
Mod           = 1
ModReturn      = 1
Total_scans    = 4375

X_90_width     = 10.238 [us]
X_acq_time     = 0.83361792 [s]
X_angle        = 30 [deg]
X_atn          = 9 [dB]
X_pulse        = 3.413 [us]
X_pulse_dec    = 21.5 [dB]
IRF_atn_dec    = 21.5 [dB]
IRF_noise      = WALZ
Decoupling     = TRUE
Initial_wait   = 1 [s]
Noe_time       = 2 [s]
Relaxation_delay = 2 [s]
Repetition_time = 2.83361792 [s]
Temp_state     = 299.0 [K]
Temp_set       = 23 [C]
Temp_get       = 23 [C]
    
```



X : parts per Million : 13C



## Analysis Info

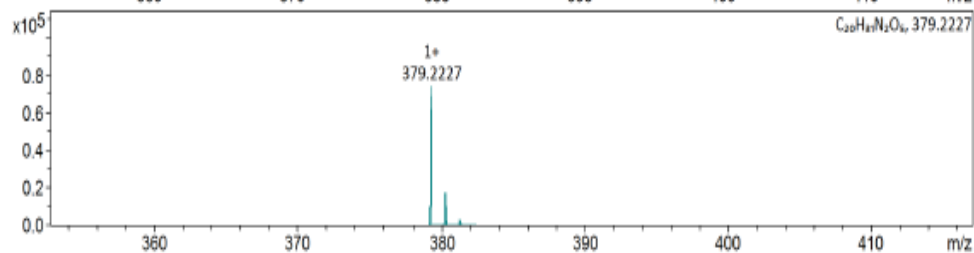
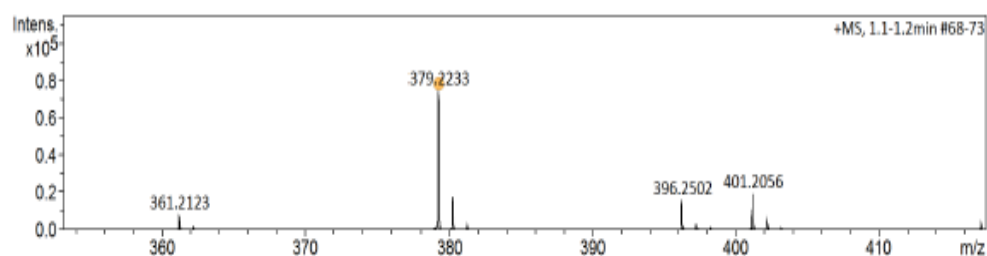
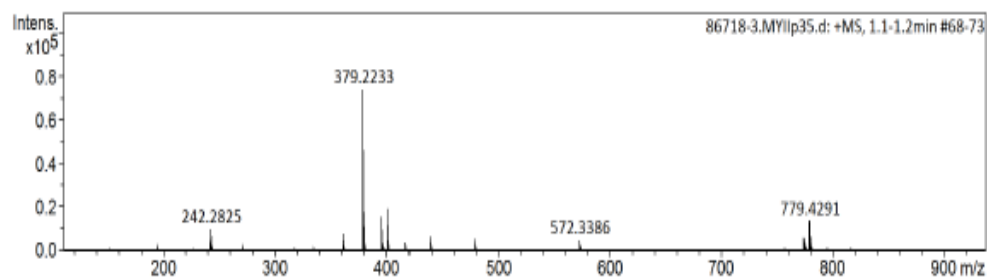
Analysis Name D:\Data\0206\86718-3.MYIIP35.d  
 Method 0118\_tune\_low\_pos\_NS.m  
 Sample Name 3.MYIIP35  
 Comment

Acquisition Date 2/6/2018 2:37:08 PM

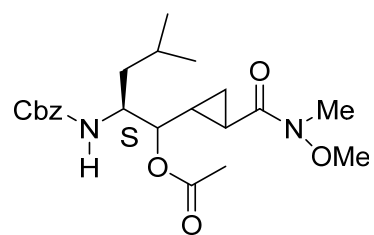
Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source

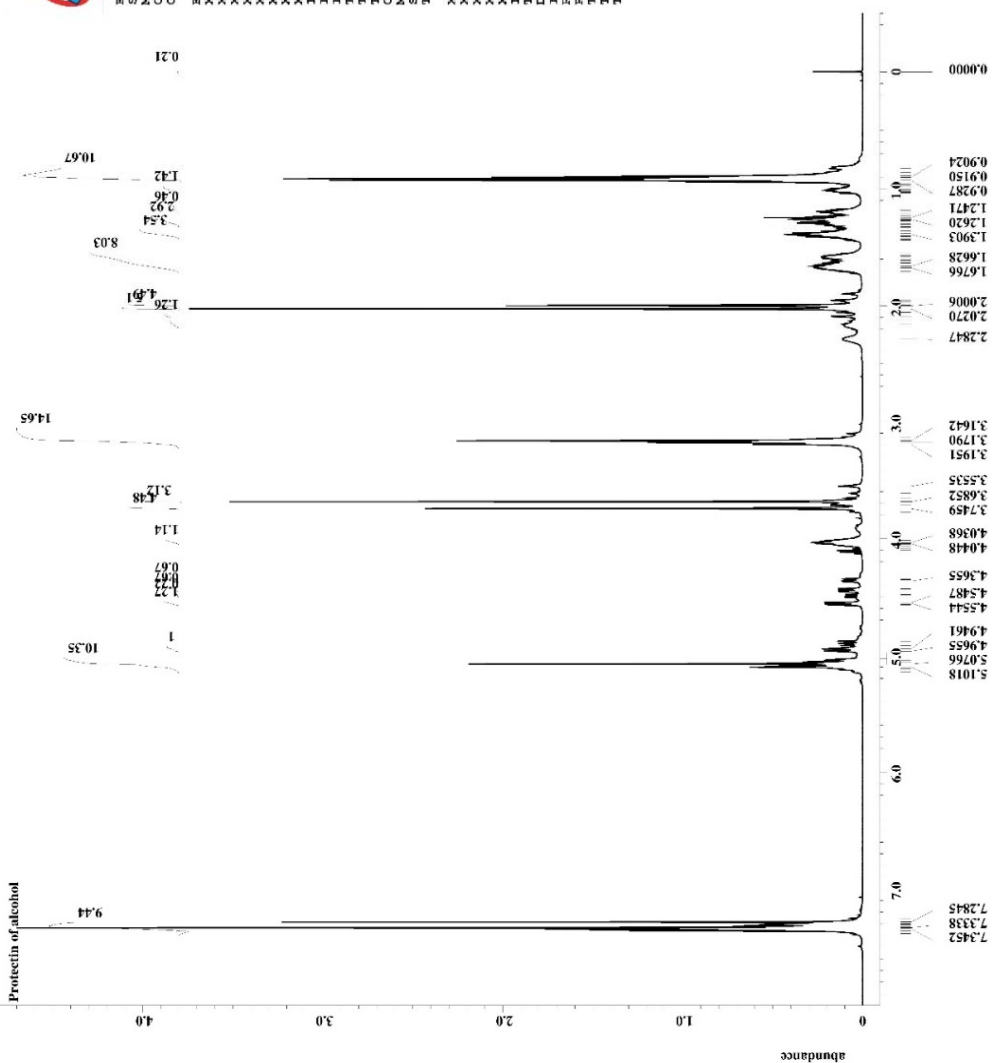


Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdB	N-Rule	e <sup>-</sup>	Conf
379.223273	1	C20H31N2O5	379.222749	-1.4	-0.7	6.5	ok	even	



$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

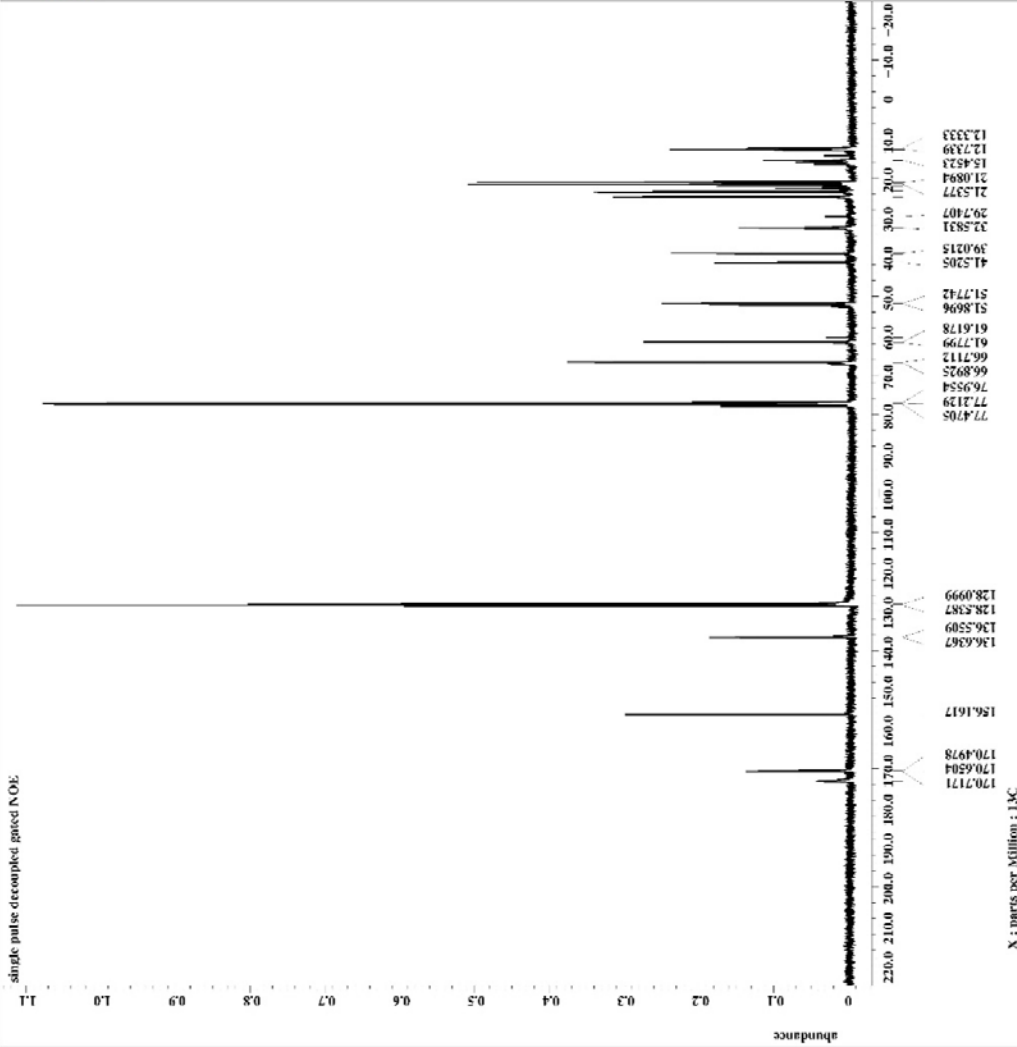
$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

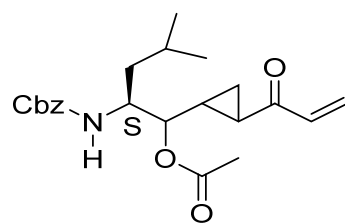


X : parts per Million : 1H



Filename = MyIsp37Carbon-3-.idf  
Sample\_id = Protection alcohol  
Acquisition\_date = 19-Oct-2017 06:02:31  
Creation\_time = single pulse decouple  
Comment =  
Field\_strength = 11.747399[T] (500[MHz])  
X\_coordination = 136  
X\_domain = 136.8361792 [s]  
X\_freq = 125.76529768 [MHz]  
X\_offset = 100 [ppm]  
X\_points = 32768  
X\_resolution = 1.19959034 [Hz]  
X\_sweep = 39.3081761 [kHz]  
Irr\_domain = IR  
Irr\_freq = 500.15991521 [MHz]  
Irr\_offset = 0 [ppm]  
Clipped = 1  
Mod\_return = 1  
Scans = 13156  
Total\_scans = 13156  
X\_90\_width = 10.239 [us]  
X\_acq\_time = 0.83361792 [s]  
X\_angle = 30 [deg]  
X\_atn = 1  
X\_pulse = 3.413 [us]  
Irr\_atn\_dec = 21.5 [dB]  
Irr\_atn\_noe = 21.5 [dB]  
Irr\_noise = WALTZ  
Spectrum = 1  
Initial\_wait = 1 [s]  
Noe = TRUE  
Noe\_time = 2 [s]  
Relaxation\_delay = 2 [s]  
Relaxation\_time = 0.83361792 [s]  
Temp\_state = TEMPO OFF  
Temp\_set = 25 [dc]  
Temp\_set = 23.7 [dc]



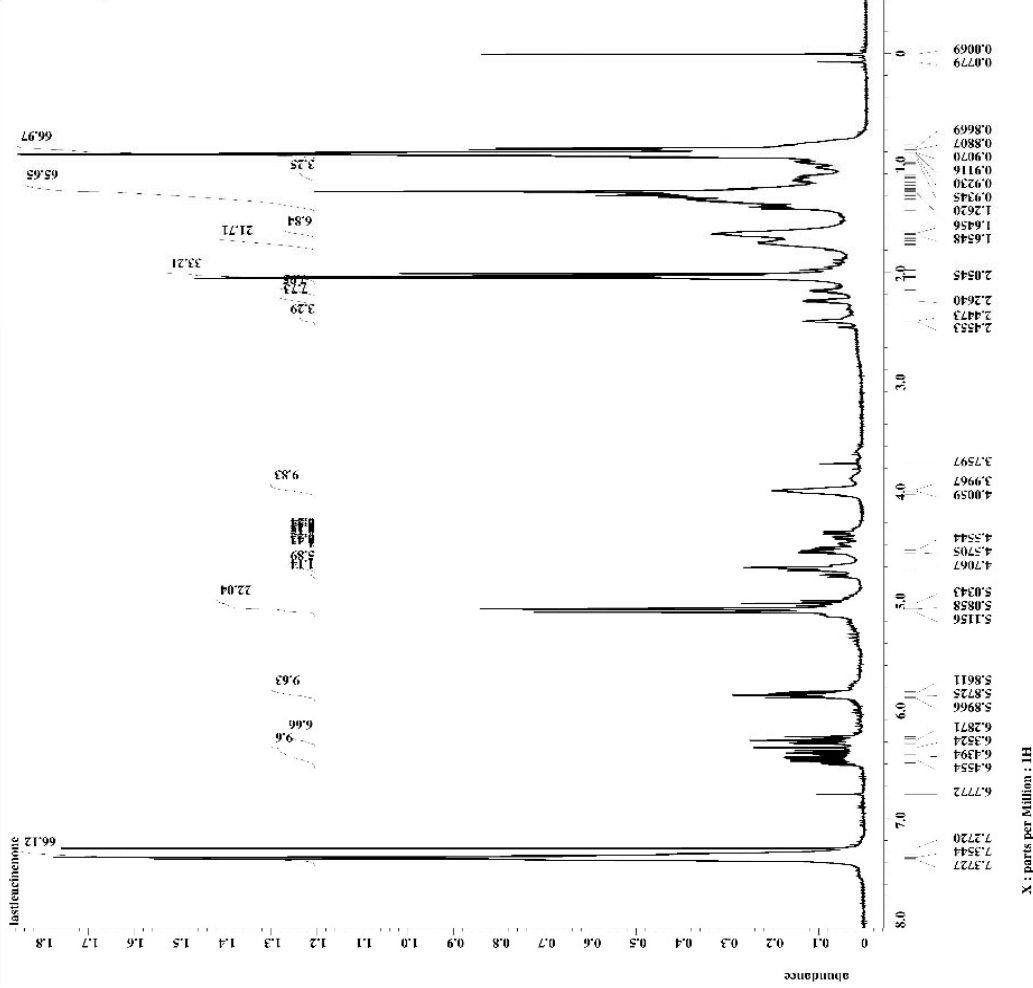


<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

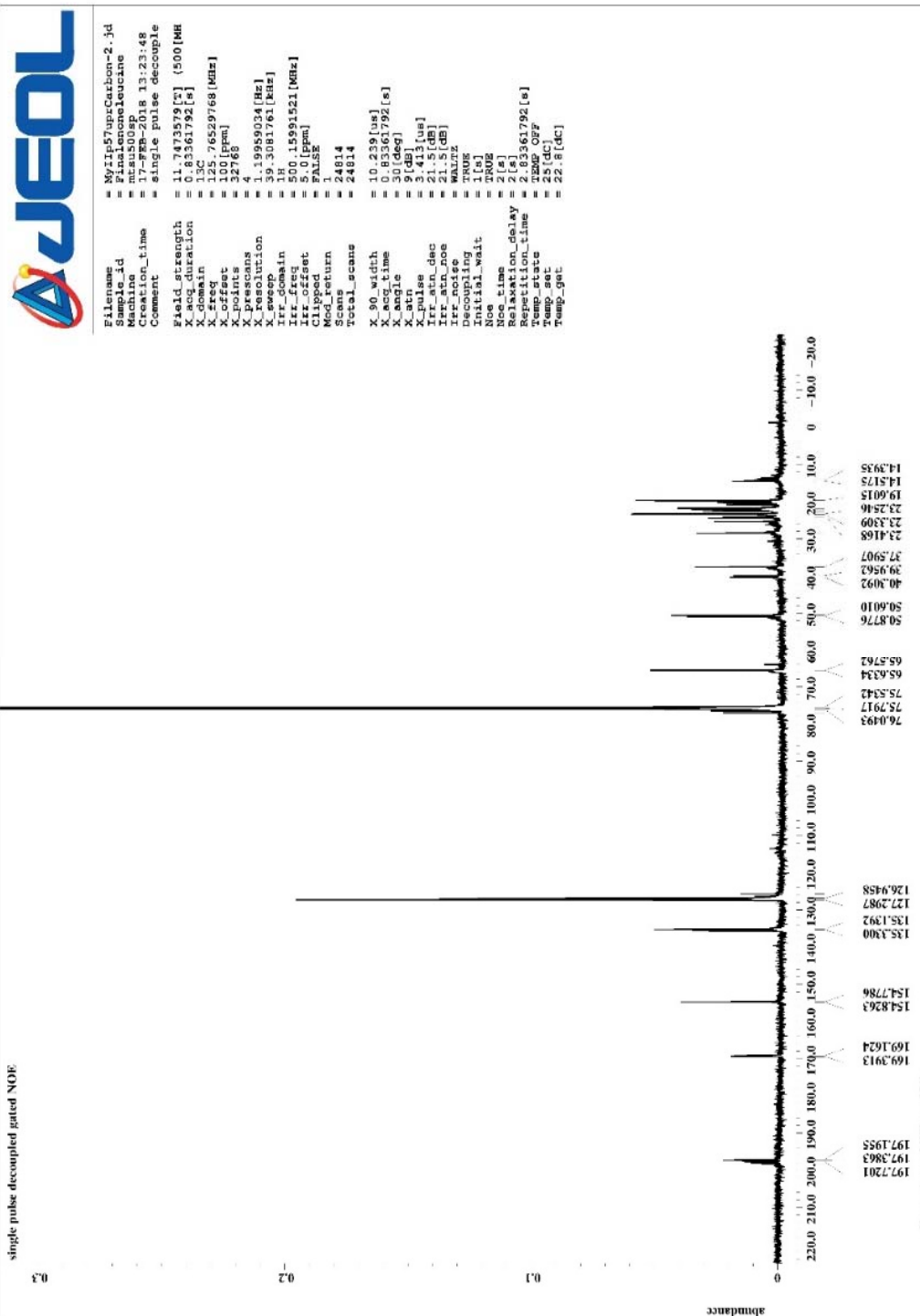
<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

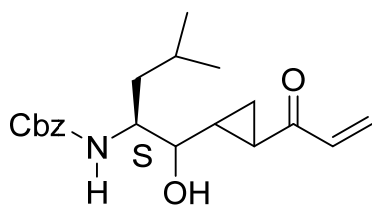


Filename = MyIps7uppreproton  
Sample\_id = MyIps7uppreproton  
Machine = msc500ep  
Creation\_time = 13/05/2018 17:59:54  
Comment = test1234567890  
Field\_strength = 11.7473575 [T] (500 MHz)  
X\_scs\_duration = 1.74587904 [s]  
X\_scs\_delay = 500.15991521 [MHz]  
X\_freq = 500.15991521 [MHz]  
X\_offset = 5.0 [ppm]  
X\_points = 16384  
X\_prescans = 1  
X\_resolution = 1.5727727 [Hz]  
X\_resolution\_h = 0.38638438 [kHz]  
Irr\_domain = 1H  
Irr\_freq = 500.15991521 [MHz]  
Irr\_offset = 5.0 [ppm]  
Irr\_domain = 1H  
Irr\_offset = 5.0 [ppm]  
T1a\_offset = 500.15991521 [MHz]  
T1a\_offset = 5.0 [ppm]  
Clipped = FALSE  
Mod\_return = 1  
Scans = 16  
Total\_scans = 16  
X\_90\_width = 14.15 [us]  
X\_scs\_time = 1.74587904 [s]  
X\_pulse = 4.0 [us]  
X\_angle = 4.0 [deg]  
X\_pulse = 7.095 [us]  
Irr\_mode = Off  
T1a\_mode = Off  
T1a\_delay = 1.0 [us]  
T1a\_offset = 1.0 [us]  
Relaxation\_delay = 4 [s]  
Repetition\_time = 5.74587904 [s]  
Temp\_set = TEMP OFF  
Temp\_set = 22.7 [degC]



1H : parts per Million : X



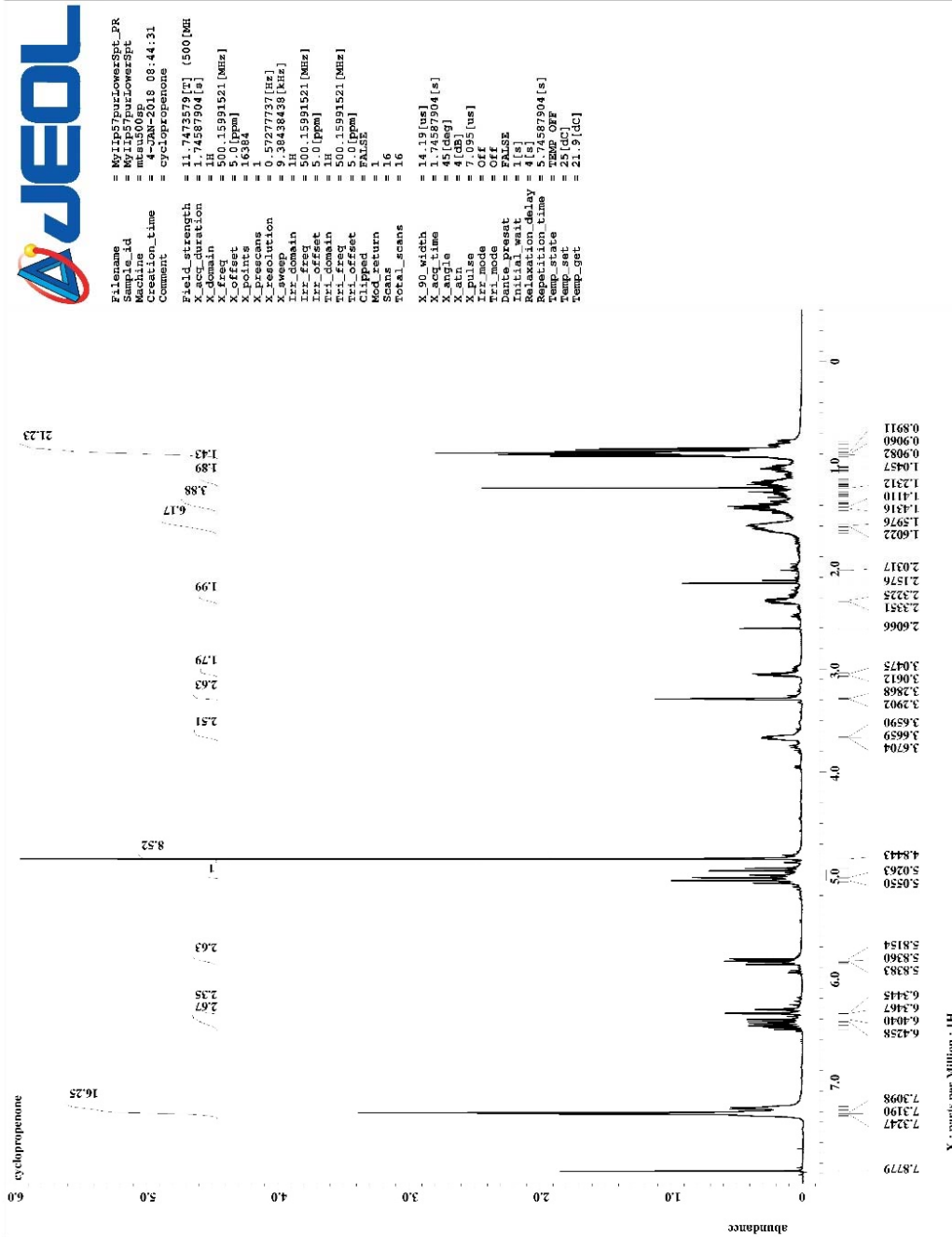


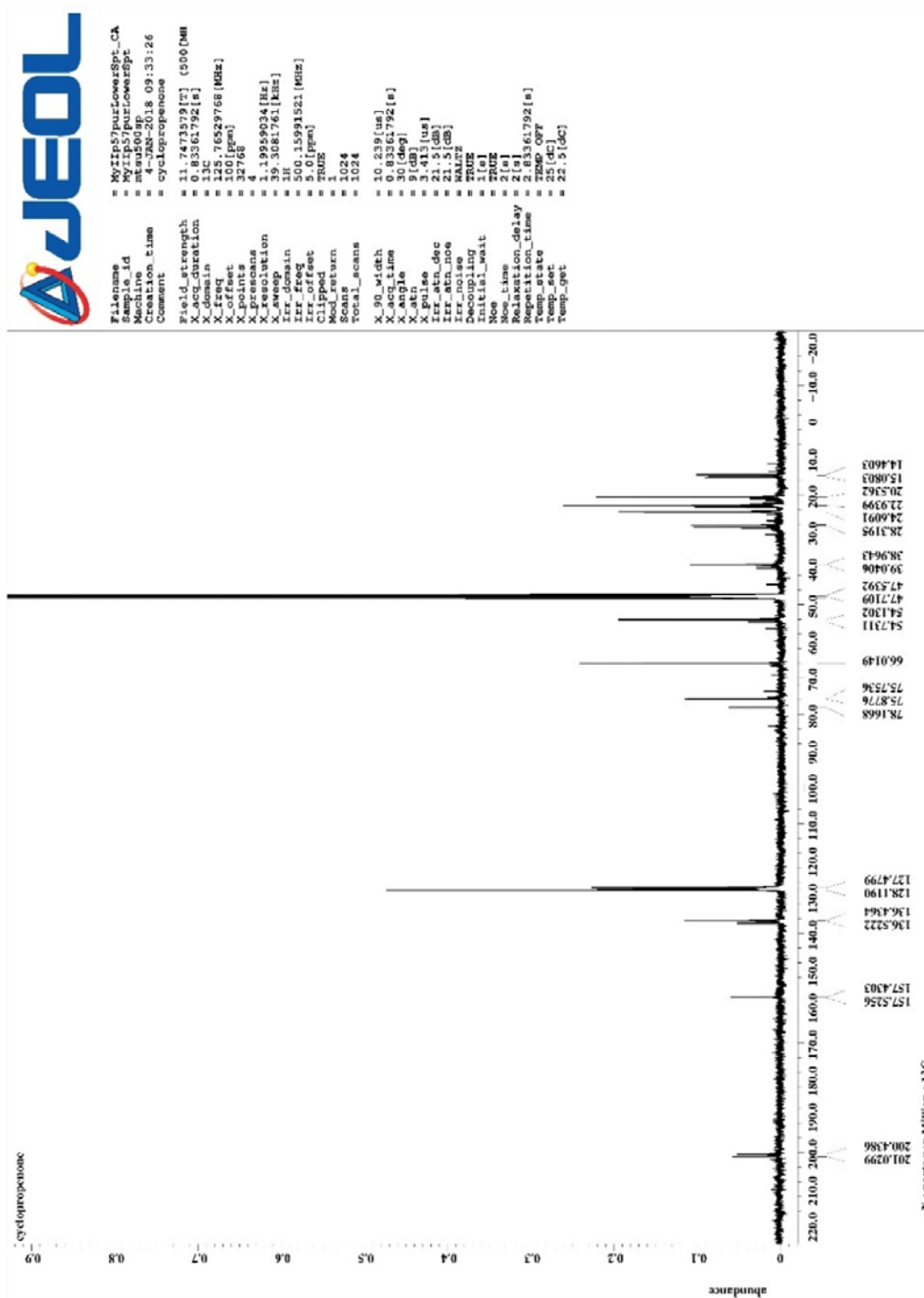
$^1\text{H-NMR}$  (500 MHz,  $\text{CD}_3\text{OD}$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CD}_3\text{OD}$ )

HRMS







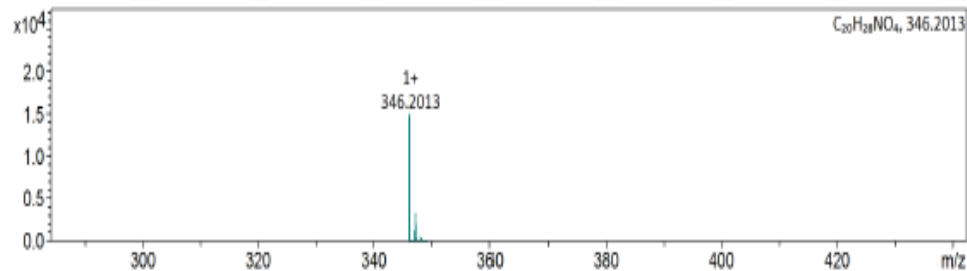
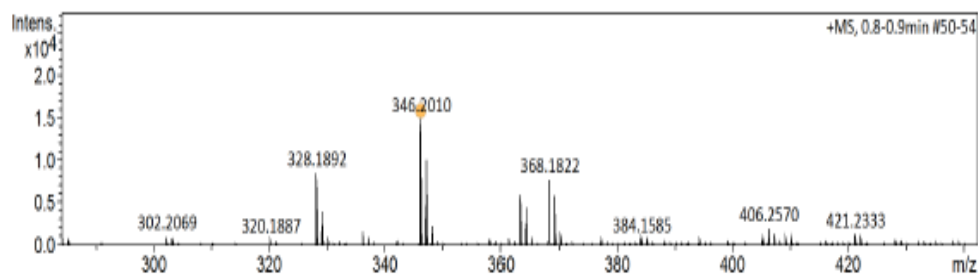
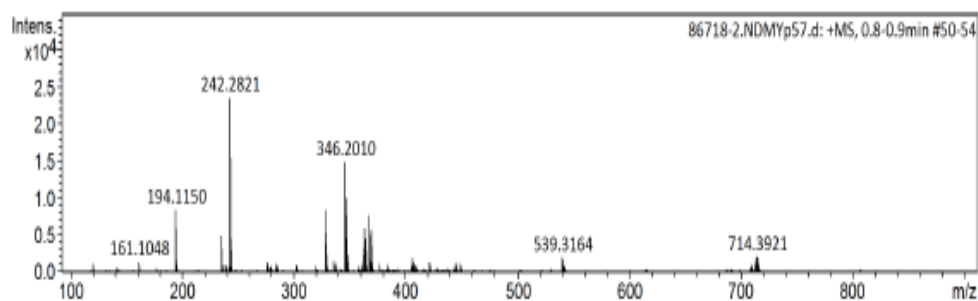
## Analysis Info

Analysis Name D:\Data\0206\86718-2.NDMYp57.d  
 Method 0118\_tune\_low\_pos\_NS.m  
 Sample Name 2.NDMYp57  
 Comment

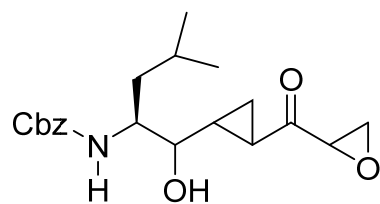
Acquisition Date 2/6/2018 2:31:06 PM  
 Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
346.200987	1	C <sub>20</sub> H <sub>28</sub> NO <sub>4</sub>	346.201285	0.9	-2.1	7.5	ok	even



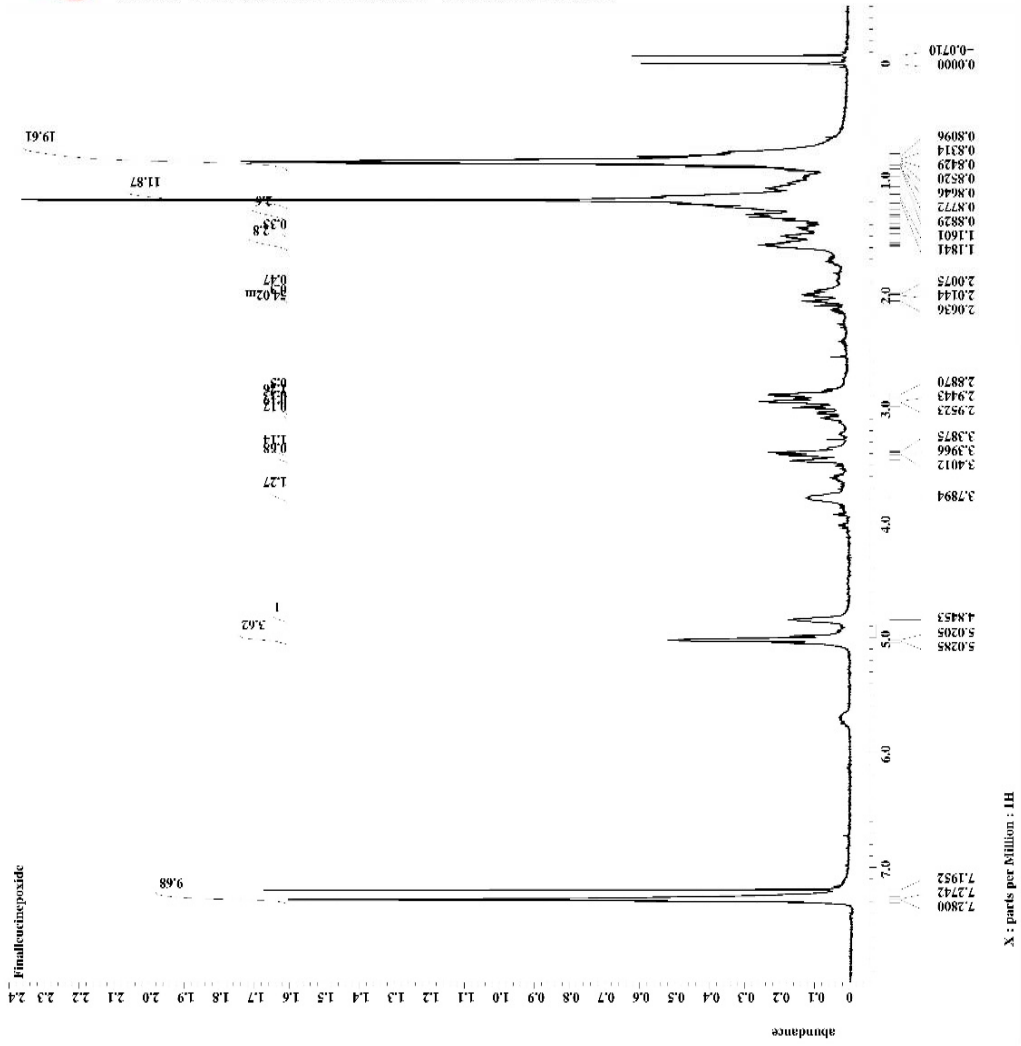
$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )



File Name = Mtl188burcense\_FRONON  
 Sample Id = Mtl188burcense  
 Machine = mtau500p  
 Creation Time = 18-FEB-2018 18:09:40  
 Comment = Finalleucinepoixide

Field strength = 11.7473579 [T] (500 [MH])  
 X\_acq\_duration = 1.74587904 [s]  
 X\_domain = 18  
 X\_freq = 500.15891521 [MHz]  
 X\_offset = 16384  
 X\_points = 1  
 X\_prescans = 0.5727737 [Hz]  
 X\_resolution = 1838438438 [Hz]  
 X\_sweep = 1838438438 [Hz]  
 Irr\_domain = 18  
 Irr\_freq = 500.15891521 [MHz]  
 Irr\_offset = 5.0 [ppm]  
 Irr\_domain = 18  
 Irr\_offset = 5.0 [ppm]  
 Clipped = FALSE  
 Mod\_return = 1  
 Scans = 16  
 Total\_scans = 16  
 X\_90\_width = 14.19 [us]  
 X\_acq\_time = 5.74587904 [s]  
 X\_pulse = 4 [dB]  
 X\_sfn = 7.095 [us]  
 X\_pulse = off  
 Irr\_mode = OFF  
 Irr\_offset = OFF  
 Initial\_wait = 1 [s]  
 Relaxation\_delay = 4 [s]  
 Repetition\_time = 5.74587904 [s]  
 Temp = 25 [C]  
 Temp\_set = 25 [dC]  
 Temp\_get = 22.6 [dC]



HI : X : parts per Billion : HI

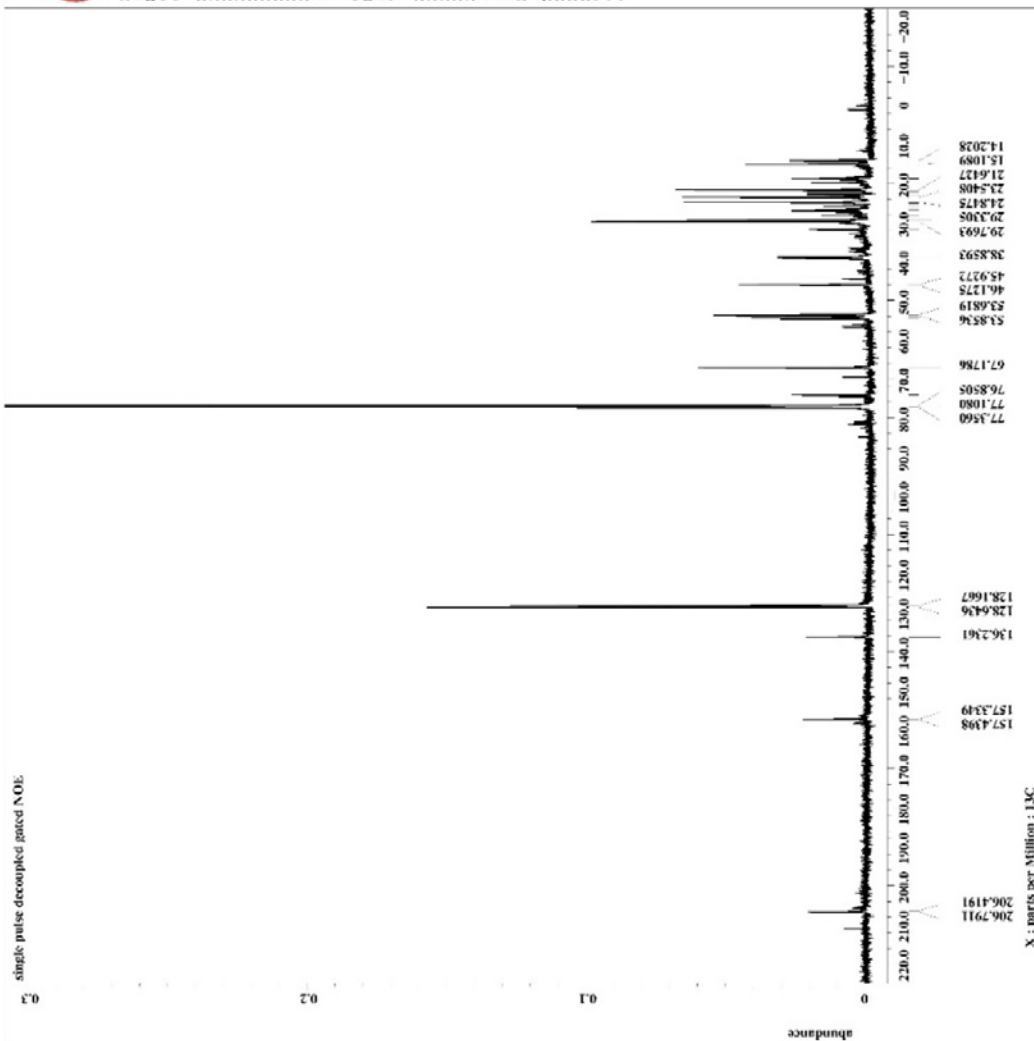


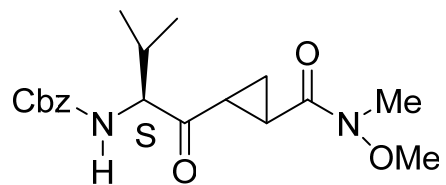
```

=====
Filename = MyTIPSCarbon-4.jdf
Sample_id = F:malleicaneoxide
Machine = straus009p
Acq_date_time = 20080807 01:03:59
Comment = single pulse decouple

Field_strength = 11.7473579[T] (500 MHz)
X_acq_duration = 0.83361792[s]
X_sweep = 32768
X_freq = 125.76529768 [MHz]
X_points = 32768
X_prescans = 4
X_resolution = 1.9858034 [Hz]
X_swap = 39.3081761 [kHz]
Irr_domain = 3H
Irr_freq = 500.15991521 [MHz]
Irr_offset = 5.0 [ppm]
C1p1_offset = 1
Mod1p1turn = 1
Scans = 47288
Total_scans = 47288

X_90_width = 10.239 [us]
X_acq_time = 0.83361792 [s]
X_angle = 30 [deg]
X_atn = 9 [dB]
X_pulse = 3.413 [us]
Irr_angle = 30 [deg]
Irr_atn = 21.5 [dB]
Irr_noise = WALTZ
Decoupling = TRUE
Initial_wait = 1 [s]
X_180_time = 2 [us]
X_180_angle = 180 [deg]
X_180_atn = 21 [dB]
Relaxation_delay = 2 [s]
Repetition_time = 2.83361792 [s]
Temp_state = 298K OFF
Temp_set = 298.15 [K]
Temp_get = 22.8 [degC]
=====
    
```

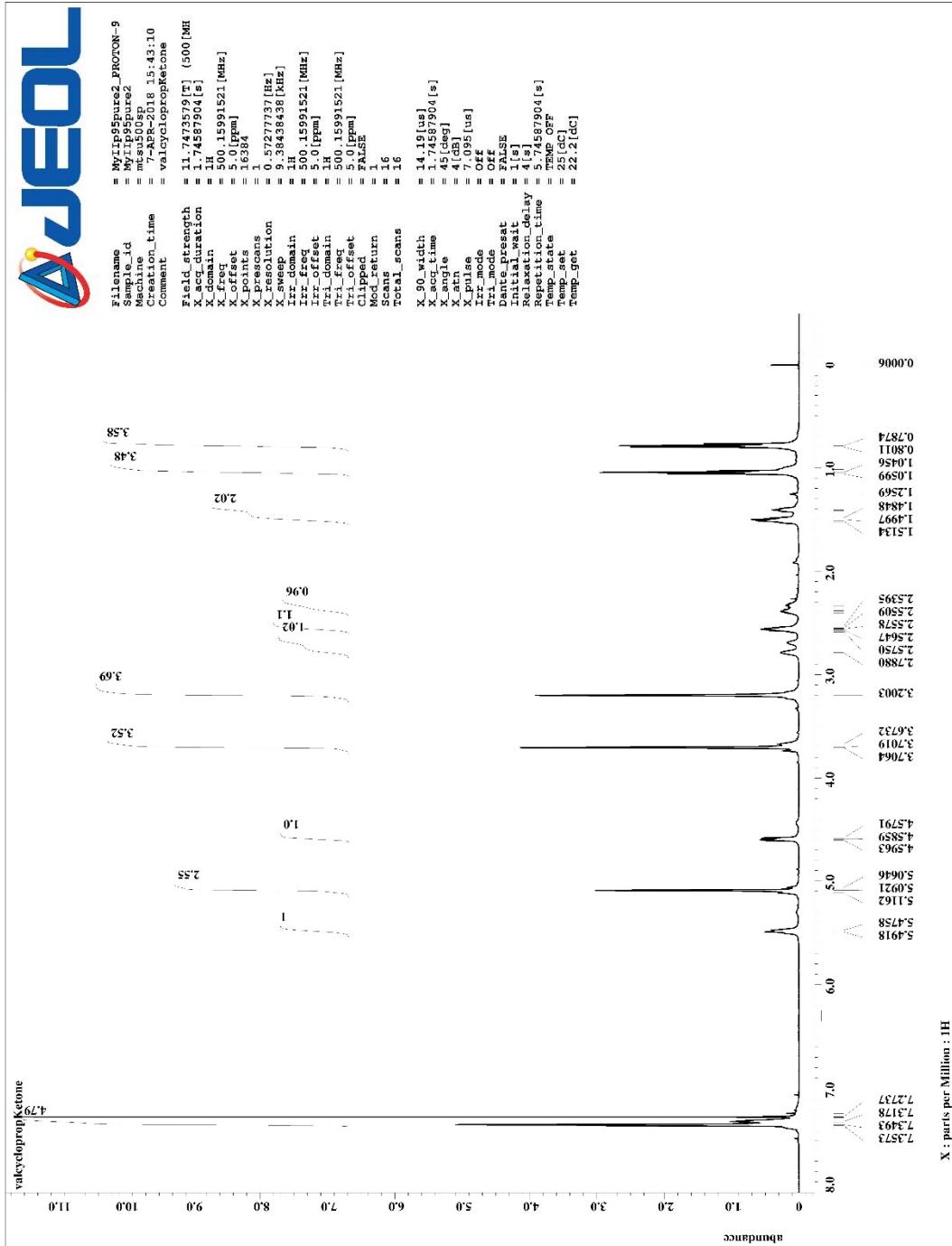




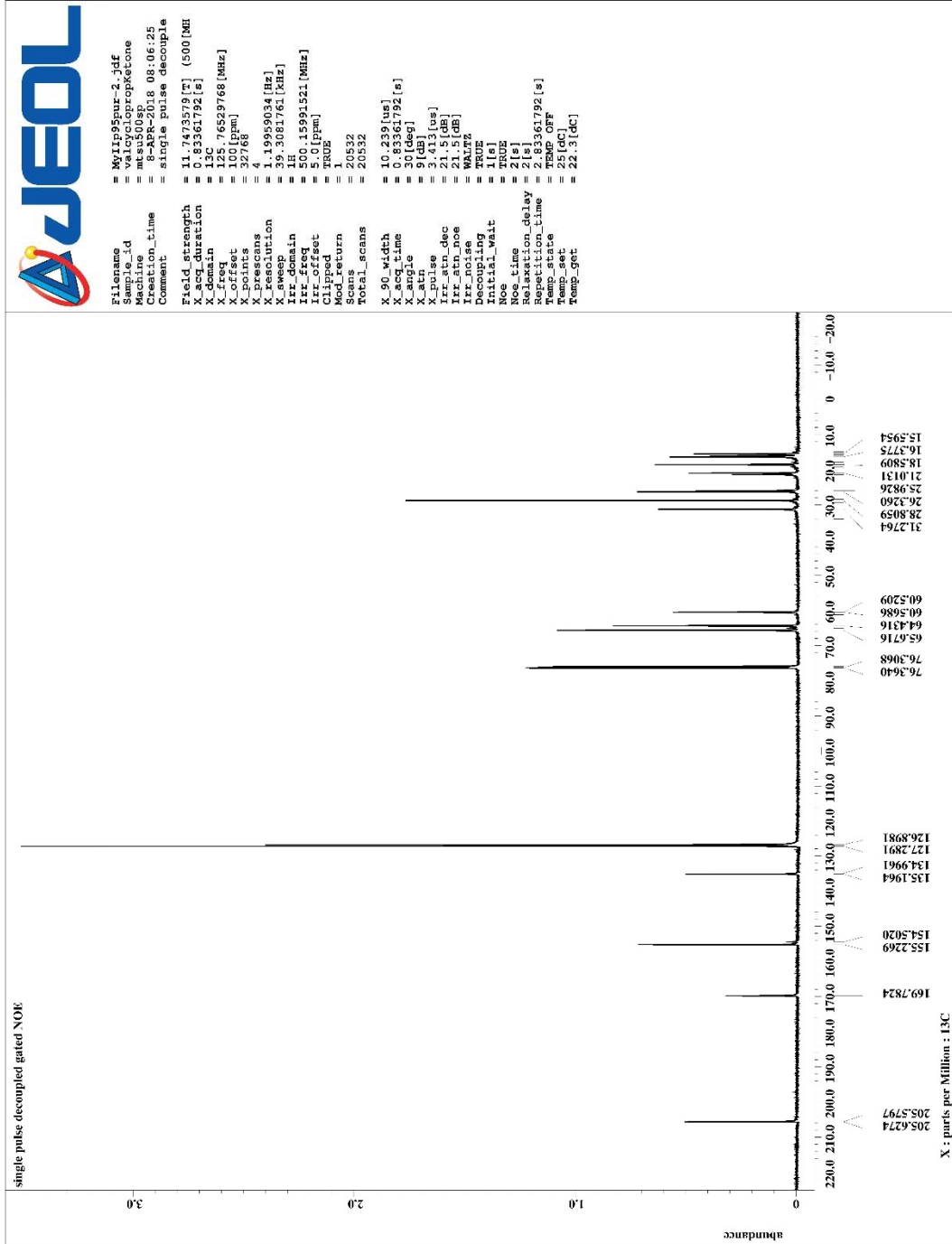
<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

HRMS







## Analysis Info

Analysis Name D:\Data\0302\88438-1.MYI\p65.d  
 Method 0118\_tune\_low\_pos\_NS.m  
 Sample Name 1.MYI\p65  
 Comment

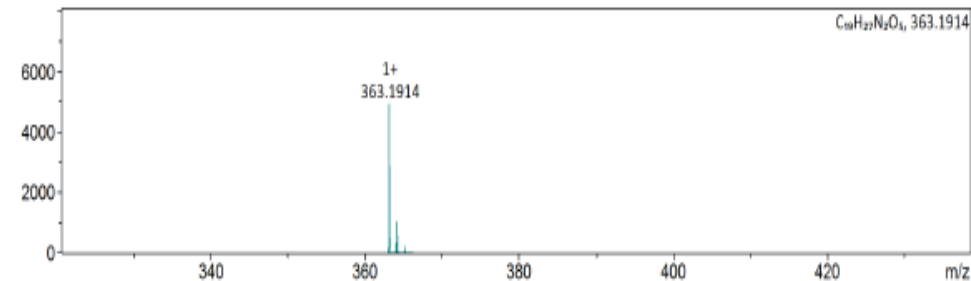
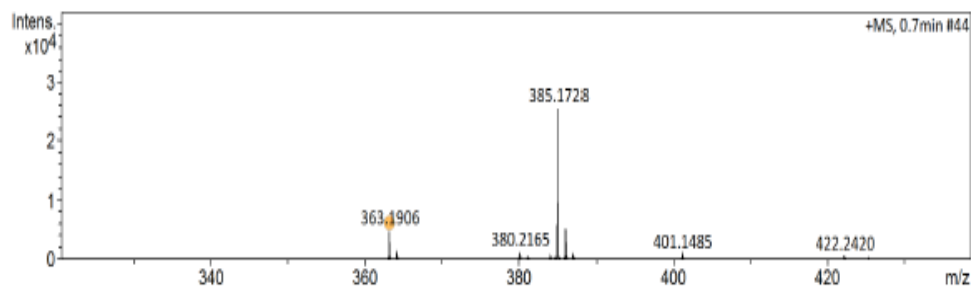
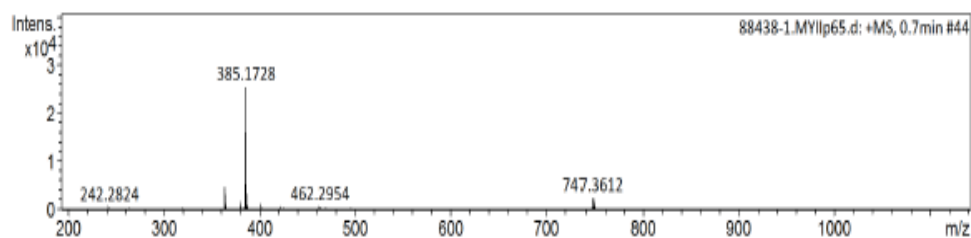
Acquisition Date 3/2/2018 3:11:50 PM

Operator BDAL@DE

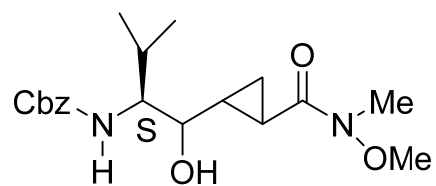
Instrument / Ser# micrOTOF II 8213750.1  
0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
363.190646	1	C <sub>10</sub> H <sub>27</sub> N <sub>2</sub> O <sub>5</sub>	363.191448	2.2	-1.3	7.5	ok	even



<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

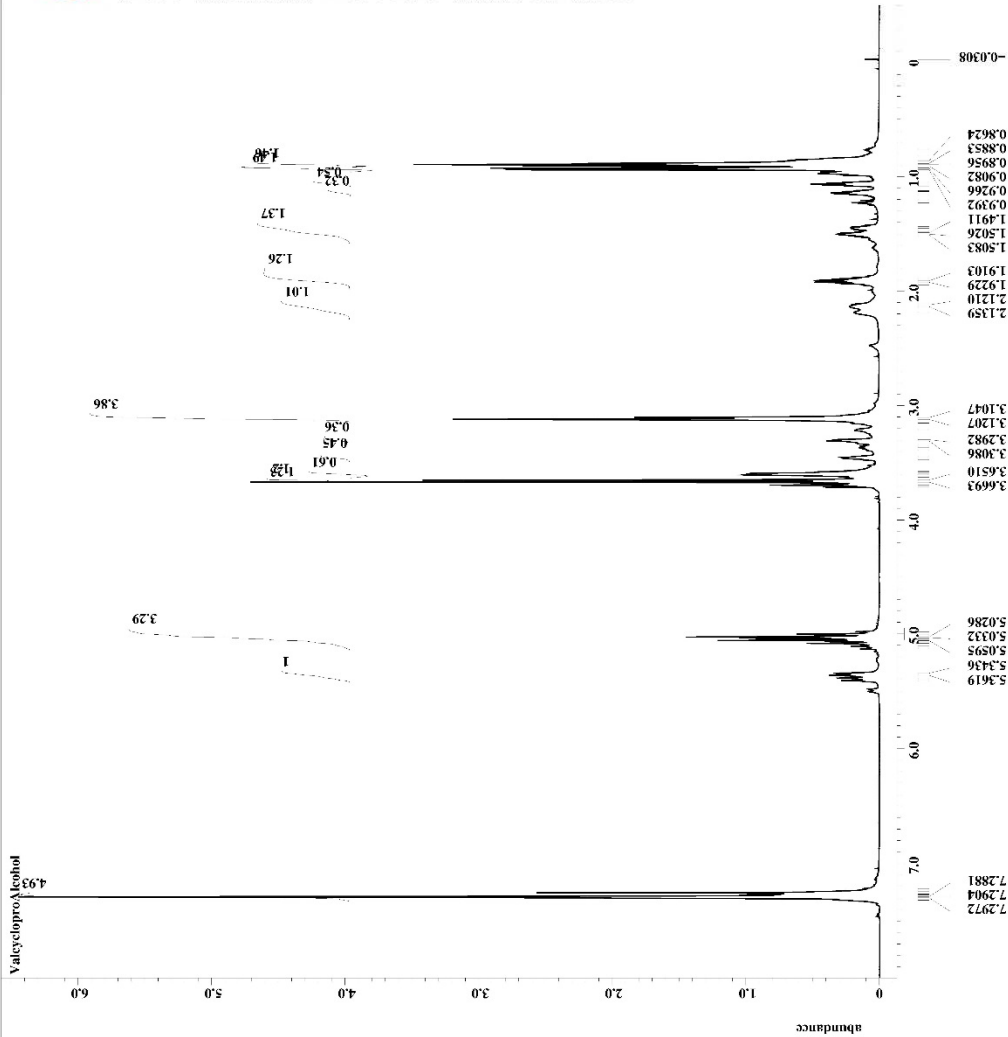
<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)



```

= MVLip97Cosy_PROTON-8.
= MVLip97Cosy
= mteu500sp
= 9-APR-2018 12:43:17
= ValicycloproAlcohol
Field strength = 11.7473579 [T] (500) [MH]
X_domain      = 1.74587904 [s]
X_acq_duration = 50.15991521 [MHz]
X_freq        = 16384
X_points      = 1
X_prescans    = 0.5727737 [Hz]
X_resolution  = 3.38438438 [KHz]
X_sweep       = 500.15991521 [MHz]
X_gain        = 5.0 [ppm]
X_ir_offset   = 1H
X_tri_domain  = 500.15991521 [MHz]
X_tri_freq    = 16384
X_tri_offset  = FALSE
X_clipper     = 1
X_mod_return  = 16
Total_scans   = 16
X_90_width   = 14.19 [us]
X_acq_time    = 1.74587904 [s]
X_angle       = 45 [deg]
X_atn         = 4 [dB]
X_pulse       = 7.099 [us]
X_mode        = Off
Dante_preset  = FALSE
Initial_wait  = 1 [s]
Relaxation_delay = 4 [s]
Repeat_time   = 1.74587904 [s]
Temp_state    = TEMP OFF
Temp_set      = 25 [cC]
Temp_get      = 22.4 [dC]

```

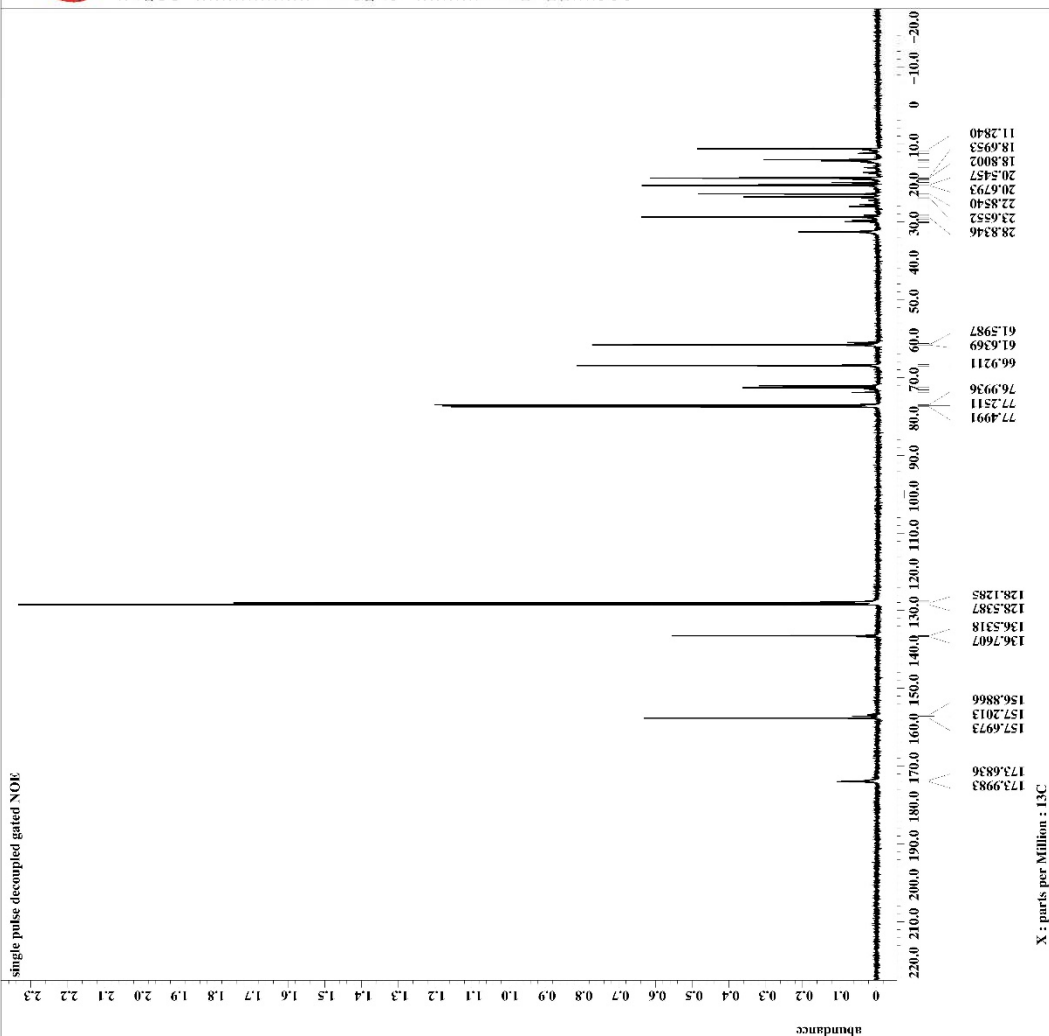


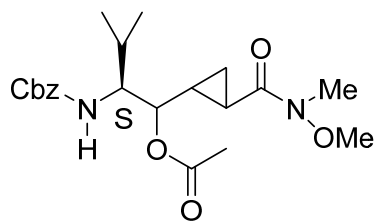
X : parts per Million : 1H



```

= MyIpr7castrn-2.jif
= valcyclopropylalcohol
= mtsu500sp
= 9-APR-2018 07:12:31
= single pulse decouple
= 11.7473579[T] (500[MH
= 0.83361792[s]
= 13C
= 100.6529768[MHz]
= 327.68
= 4
= 1.19959034[Hz]
= 39.3081761[KHz]
= 500.15991521[MHz]
= 5.0[ppm]
= TRUE
= TRUE
= 20196
= 20196
= 10.239[us]
= 0.83361792[s]
= 0.189g]
= 3.413[us]
= 21.5[dB]
= 21.5[dB]
= WALTZ
= 1[s]
= TRUE
= 2[s]
= 2[s]
= 33.61792[s]
= TRMP OFF
= 25[dc]
= 22.6[dc]
  
```



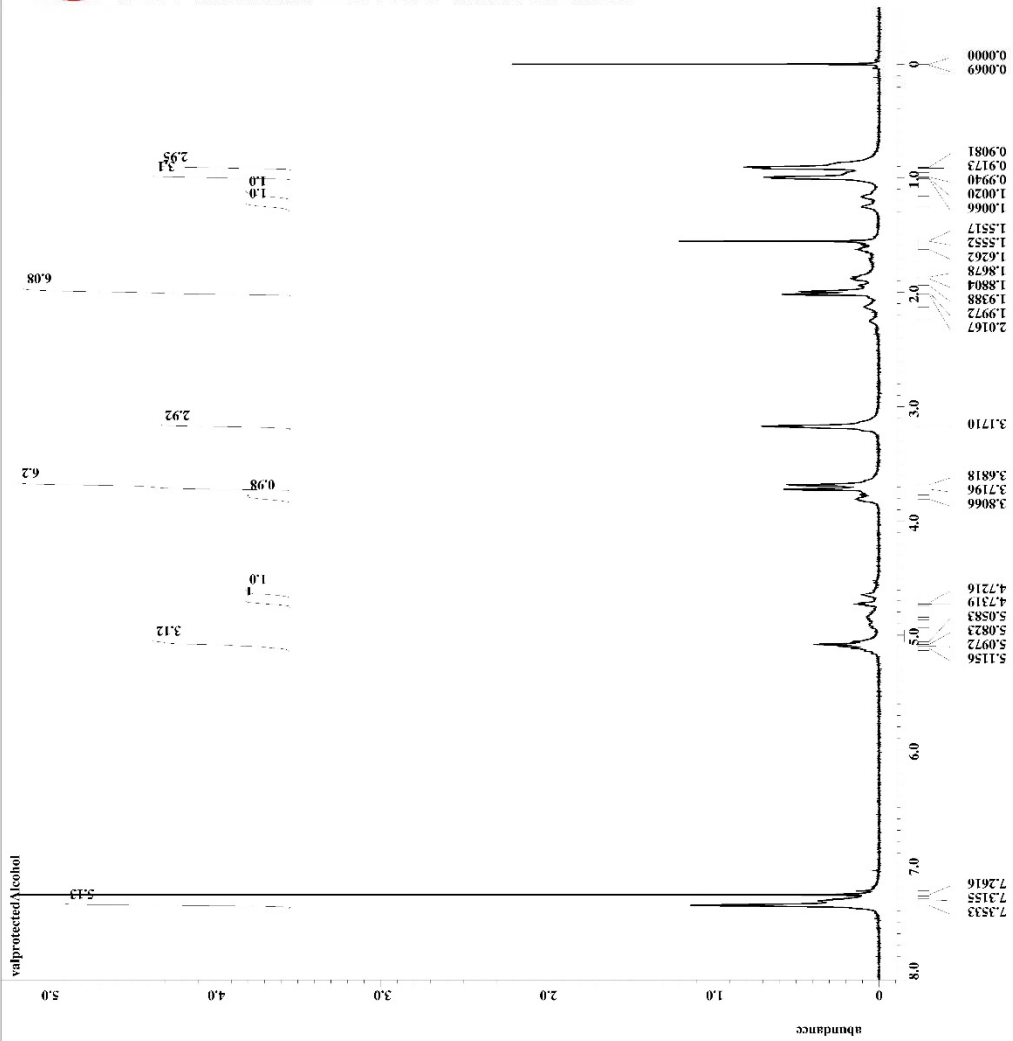


$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )



= MyIIP101pure\_PROTON-6  
 Sample\_id = MyIIP101pure  
 Machine = mteus500sp  
 Creation\_time = 14-APR-2018 13:18:22  
 Comment = valprotectedAlcohol  
 Field\_strength = 11.7473579 [T] (500 [MH  
 X\_acq\_duration = 1.74587904 [s]  
 X\_domain = 2H  
 X\_freq = 500.15991521 [MHz]  
 X\_f2 = 500.15991521 [MHz]  
 X\_points = 16384  
 X\_prescans = 1  
 X\_resolution = 0.5727737 [Hz]  
 X\_sweep = 9.38438438 [kHz]  
 X\_swept\_domain = 2H  
 Irr\_freq = 500.15991521 [MHz]  
 Irr\_offset = 5.0 [ppm]  
 Tri\_domain = 1H  
 Tri\_freq = 500.15991521 [MHz]  
 Tri\_offset = 5.0 [ppm]  
 Clipped = FALSE  
 Mod\_return = 1  
 Scans = 16  
 Total\_scans = 16  
 X\_90\_width = 14.19 [us]  
 X\_acq\_time = 1.74587904 [s]  
 X\_angle = 45 [deg]  
 X\_atn = 4 [dB]  
 X\_pulse = 7.095 [us]  
 Tri\_mode = OFF  
 Tri\_mode = OFF  
 Dante\_preset = FALSE  
 Initial\_wait = 1 [s]  
 Relaxation\_delay = 4 [s]  
 Acquisition\_time = 1.74587904 [s]  
 Temp\_set = 25 [dC]  
 Temp\_get = 22.5 [dC]



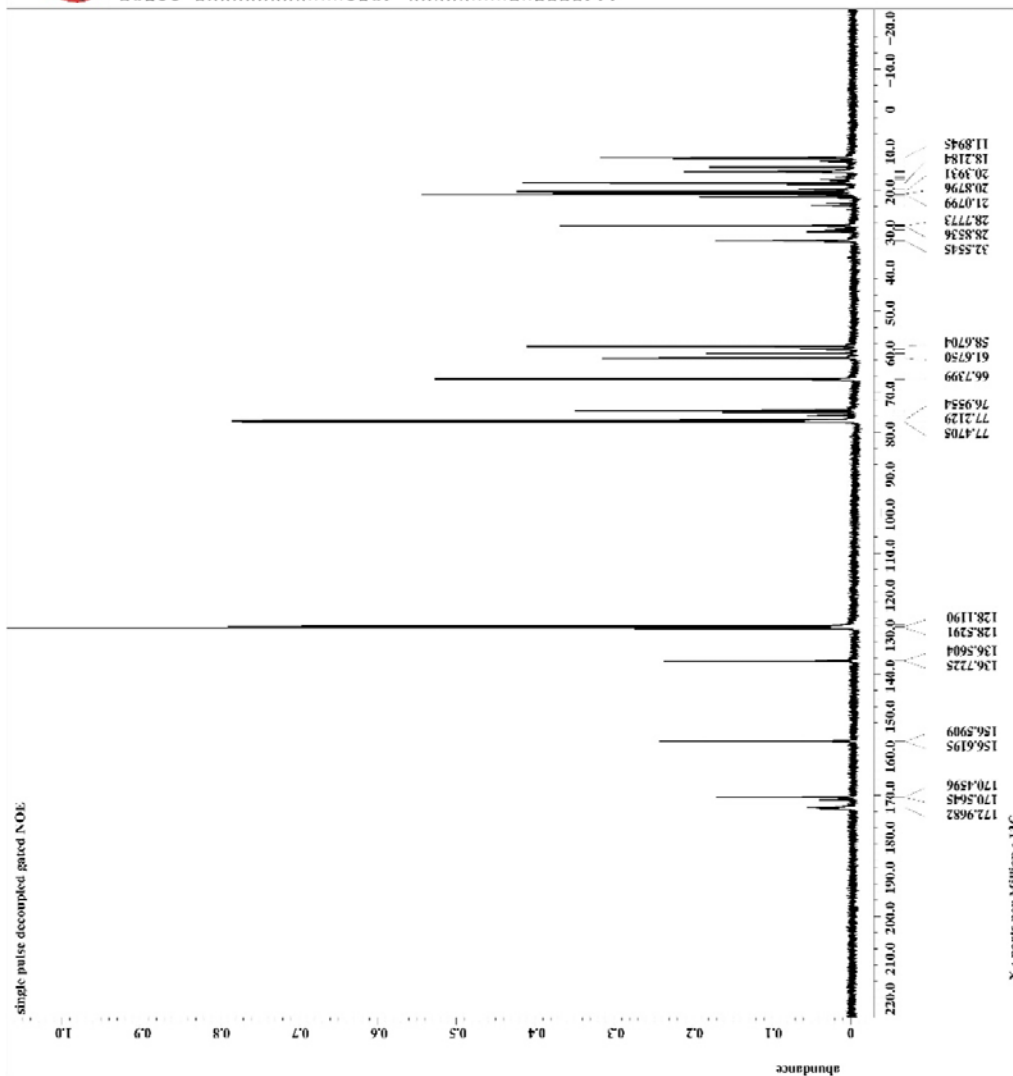
X : parts per Million : 1H



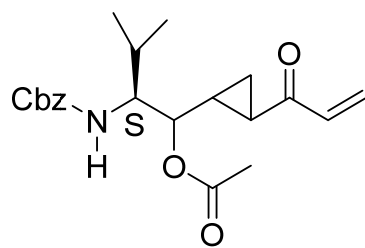
single pulse decoupled gated NOE

```

Filename = MyIpr1Carbon-4.jdr
Sample_id = Valproalcohol
AcqDate = 18-JAN-2018 07:49:39
Creation_time = single pulse decouple
Comment
Field_strength = 11.747159 [T] (500 [MHz])
X_domain = 13C
X_freq = 125.7652768 [MHz]
X_offset = 100 [ppm]
X_points = 4
X_pulse = 1.19959034 [Hz]
X_resolution = 39.3081761 [kHz]
X_sweep = 10
X_domain = 50.15991521 [MHz]
X_offset = TRUE
Clipped = 1
Mod_return = 1
Scans = 17563
Total_scans = 17563
X_90_width = 10.239 [us]
X_acc_time = 0.83361792 [s]
X_angle = 9 [deg]
X_atn = 3.415 [us]
X_pulse = 3.415 [us]
Irr_atn_dec = 21.5 [dB]
Irr_atn_noe = 21.5 [dB]
Decoupling = TRUE
Initial_west = 1 [s]
Noe = TRUE
Noe_time = 2 [s]
Noe_time_delay = 2.83361792 [s]
Repetition_time = 2.83361792 [s]
Temp_state = TEMPO OFF
Temp_sec = 25 [dc]
Temp_get = 22.8 [dc]
    
```

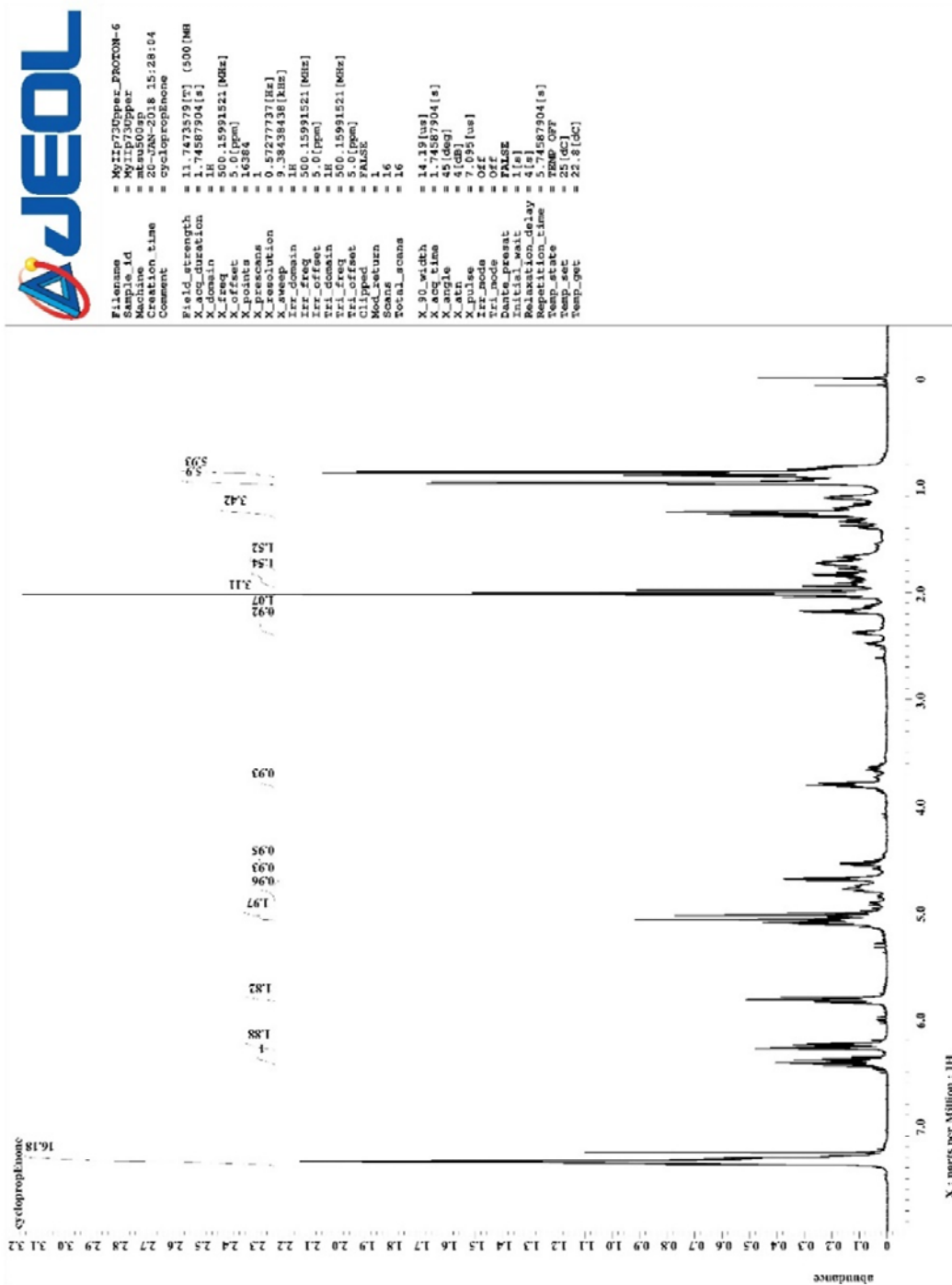






$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )



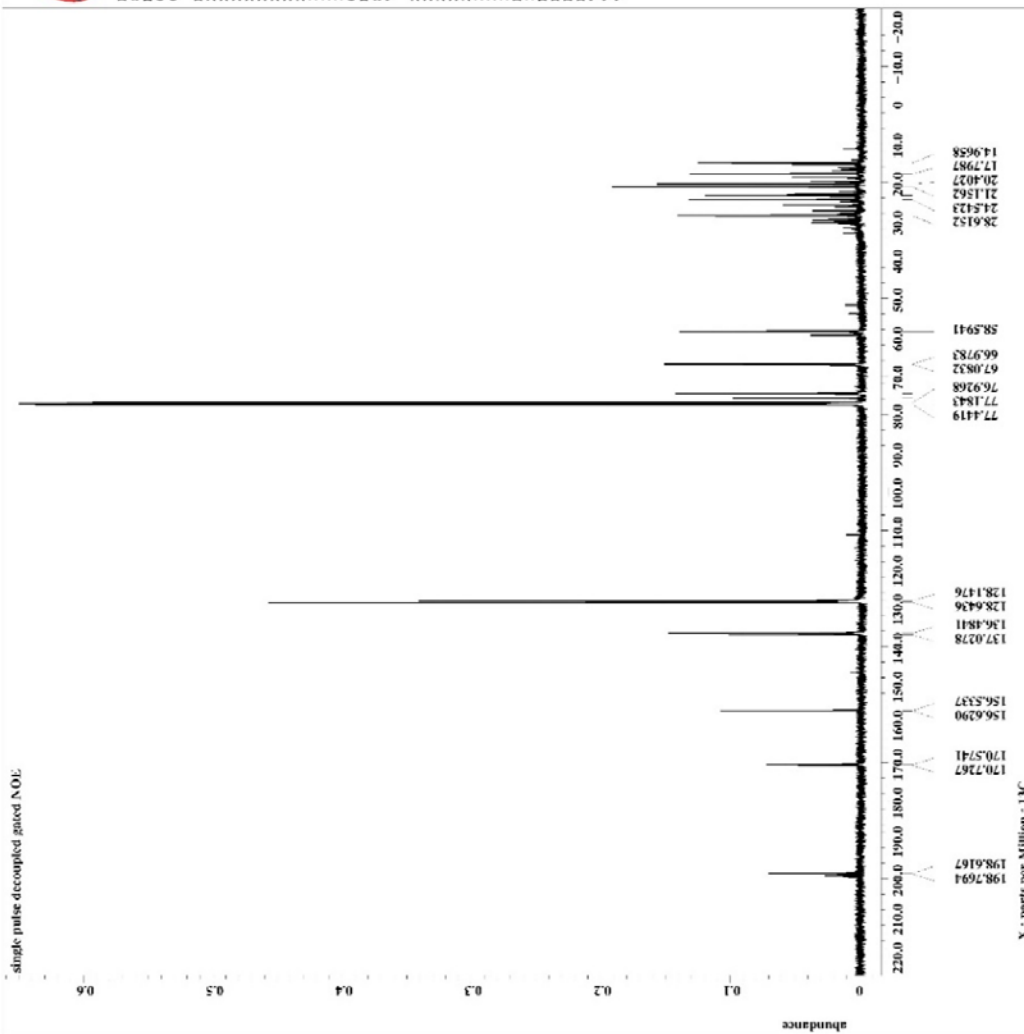


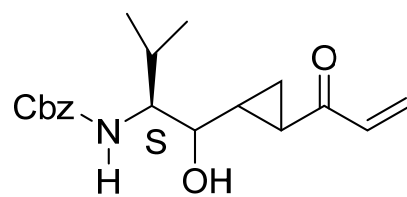
```

= MyZipUpperCarb-6.jd
= LastName
= Sample_id
= Machine
= Date_08_07_22_43
= Comment
= single pulse decouple

Field_strength = 11.7435792[T] (500[MH]
X_acq_duration = 1.83361792[s]
X_freq = 125.76529768[MHz]
X_offset = 100[ppm]
X_points = 32768
X_prescans = 1
X_resolution = 1.39949034[Hz]
X_sweep = 39.3081761[kHz]
Irr_freq = 500.15991521[MHz]
C11_offset = 70[ppm]
Mod_return = 1
Total_scans = 17521

X_90_width = 10.239[us]
X_acq_time = 0.83361792[s]
X_angle = 30[deg]
X_atn = 9[db]
Irr_atn_us = 21.5[us]
Irr_atn_db = 21.5[db]
Irr_atn_pwr = 21.5[db]
Decoupling = WALTZ
Mita_west = TRUE
Noe_time = 2[s]
Relaxation_delay = 2[s]
Repetition_time = 2.83361792[s]
Temp_set = 25[degC]
Temp_get = 25[degC]
Temp_set = 22.9[degC]
    
```





<sup>1</sup>H-NMR (500 MHz, CDCl<sub>3</sub>)

<sup>13</sup>C-NMR (125 MHz, CDCl<sub>3</sub>)

HRMS



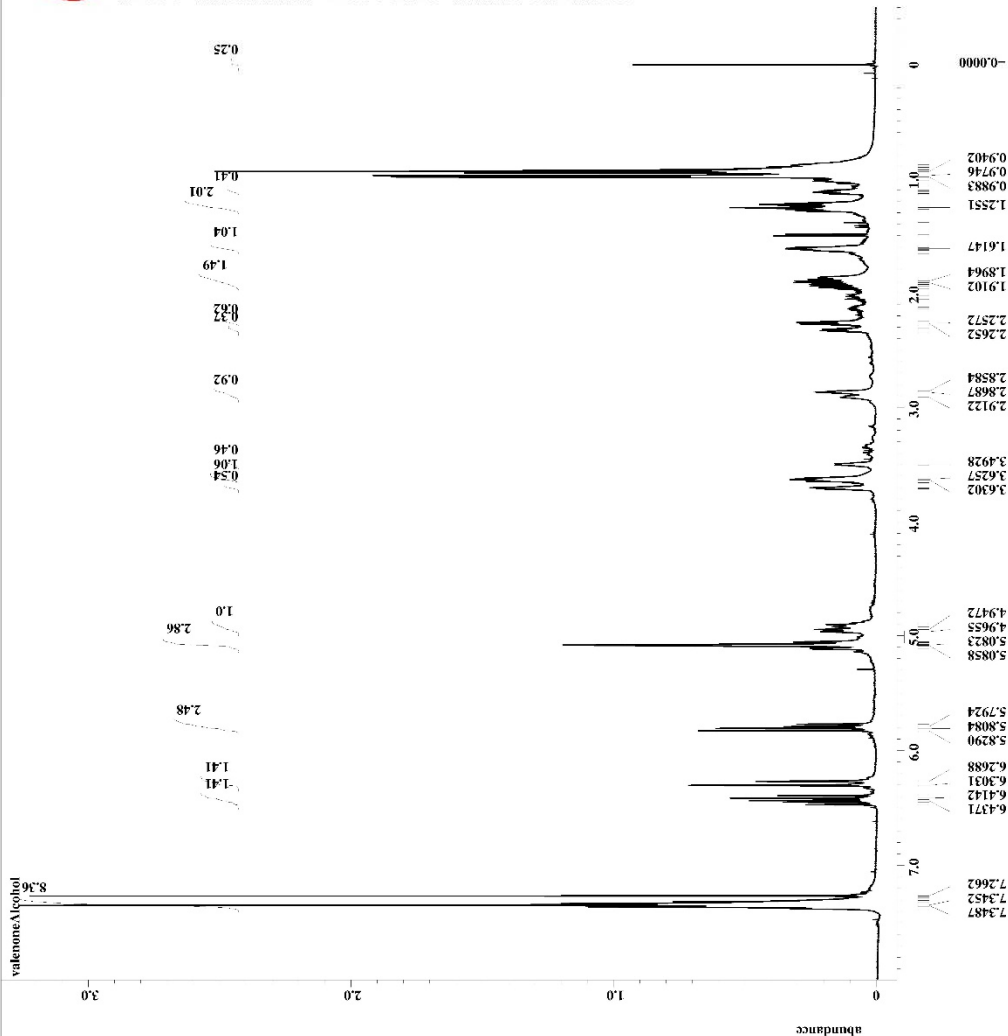
```

File Name      = MyJIP75LwrsPt_PROTON-
Sample_id     = MyJIP75LwrsPt
Machine       = mtsu500sp
Creation_time = 11-APR-2018 16:52:14
Comment      = valenoneAlcohol

Field strength = 11.7473579 [T] (500 [MH
X_acq_duration = 1.74587904 [s]
X_domain       = 1H
X_freq         = 500.15991521 [Mhz]
X_f2          = 500.15991521 [Mhz]
X_points      = 16384
X_prescans    = 1
X_resolution  = 0.5727737 [Hz]
X_sweep       = 9.38438438 [KHz]
X_sweep_time  = 0.185185185 [s]
X_t1          = 5.00 [ppm]
Irr_freq      = 500.15991521 [Mhz]
Irr_offset    = 5.0 [ppm]
Tri_domain    = 1H
Tri_freq      = 500.15991521 [Mhz]
Tri_offset    = 5.0 [ppm]
Clipped      = FALSE
Mod_return    = 1
Total_scans   = 16

X_90_width   = 14.19 [us]
X_acq_time   = 1.74587904 [s]
X_angle      = 45 [deg]
X_atn        = 4 [dB]
X_pulse      = 7.095 [us]
Tri_mode     = OFF
Dante_presat = FALSE
Initial_wait = 1 [s]
Relaxation_delay = 4 [s]
Acq_start_time = 11-APR-2018 16:52:14
Temp_set     = 25 [dC]
Temp_get     = 22.5 [dC]

```



X : parts per Million : 1H

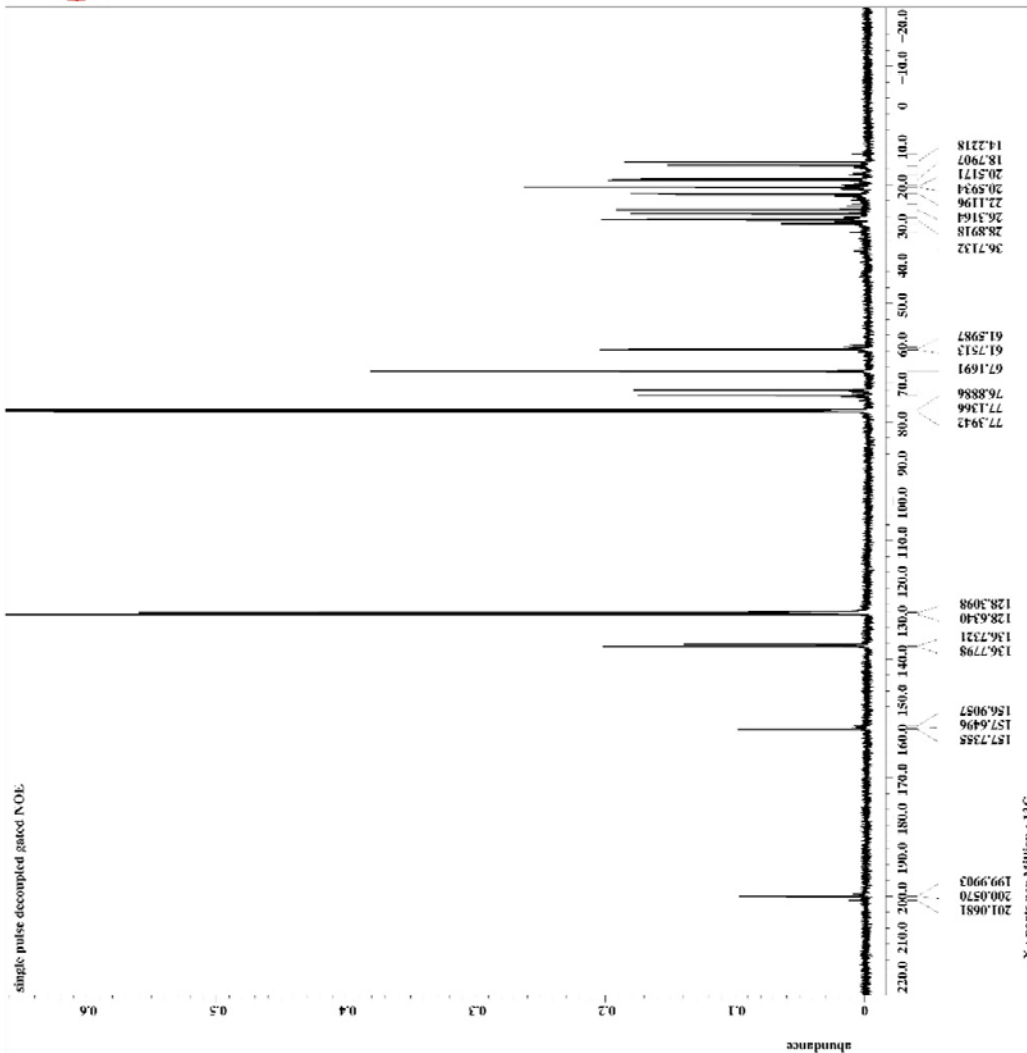


```

=====
Filename = MyIP7LowCaron-2.jd
Sample_id = cyclopropenone
Machine = mnu500p
Acquisition_time = 8 07:06:22
Date_UTC = 20110708
Comment = single pulse decouple

Field_strength = 11.7473579[T] (500[MH]
X_acq_duration = 0.83361792[s]
X_resolution = 125.76529768 [MHz]
X_freq = 100[ppm]
X_offset = 32768
X_points = 32768
X_prescans = 1
X_swept = 1.19959034[Hz]
X_resolution = 59.3081761[kHz]
X_sweep = 1H
IRF_domain = 500.15991521 [MHz]
IRF_freq = 5.0[ppm]
IRF_offset = 1.000
Mod_return = 1
Total_scans = 19057

X_90_width = 10.239[us]
X_acq_time = 0.83361792[s]
X_angle = 30[deg]
X_atn = 8[dB]
X_pulse_dec = 21[us]
IRF_atn_pos = 21.5[dB]
IRF_atn_neg = 21.5[dB]
IRF_noise = MALTZ
Decoupling = TRUE
Initial_wait = 700[us]
Nox_time = 2[s]
Relaxation_delay = 2[s]
Repetition_time = 2.83361792[s]
Temp_state = 25[degC]
Temp_offset = 25[degC]
Temp_set = 22.7[degC]
=====
    
```



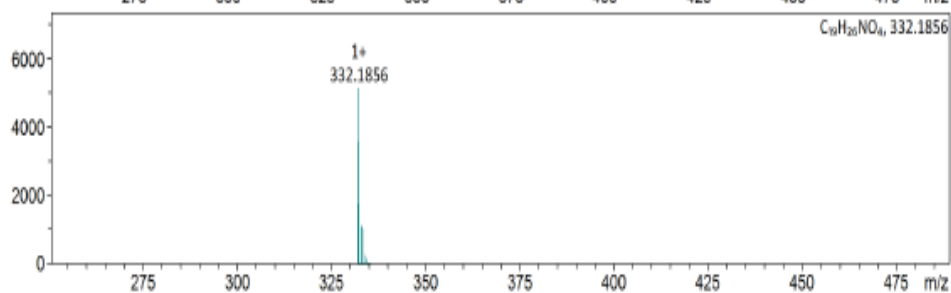
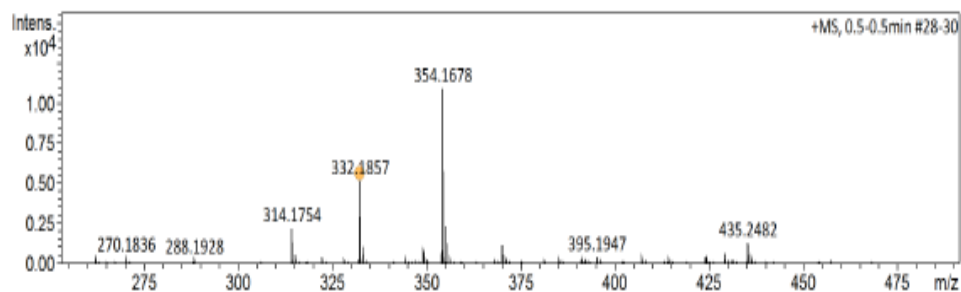
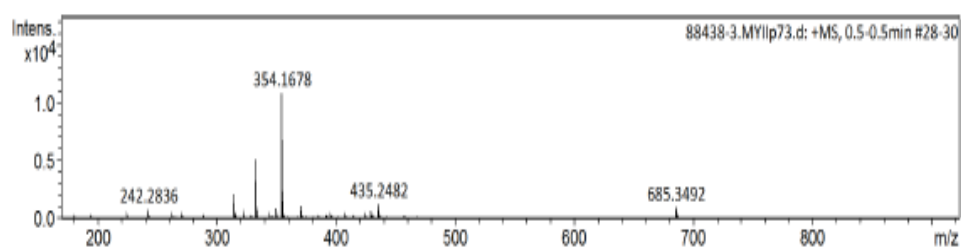
## Analysis Info

Analysis Name D:\Data\0302\88438-3.MYIIP73.d  
 Method 0118\_tune\_low\_pos\_NS.m  
 Sample Name 3.MYIIP73  
 Comment

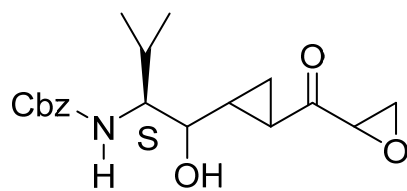
Acquisition Date 3/2/2018 3:19:47 PM  
 Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdB	N-Rule	e <sup>-</sup> Conf
332.185672	1	C <sub>19</sub> H <sub>28</sub> NO <sub>4</sub>	332.185635	-0.1	-2.3	7.5	ok	even

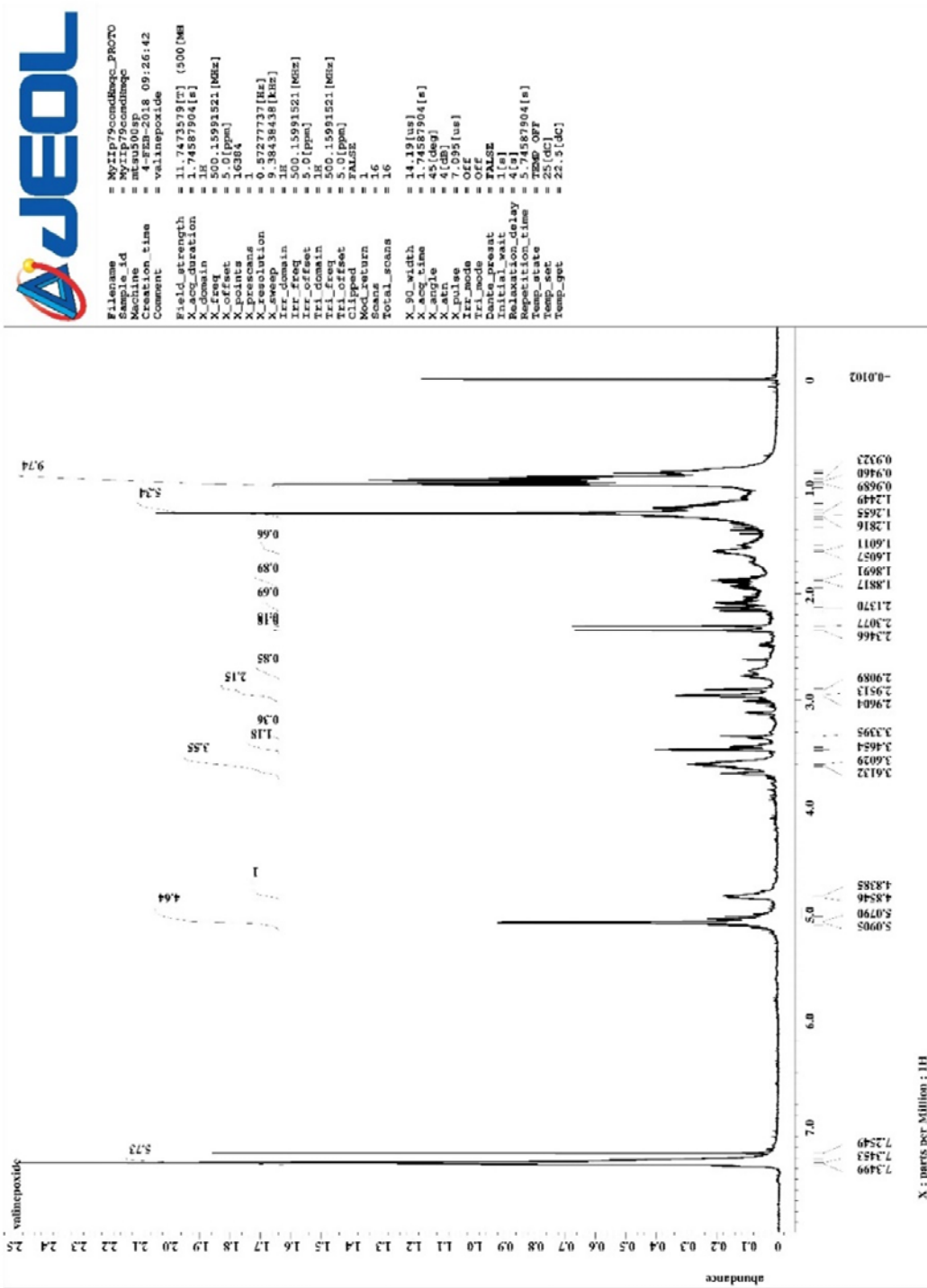


$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

HRMS





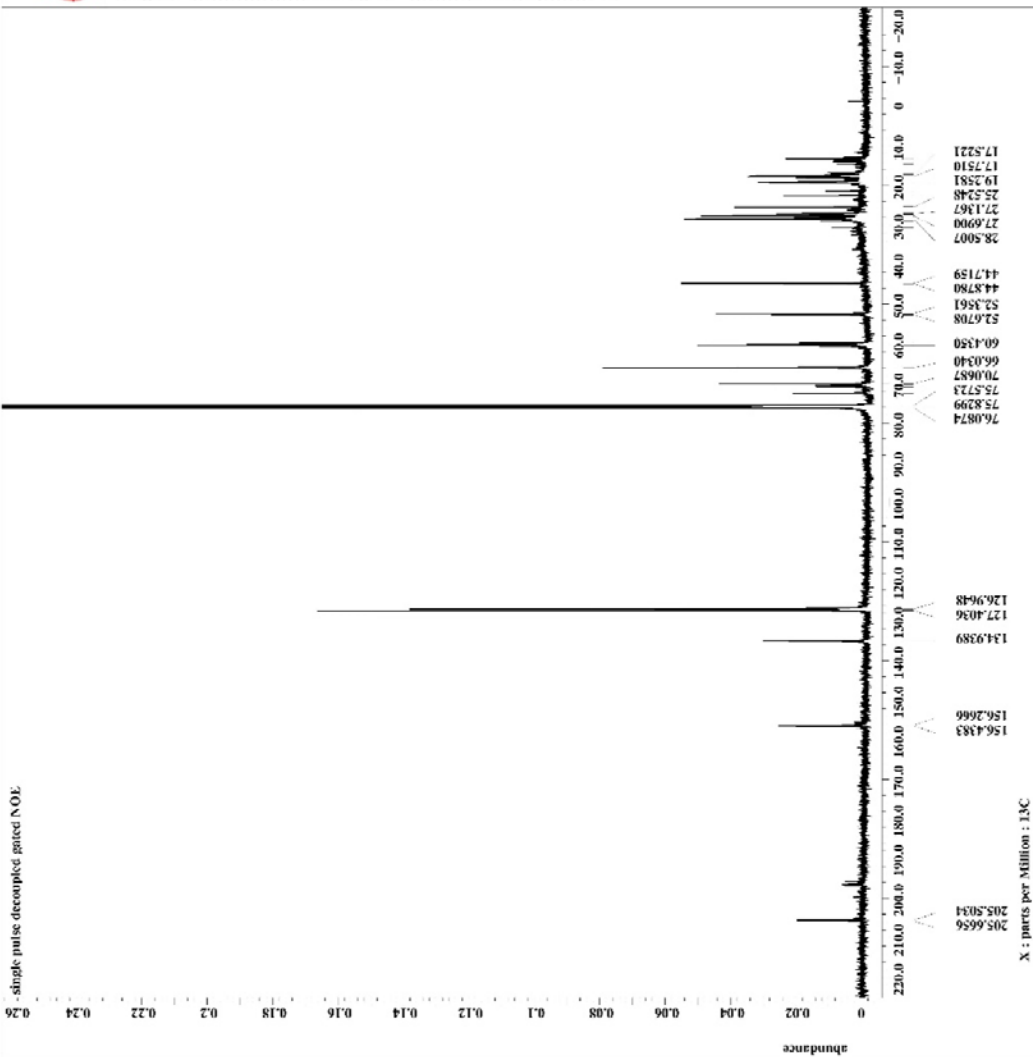
X : parts per Million : 1H



```

File Name = My1p7cmdbur-5.jdf
Sample ID = Zinalvalinepoxide
Machine = mnu500sp
Creation Time = 4-Feb-2018 00:15:41
Comment = single pulse decouple

Field_strength = 11.7473579 [T] (500 [MH])
X_acq_duration = 0.83361792 [s]
X_domain = 120
X_offset = 76529768 [MHz]
X_points = 32768
X_prescans = 4
X_resolution = 1.9950934 [Hz]
X_resolution_h = 39.3681761 [kHz]
Irr_domain = 18
Irr_freq = 500.15991521 [MHz]
Irr_offset = 5.0 [ppm]
Irr_phase = TRUE
Mag_return = 40000
Total_scans = 40000
X_90_width = 10.238 [us]
X_acq_time = 0.83361792 [s]
X_angle = 30 [deg]
X_atn = 9 [dB]
X_pulse = 3.41 [us]
Irr_atn_dec = 21.5 [dB]
Irr_atn_noise = 21.5 [dB]
Irr_noise = WALTZ
Decoupling = TRUE
Initial_wait = 1 [s]
New_time = 2 [s]
Relaxation_delay = 2 [s]
Relaxation_time = 2.83361792 [s]
Temp_state = 75deg OFF
Temp_set = 22.8 [degC]
Temp_spt = 22.8 [degC]
    
```



## Analysis Info

Analysis Name D:\Data\0302\88438-2.MYI\p79.d  
 Method 0118\_tune\_low\_pos\_NS.m  
 Sample Name 2.MYI\p79  
 Comment

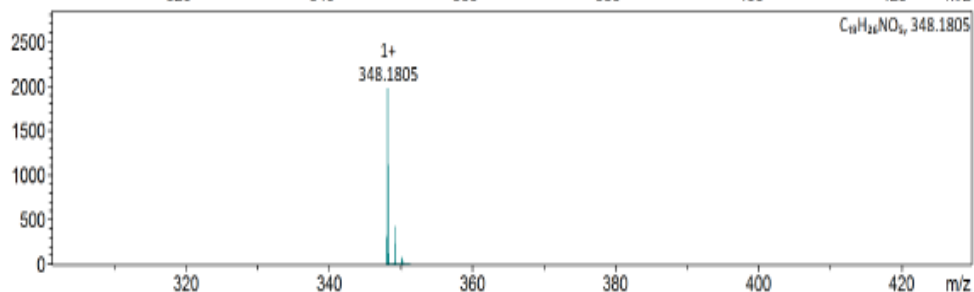
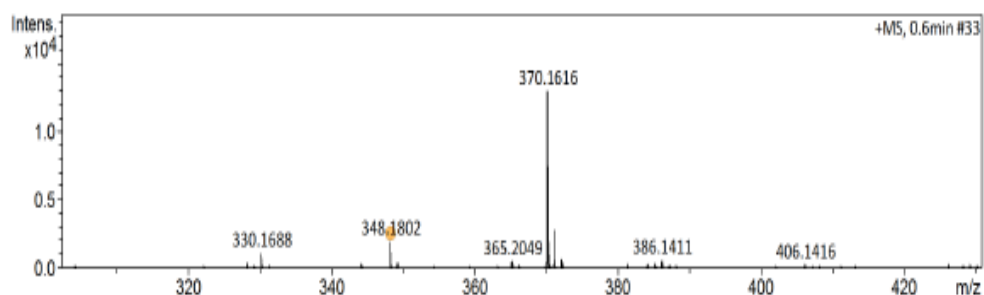
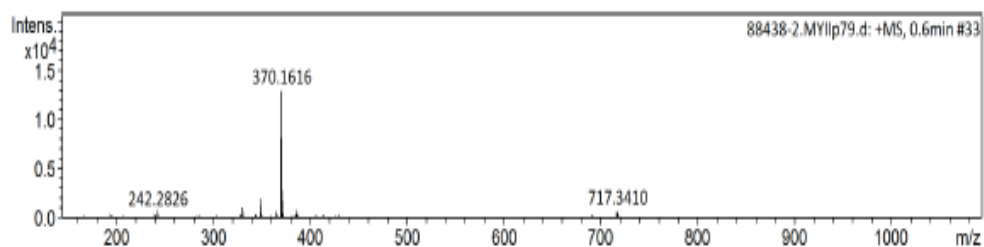
Acquisition Date 3/2/2018 3:16:42 PM

Operator BDAL@DE

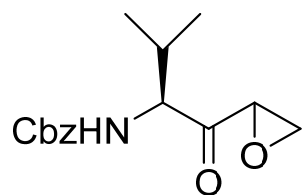
Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
348.180233	1	C <sub>19</sub> H <sub>28</sub> NO <sub>5</sub>	348.180549	0.9	2.0	7.5	ok	even



$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

HRMS

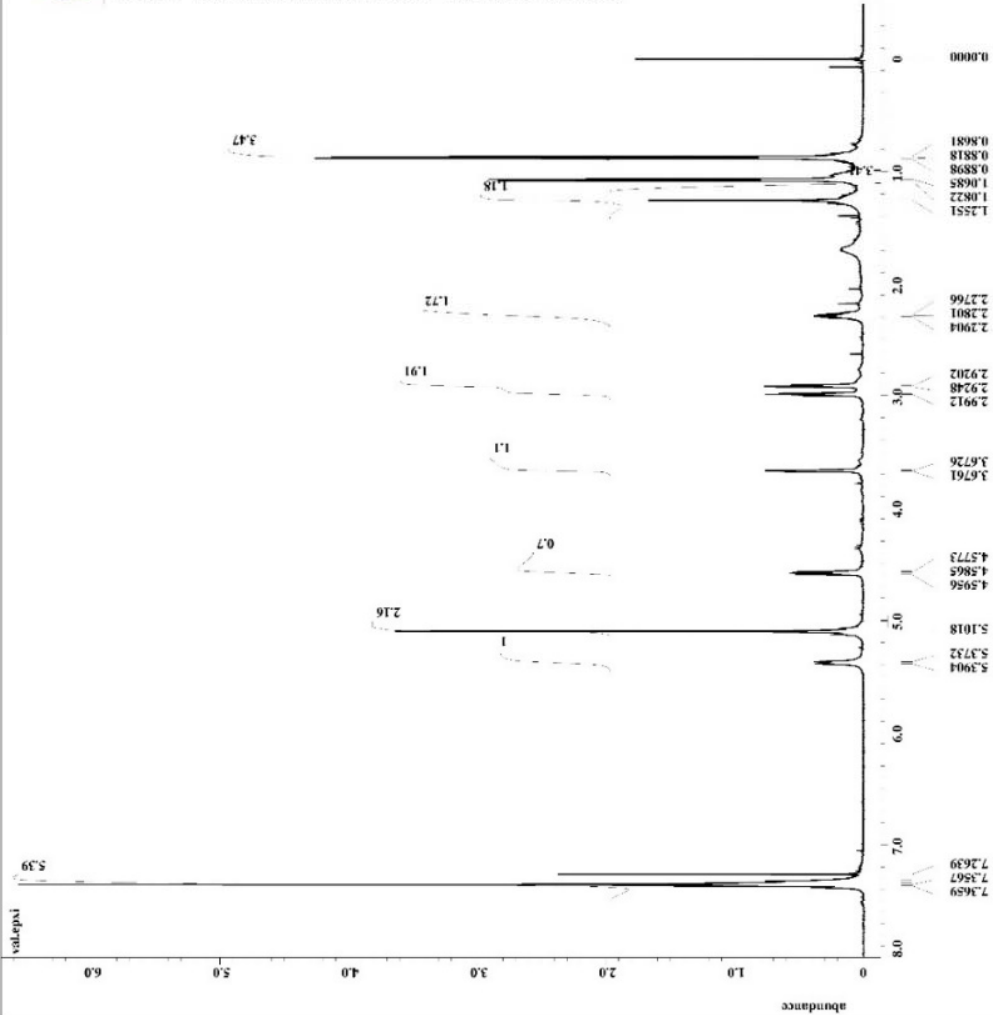


```

Filename = Myr149entB_PROTON-7.3
Sample_id = Myr149entB
Machine = mnu500sp
Acquisition_time = 2017.10.29.34
Comment = val.opxl

Field_strength = 11.7473579 [T] (500 [MHz])
X_acq_duration = 1.74587904 [s]
X_sweep = 1.74587904 [s]
X_freq = 500.15991521 [MHz]
X_offset = 5.0 [ppm]
X_points = 16384
X_prescans = 0
X_resolution = 0.5227737 [Hz]
X_sweep = 9.38438438 [MHz]
Irr_domain = 1H
Irr_freq = 500.15991521 [MHz]
Irr_offset = 5.0 [ppm]
Irr_domain = 1H
Irr_freq = 500.15991521 [MHz]
Irr_offset = 5.0 [ppm]
Mod_return = FALSE
Total_scans = 16

X_90_width = 14.19 [us]
X_acq_time = 1.74587904 [s]
X_sweep = 1.74587904 [s]
X_gain = 4 [dB]
X_pulprg = 7.095 [us]
Irr_mode = Off
Irr_pulprg = 7.095 [us]
Initial_wait = 1 [s]
Relaxation_delay = 4 [s]
Repetition_time = 5.74587904 [s]
Temp_state = Off
Temp_set = 23 [C]
Temp_get = 20.5 [C]
    
```



X : parts per Million : 1H



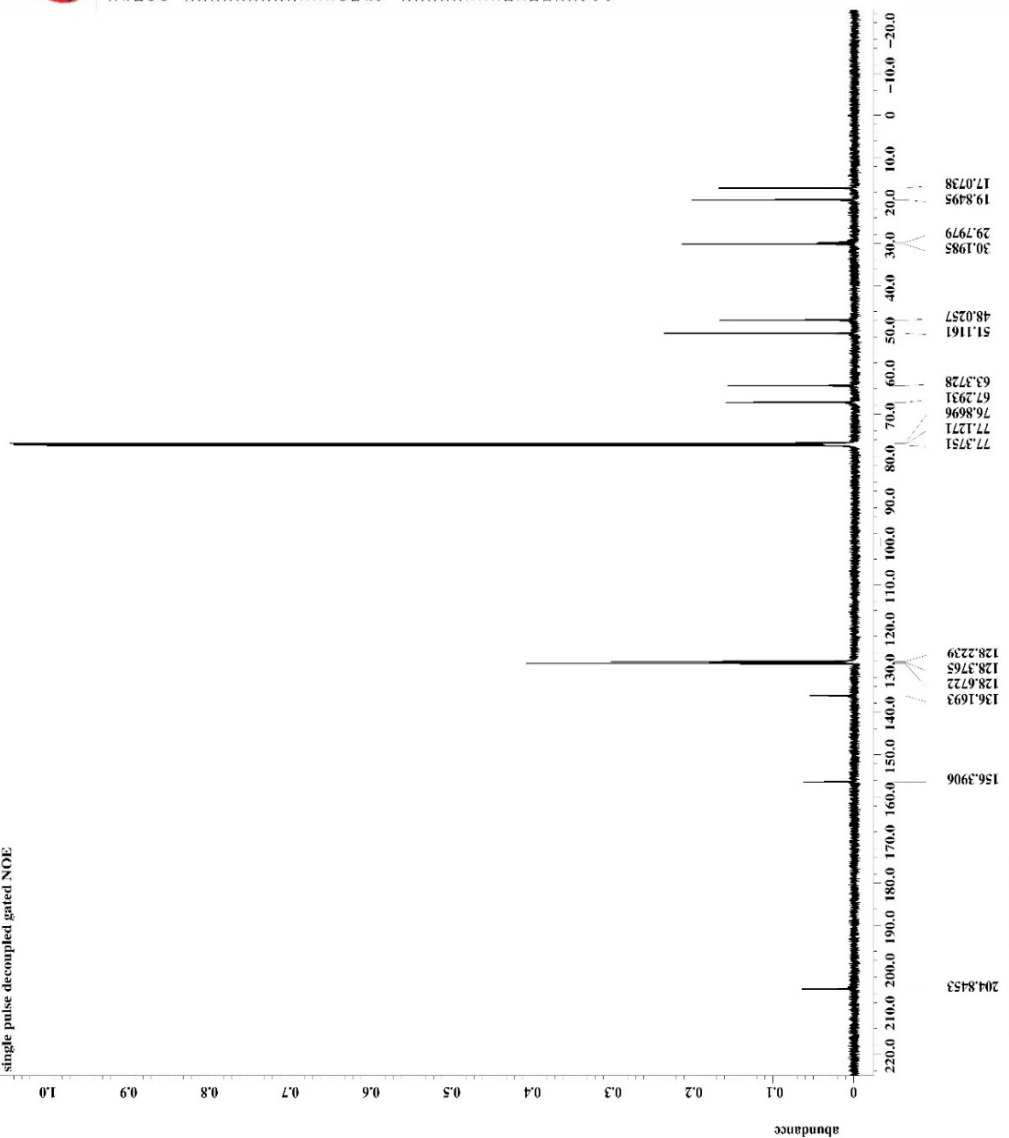
single pulse decoupled gated NOE

```

= MYP149spotBcarbon-2.j
Sample_id = val.enone
Machine = ncsu505sp
Date_7 = 10.00.44
Time = 10.00.44
Comment = single pulse decouple

Field_strength = 11.7473579[T] (500 [MH
X_acq_duration = 0.83361792 [s]
X_chain =
X_freq = 125.76529768 [MHz]
X_offset = 100 [ppm]
X_points = 32768
X_prescans = 4
X_resolution = 1.9959034 [Hz]
X_sweep = 39.3081761 [kHz]
Irr_domain = 1H
Irr_freq = 500.15991521 [MHz]
Irr_offset = 5.0 [ppm]
Irr_phase = TRUE
Mod_return = 1
Scans = 17071
Total_scans = 17071
X_90_width = 10.239 [us]
X_acq_time = 0.83361792 [s]
X_angle = 30 [deg]
X_atn = 9 [dB]
X_pulse_dec = 2.413 [us]
X_atn_noe = 21.5 [dB]
Irr_noise = WALTZ
Decoupling = TRUE
Initial_wait = 1 [s]
Noe_time = 2 [s]
Relaxation_delay = 2 [s]
Repetition_time = 2.83361792 [s]
Temp_state = TRNG OFF
Temp_set = 25 [C]
Temp_get = 21.2 [dc]

```



X : parts per Million : 13C

## Analysis Info

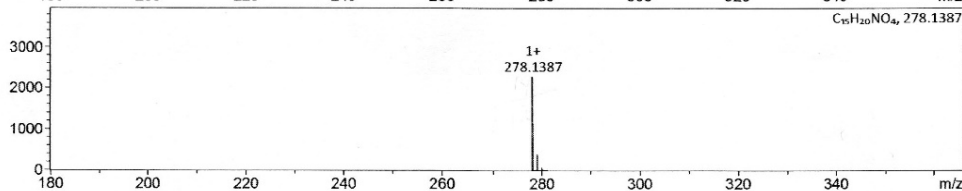
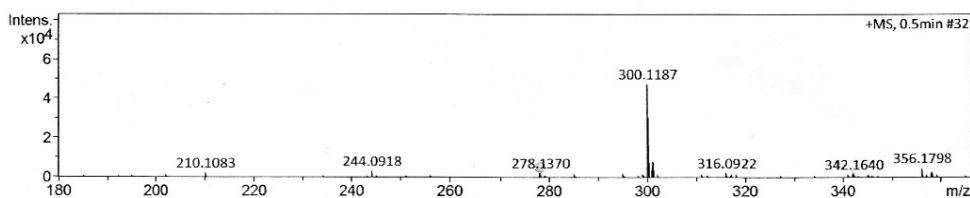
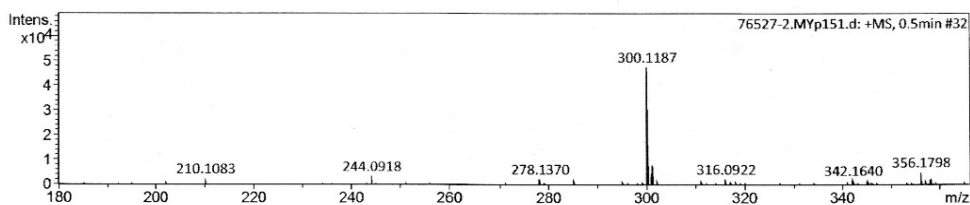
Analysis Name D:\Data\0726\76527-2.MYp151.d  
 Method 061517\_tune\_low\_pos\_NS.m  
 Sample Name 2.MYp151  
 Comment direct infusion

Acquisition Date 7/26/2017 2:41:45 PM

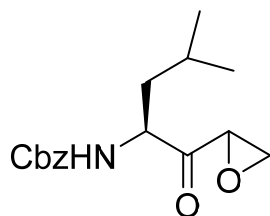
Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
278.137027	1	C <sub>15</sub> H <sub>20</sub> NO <sub>4</sub>	278.138685	6.0	1495.6	6.5	ok	even



$^1\text{H-NMR}$  (500 MHz,  $\text{CDCl}_3$ )

$^{13}\text{C-NMR}$  (125 MHz,  $\text{CDCl}_3$ )

HRMS





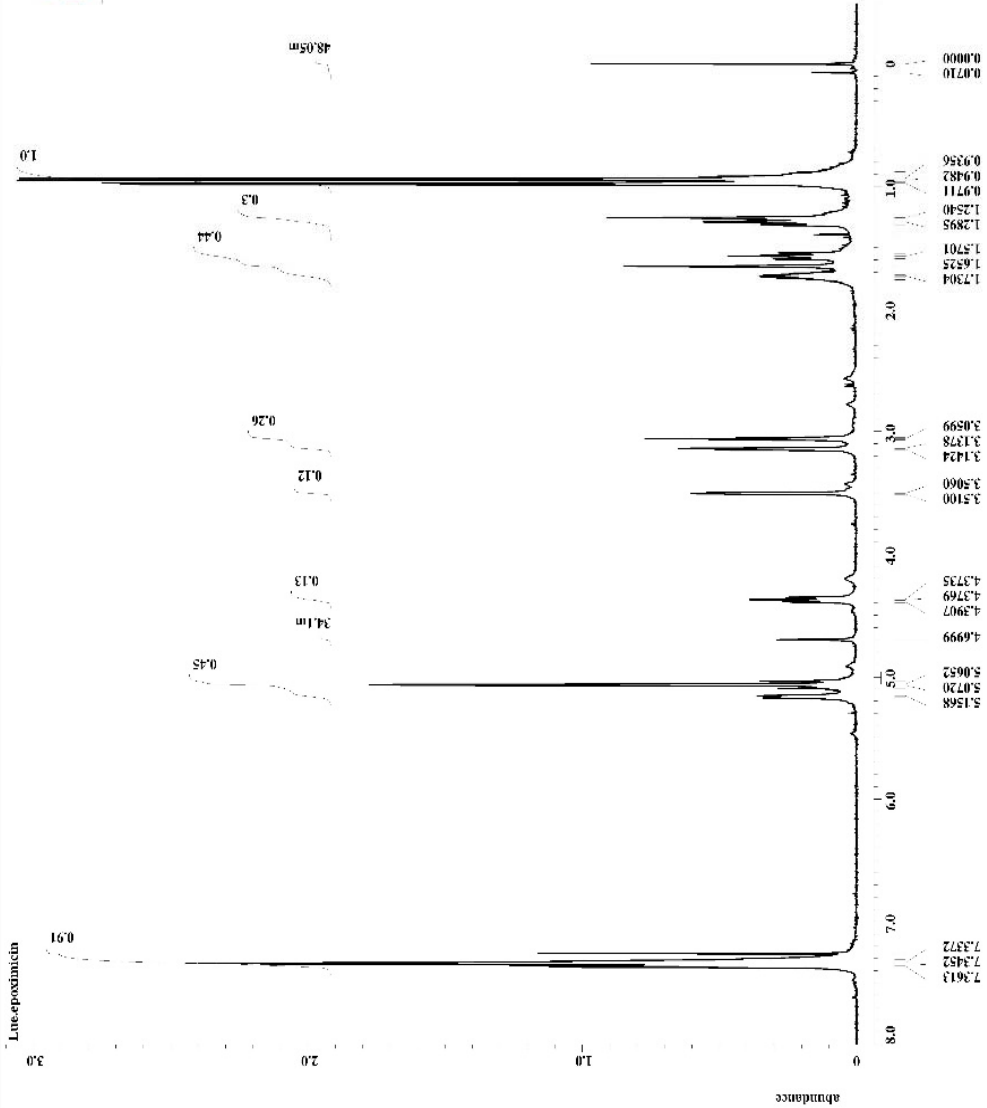
```

Filename = MY2pSepta_PROTON-7_3d
Sample_id = MY2pSepta
Machine = JEOL JNM-PS100
Collection_time = 19 JUN 2017 17:19:30
Comment = Late epoxy resin

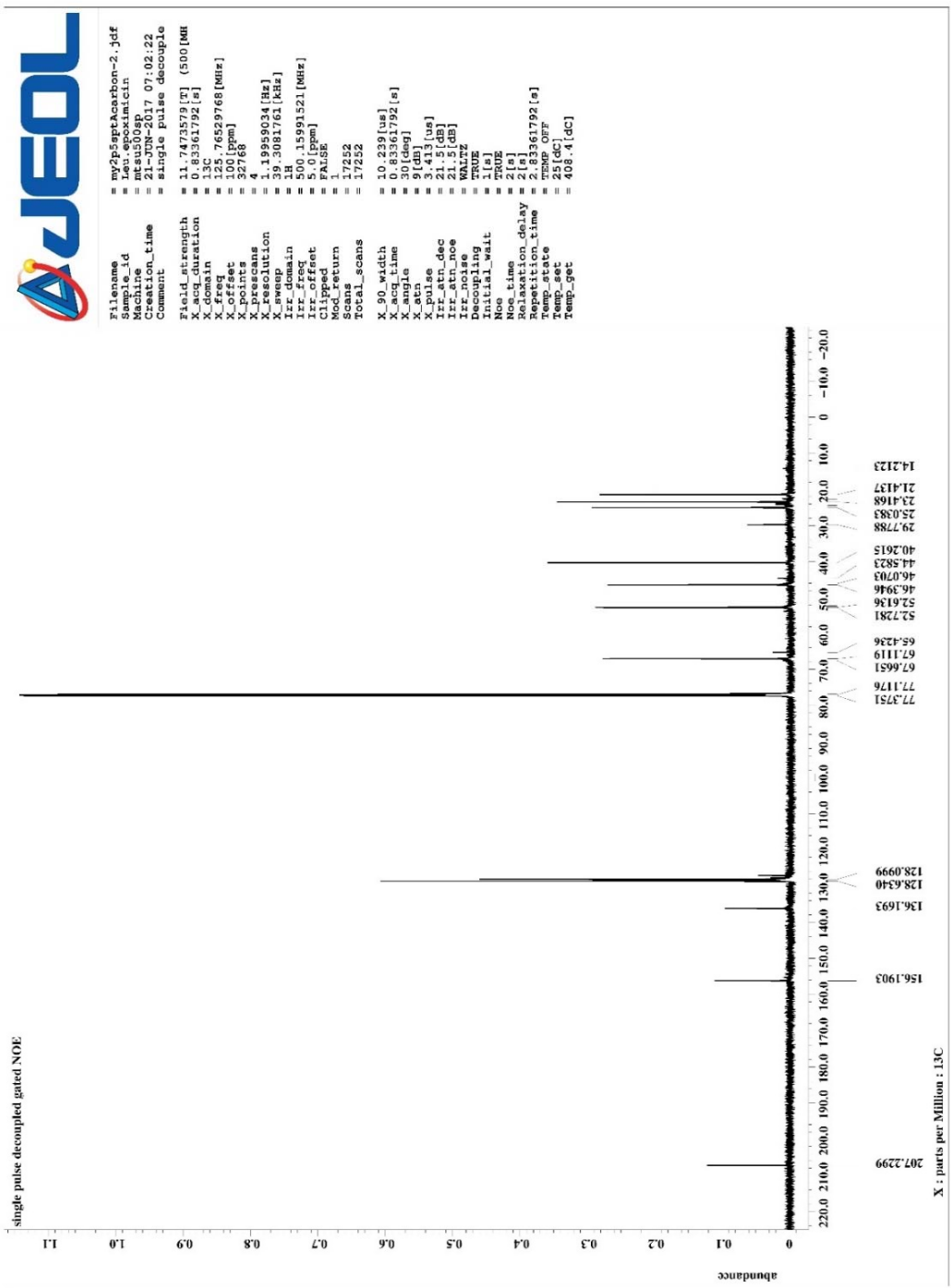
Field_strength = 11.747372 [T] (500 DMH)
X_acq_duration = 1X 74587904 [s]
X_freq = 500.15991521 [MHz]
X_offset = 5.0 [ppm]
X_points = 16384
X_resolution = 0.5727737 [Hz]
X_sweep = 9.38438438 [kHz]
X_irr_domain = 1X
X_irr_freq = 500.15991521 [MHz]
X_irr_offset = 5.0 [ppm]
X_tri_domain = 1X
X_tri_freq = 500.15991521 [MHz]
X_tri_offset = 5.0 [ppm]
Clipped = FALSE
Noisy_return = 1
Total_scans = 16

X_90_width = 14.18 [us]
X_pulse_program = 4587904 [s]
X_angle = 45 [deg]
X_atn = 4 [dB]
X_atn = 7.095 [us]
X_mode = Off
X_preamplifier = Off
Dante_preamplifier = FALSE
Initial_wait = 1 [s]
Relaxation_delay = 4 [s]
Repetition_time = 5.000000 [s]
Repetition_rate = 20.0 [Hz]
Temp_set = 25 [degC]
Temp_get = 408.2 [degC]

```



X : parts per Million : 1H



## Analysis Info

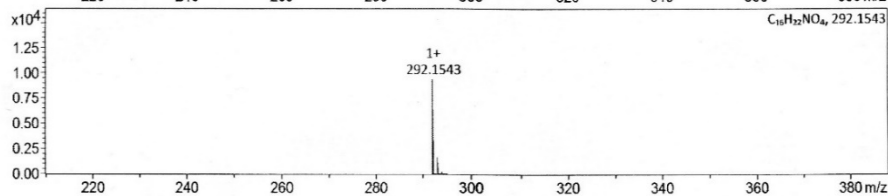
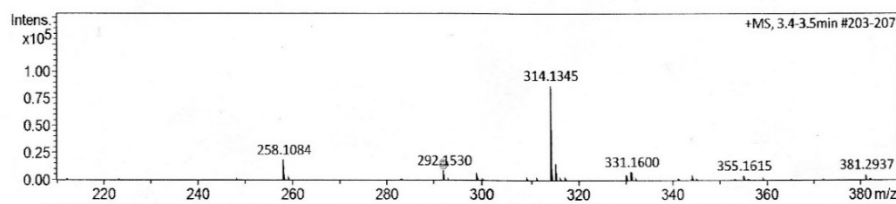
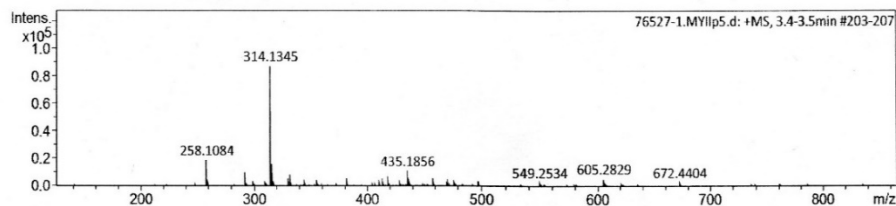
Analysis Name D:\Data\0726\76527-1.MYIIP5.d  
 Method 061517\_tune\_low\_pos\_NS.m  
 Sample Name 1.MYIIP5  
 Comment direct ifusion

Acquisition Date 7/26/2017 2:33:24 PM

Operator BDAL@DE  
 Instrument / Ser# micrOTOF II 8213750.1  
 0314

## Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1650 m/z	n/a	n/a	Set Divert Valve	Source



Meas. m/z	#	Ion Formula	m/z	err [ppm]	Mean err [ppm]	rdb	N-Rule	e <sup>-</sup> Conf
292.152982	1	C <sub>16</sub> H <sub>22</sub> NO <sub>4</sub>	292.154335	4.6	3.0	6.5	ok	even